

LAMPIRAN A.
Pembuatan Reagen

Komposisi Pereaksi 1 Metode Lowry, Komposisi Larutan Buffer Fosfat pH 7, Komposisi Media Agar Susu Skim dan Media Produksi

- a. Komposisi Pereaksi 1 menurut Pakpahan (2009) : Pereaksi A (2 gr Na_2CO_3 dilarutkan dalam NaOH 0,1 N hingga batas 100 mL dalam labu takar), pereaksi B (0,5 gr $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ dilarutkan dalam Natrium Kalium Tartrat 1% hingga batas 100 mL dalam labu takar) dan pereaksi 1 (50 mL pereaksi A ditambah 1 mL pereaksi B dan dihomogenkan).
- b. Komposisi Larutan Buffer Fosfat, yaitu ditimbang NaH_2PO_4 50 mM sebanyak 0,4024 gram dan $\text{Na}_2\text{HPO}_4 \cdot \text{H}_2\text{O}$ 50 mM sebanyak 0,9228 gram. Lalu kedua bahan dilarutkan dengan akuades dan dituang dalam beaker 500 ml.
- c. Komposisi media agar susu skim, yaitu media agar susu skim mengandung 4,9 gr susu skim dan 3 gr agar. 4,9 gr susu skim dilarutkan dalam 200 mL akuades kemudian disterilkan dengan otoklaf dengan tekanan 10 lbs. 3 gr agar dilarutkan dalam 200 mL akuades, disterilkan dengan otoklaf dengan tekanan 15 lbs dan dicampur dengan 20 mL larutan susu waktu masih panas.
- d. Komposisi media produksi, yaitu susu skim tanpa agar (4,9 gr susu skim dilarutkan dalam 200 mL akuades kemudian disterilkan

dengan otoklaf dengan tekanan 10 lbs. 200 mL akuades, disterilkan dengan otoklaf dengan tekanan 15 lbs dan dicampur dengan 20 mL larutan susu waktu masih panas.

LAMPIRAN B.

Contoh Perhitungan Aktivitas Enzim Protease

- **Perhitungan kurva baku**

Larutan baku @ 1ml + 5 ml pereaksi 1 + 0,5 ml pereaksi folin →

$$\text{vol}_{\text{tot}} = 6,5 \text{ ml}$$

No	Konsentrasi Albumin $\mu\text{g/ml}$	Konsentrasi dalam Kuvet $\mu\text{g/ml}$	Absorbansi
1	200	30,79	0,259
2	600	92,31	0,376
3	1000	153,85	0,523
4	1200	184,62	0,535
5	1500	230,77	0,584

Cara perhitungan :

C dalam kuvet

- ❖ $200 \text{ ppm} \rightarrow 1/6,5 \times 200 = 30,79 \mu\text{g/ml}$
- ❖ $600 \text{ ppm} \rightarrow 1/6,5 \times 600 = 92,31 \mu\text{g/ml}$
- ❖ $1000 \text{ ppm} \rightarrow 1/6,5 \times 1000 = 153,85 \mu\text{g/ml}$
- ❖ $1200 \text{ ppm} \rightarrow 1/6,5 \times 1200 = 184,62 \mu\text{g/ml}$
- ❖ $1500 \text{ ppm} \rightarrow 1/6,5 \times 1500 = 230,77 \mu\text{g/ml}$

RL : C kuvet Vs A

$$A = 0,2225$$

$$B = 1,6820 \times 10^{-3}$$

$$r = 0,9812$$

• **Contoh Perhitungan Kadar Protein Enzim dan Substrat Terhidrolisis pada Enzim kasar dengan waktu inkubasi 24 jam (sampel):**

1. 0,5 ml enzim + 1 ml buffer Fosfat + 5 ml pereaksi 1 + 0,5 ml pereaksi folin $\rightarrow V_{\text{tot}} = 7\text{ml}$

