

BAB V

KESIMPULAN DAN SARAN

V.1 Kesimpulan

Berdasarkan hasil penelitian, komposit mRGO/Co yang dikombinasikan dengan Co dapat mengadsorpsi metilena biru dengan waktu setimbang yang singkat yaitu <30 menit. Hasil kinetika adsorpsi komposit tersebut sesuai dengan model kinetika order satu semu yang berarti adsorpsi terjadi secara fisik. Nilai Q_e pada suhu 303 K dinyatakan oleh persamaan order satu semu sebesar 56,115 mg/g. Nilai Q_{max} dari persamaan langmuir menyatakan kapasitas adsorpsi maksimum sebesar 180,313 mg/g pada suhu 323 K. Percobaan adsorpsi isoterm sesuai dengan model Sips dan Redlich Peterson yang berarti melibatkan adsorpsi monolayer dengan adanya permukaan yang homogen dan juga heterogen. Nilai negatif dari ΔG^0 dan ΔH^0 , serta nilai positif dari ΔS^0 menunjukkan karakteristik spontan dan eksotermik dari proses adsorpsi metilena biru menggunakan mRGO/Co.

V.2 Saran

Penulis merekomendasikan beberapa saran untuk penelitian mRGO/Co dengan *green reductor* di masa depan. Pertama, untuk mendapatkan pemahaman yang lebih baik tentang adsorpsi isoterm, disarankan untuk melakukan lebih banyak variasi konsentrasi metilena biru. Variasi konsentrasi metilena biru dapat memperjelas jenis isoterm adsorpsi yang sesuai untuk sistem ini.

Dalam penelitian ini, model isoterm Langmuir, Freundlich, Redlich-Peterson, dan Sips telah digunakan untuk menganalisis data adsorpsi. Namun, selama percobaan, penulis mengamati bahwa ada beberapa ketidakpastian

dalam penyesuaian model dengan data, yang mungkin disebabkan oleh kompleksitas sistem adsorpsi yang melibatkan metilena biru dan mRGO/Co.

Oleh karena itu, penulis merekomendasikan penggunaan model isoterm yang berbeda dalam penelitian selanjutnya. Salah satu persamaan yang dapat dipertimbangkan adalah model isoterm BET. Model BET mempertimbangkan adsorpsi *multilayer*, yang mungkin memberikan penjelasan yang lebih baik dalam kondisi ketika metilena biru menempel pada lapisan mRGO/Co yang sudah mengadsorpsi molekul lain.

Sebagai alternatif lain, penulis juga merekomendasikan penggunaan model isoterm Temkin dalam penelitian selanjutnya. Model ini mempertimbangkan efek interaksi antara adsorbat dan adsorben, dan mungkin memberikan penjelasan yang lebih baik tentang proses adsorpsi di permukaan mRGO/Co.

DAFTAR PUSTAKA

- [1] S. J. Salih, A. S. Abdul Kareem, and S. S. Anwer, "Adsorption of anionic dyes from textile wastewater utilizing raw corncob," *Heliyon*, vol. 8, no. 8, pp. 1–9, 2022, doi: 10.1016/j.heliyon.2022.e10092.
- [2] M. Rafique, F. Shafiq, S. S. Ali Gillani, M. Shakil, M. B. Tahir, and I. Sadaf, "Eco-friendly green and biosynthesis of copper oxide nanoparticles using *Citrofortunella microcarpa* leaves extract for efficient photocatalytic degradation of Rhodamin B dye from textile wastewater," *Optik (Stuttg.)*, vol. 208, pp. 1–28, 2020, doi: 10.1016/j.ijleo.2019.164053.
- [3] O. A. Attallah, M. A. Al-Ghobashy, M. Nebsen, and M. Y. Salem, "Removal of cationic and anionic dyes from aqueous solution with magnetite/pectin and magnetite/silica/pectin hybrid nanocomposites: Kinetic, isotherm and mechanism analysis," *RSC Adv.*, vol. 6, no. 14, pp. 11461–11480, 2016, doi: 10.1039/c5ra23452b.
- [4] M. Athari, M. Fattahi, M. Khosravi-Nikou, and A. Hajhariri, "Adsorption of different anionic and cationic dyes by hybrid nanocomposites of carbon nanotube and graphene materials over UiO-66," *Sci. Rep.*, vol. 12, no. 1, pp. 1–15, 2022, doi: 10.1038/s41598-022-24891-2.
- [5] N. U. M. Nizam, M. M. Hanafiah, E. Mahmoudi, A. A. Halim, and A. W. Mohammad, "The removal of anionic and cationic dyes from an aqueous solution using biomass-based activated carbon," *Sci. Rep.*, vol. 11, no. 1, pp. 1–17, 2021, doi: 10.1038/s41598-021-88084-z.
- [6] M. A. Abid, D. A. Abid, W. J. Aziz, and T. M. Rashid, "Iron oxide nanoparticles synthesized using garlic and onion peel extracts rapidly

- degrade methylene blue dye,” *Phys. B Condens. Matter*, vol. 622, no. February, pp. 1–11, 2021, doi: 10.1016/j.physb.2021.413277.
- [7] T. A. Salman and M. I. Ali, “Potential Application of Natural and Modified Orange Peel as an Eco–friendly Adsorbent for Methylene Blue Dye,” *Iraqi J. Sci.*, vol. 57, no. 1A, pp. 1–13, 2016.
- [8] V. S. Munagapati and D. S. Kim, “Adsorption of anionic azo dye Congo Red from aqueous solution by Cationic Modified Orange Peel Powder,” *J. Mol. Liq.*, vol. 220, pp. 540–548, 2016, doi: 10.1016/j.molliq.2016.04.119.
- [9] N. Jahan *et al.*, “A comparative study on sorption behavior of graphene oxide and reduced graphene oxide towards methylene blue,” *Case Stud. Chem. Environ. Eng