0,5 ml enzim (wkt inkubasi 24 jam₁) dengan absorbansi 0,570

$$Y = 0,2225 + 1,6820 \times 10^{-3} x$$

$$0,570 = 0,2225 + 1,6820 \times 10^{-3} x$$

$$X = 206,60$$

0,5 ml enzim (wkt inkubasi 24 jam₂) dengan absorbansi 0,590

$$Y = 0,2225 + 1,6820 \times 10^{-3} x$$

$$0,590 = 0,2225 + 1,6820 \times 10^{-3} x$$

$$X = 218,49 \mu\text{g/ml}$$

\rightarrow rata-rata hasil dari enzim 24 jam dan 48 jam = 212,545 $\mu\text{g/ml}$

Jadi, protein yang terdapat pada 7 ml larutan = 7 x 212,542 =

1487,815 μg .

2. Untuk kadar protein 0,5 ml protease kasar + 0,5 ml substrat (BSA 1000 $\mu\text{g/ml}$) + 5 ml pereaksi 1 + 0,5 ml pereaksi folin $\rightarrow V_{\text{tot}} = 7\text{ml}$

Diperoleh absorbansi 0,574 nm

$$Y = 0,2225 + 1,6820 \times 10^{-3} x$$

$$0,574 = 0,2225 + 1,6820 \times 10^{-3} x$$

$$X = 208,98 \mu\text{g/ml}$$

Jadi, jumlah protein dalam 7 ml larutan hasil reaksi E+S = 7 x 208,98 =

1462,86 μg .

3. Kadar protein dalam 0,5 ml substrat (BSA 1000 $\mu\text{g/ml}$) = 500 μg .

$$S = (a + b) - c$$

Keterangan :

S = Banyaknya substrat yang terhidrolisis

a = kadar protein dalam substrat

b = kadar protein dalam enzim

c = kadar protein hasil reaksi Enzim dan substrat

$$S = (a+b) - c$$

$$= 500 + 1487,815 - 1462,86$$

$$= \mathbf{524,955 \mu\text{g}}$$

Jadi, Banyaknya substrat yang terhidrolisis sebesar **524,955 μg** .

4. Aktivitas enzim (V) = $\Delta [s] / \Delta s$

$$= 524,955 / 10$$

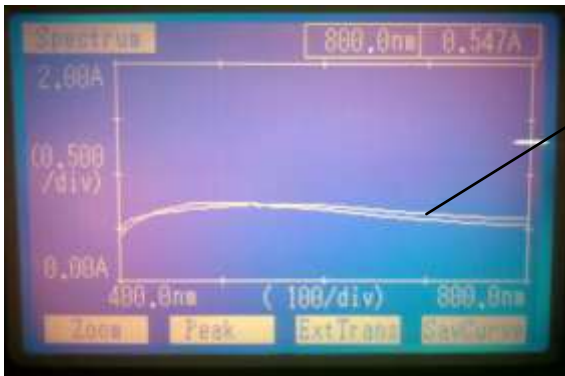
$$= \mathbf{52,4955 \text{ Unit}}$$
 (untuk 0,5 ml enzim)

$$= 104,991 \text{ U/ml}$$

LAMPIRAN C.
Pemilihan Panjang Gelombang



Gambar L3.1. Hasil serapan yang diberikan oleh baku Albumin (C5)



λ pada puncak spektrum adalah 715 nm

Gambar L3.2. Hasil tumpang tindih antara spektrum blanko enzim dengan spektrum sistem enzim+substrat

LAMPIRAN D.

Alat Inkubasi Bakteri Disertai Pengocokan 150 rpm



LAMPIRAN E.

Alat Sentrifugasi dengan Kecepatan 3000 rpm dan suhu 4°C



LAMPIRAN G.

Pengelompokan Spesies *Bacillus*



Lanjutan diagram Hidrolisis kasein (protease) positif (+)



LAMPIRAN H.
Pembacaan Microbact Kit 12A dan 12B

TABLE OF SUBSTRATES AND REACTIONS (12A/12E/24E):

Well No.	Designation	Reaction Principle	Reaction colours		Comments
			Negative	Positive	
1	Lysine	Lysine decarboxylase	Yellow	Blue-green	Green or blue is positive reaction. Bromothymol blue indicates formation of the specific amine cadaverine. Green should be regarded as a negative reaction. The pH shift indicated by bromothymol blue caused by formation of the specific amine putrescine is greater than that caused by lysine decarboxylation.
2	Ornithine	Ornithine decarboxylase	Yellow-green	Blue	H ₂ S is produced from thiosulphate.
3	H ₂ S	H ₂ S production	Straw colour	Black	H ₂ S reacts with ferric salts in the medium to form a black precipitate.
4	Glucose	Glucose fermentation	Blue-green	Yellow	Bromothymol blue indicator changes from blue to yellow when the carbohydrate is utilised to form acid.
5	Mannitol	Mannitol fermentation	Blue-green	Yellow	
6	Xylose	Xylose fermentation	Blue-green	Yellow	
		Hydrolysis of o-nitrophenyl-β-D-			β-galactosidase hydrolysis of the

7	ONPG	<i>galactopyranoside</i> ONPG by action of β -galactosidase	Colourless	Yellow	<i>indole</i> ONPG releases yellow ortho-nitrophenol. Indole is formed from metabolism of tryptophan.
8	Indole	Indole production from tryptophan	Colourless	Pink-red	Indole Kovacs reagent forms a pink-red complex with indole. Ammonium released from splitting of urea causes the pH to rise - indicated by phenol red changing from yellow to pink-red.
9	Urease	Urea hydrolysis	Straw colour	Pink-red	Acetoin is produced from glucose indicated by the formation of a pink-red complex after the addition of alpha-naphthol and creatine.
10	VP	Acetoin production (Voges-Proskauer reaction)	Straw colour	Pink-red	Citrate is the sole carbon source, which if utilized results in a pH rise, indicated by bromothymol blue, with a colour change from green to blue.
11	Citrate	Citrate utilization (citrate is the only source of carbon)	Green	Blue	Tryptophan deaminase forms indolepyruvic acid from tryptophan which produces a brown colour in the presence of ferric ions.
12	TDA	Production of indolepyruvate by deamination of tryptophan	Straw colour	Cherry red	Indole positive organisms may produce a brown colour. This is a negative reaction.

Well No. 12B/24E	Designation	Reaction Principle	Reaction colours		Comments
			Negative	Positive	
1/13	Gelatin	Gelatin liquefaction	Colourless	Black	<p>Liquefaction of gelatin by proteolytic enzymes diffuses the black pigment. Solid gelatin particles which may drift across the well after rehydration should be considered as a negative reaction.</p> <p>Sodium malonate is the sole carbon source and this inhibits the conversion of succinic acid to fumaric acid. An organism unable to utilize this substrate results in the accumulation of succinic acid and the organism cannot grow. Bromothymol blue is the indicator. Yellow-green is indicative of a negative result.</p> <p>Utilisation of Na malonate at the same time that ammonium sulphate is utilised as the nitrogen source produces sodium hydroxide resulting in increased alkalinity and a blue colouration.</p>
2/14	Malonate	Malonate inhibition	Green	Blue	<p>Utilisation of Na malonate at the same time that ammonium sulphate is utilised as the nitrogen source produces sodium hydroxide resulting in increased alkalinity and a blue colouration.</p>
3/15	Inositol	Inositol fermentation	Blue- green	Yellow	

4/16	Sorbitol	Sorbitol fermentation	Blue-green	Yellow	
5/17	Rhamnose	Rhamnose fermentation	Blue-green	Yellow	
6/18	Sucrose	Sucrose fermentation	Blue-green	Yellow	Bromothymol blue indicator changes from blue to yellow when the carbohydrate is fermented.
7/19	Lactose	Lactose fermentation	Blue-green	Yellow	
8/20	Arabinose	Arabinose fermentation	Blue-green	Yellow	
9/21	Adonitol	Adonitol fermentation	Blue-green	Yellow	
10/22	Raffinose	Raffinose fermentation	Blue-green	Yellow	
11/23	Salicin	Salicin fermentation	Blue-green	Yellow	
		Arginine dihydrolase			
12/24	Arginine	24 hours	Yellow	Green-blue	
		48 hours	Yellow-green	Blue	

LAMPIRAN I.

Hasil Uji Fisiologis Isolat

Tabel 1. Isolat bakteri A pada uji amilase positif (+)

No	Jenis uji	A	1	2	3	4	5	6	7	9	10	11	12	14	16	18	19	22	24
1	Pewarnaan Gram	+	+	+	+	d	d	+	+	+	+	+	+	d	d	-	-	d	d
2	Bentuk sel	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt	bt
3	Sel yang membentuk rantai	-	+	+	+	d	d	d	d	d	d	d	d	d	d	-	-	d	d
4	Motilitas	-	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
5	Panjang sel > 3 µm	+	+	+	+	+	+	-	d	-	-	-	-	+	d	d	d	-	-
6	Bentuk dan ukuran spora	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX	VX
7	Pembungkakan sel spora	-	-	-	-	-	d	d	-	-	-	-	d	+	+	+	+	d	+
Uji produksi asam dari fermentasi karbohidrat :																			
10	Glucose	-	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
11	Cellulose (Hydrolysis selulase)	-	d	d	d	-	d	+	+	+	+	+	d	+	+	+	+	d	d
12	Galactose (Gelatin)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
13	Mannose (Mannitol)	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
14	Raffinose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
15	Salicin	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
16	Xylose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
17	ONPG	+	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
18	Citrate	-	-	d	d	+	+	+	+	+	+	+	+	+	+	+	+	+	+
19	Urease	d	-	d	d	-	-	-	-	-	-	-	-	-	-	-	-	-	-
20	Indole	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
21	VP (<i>Voges Proskauer</i>)	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
22	Reduksi nitrate	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
23	Oksidase	+	d	d	d	d	+	+	+	+	+	+	+	+	+	+	+	+	+
24	Hidrolisis kasein (protein)	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
24	Hidrolisis tepung (amilum)	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+	+
Koefisien sebanding (%)		78	74	70	65	65	52	48	48	48	48	43	43	37	43	48	48	43	48

Keterangan :

(T) spora terminal, (V) spora sentral atau subterminal, (X) spora oval, (Y) spora bulat

- (1) *Bacillus anthracis*; (2) *Bacillus cereus*, *Bacillus anthracoides*; (3) *Bacillus mycoideus*; (4) *Bacillus thuringiensis*; (5) *Bacillus firmus*; (6) *Bacillus licheniformis*; (7) *Bacillus megaterium*; (9) *Bacillus subtilis*; (10) *Bacillus licheniformis*; (11) *Bacillus amyloliquefaciens*; (12) *Bacillus coagulans*; (14) *Bacillus alvei*; (16) *Bacillus circulans*; (18) *Bacillus macerans*; (19) *Bacillus polymyxa*; (22) *Bacillus stearothermophilus*; (24) *Bacillus stearothermophilus*

LAMPIRAN J.

Perhitungan Angka Koefisien Sebanding *Bacillus*

Perhitungan persentase indeks kesamaan menggunakan koefisien sebanding (Ss) mencakup kesamaan positif dan negatif (Stainer *et al.*, 1986). Perhitungan menggunakan rumus:

$$Ss = \frac{a+d}{b+c+d} \times 100\%$$

Keterangan :

Ss= Koefisien sebanding

a= jumlah ciri positif pada kedua galur bakteri

b= Jumlah ciri positif pada galur 1 dan negatif pada galur 2

c= Jumlah ciri negatif pada galur 1 dan positif pada galur 2

d= Jumlah ciri negatif pada kedua galur bakteri

Contoh :

(koefisien sebanding dengan *Bacillus anthracis*)

a = 9

b = 3

c = 2

d = 9

$$\begin{aligned} Ss &= \frac{9+9}{3+2+9} \times 100\% \\ &= 78\% \end{aligned}$$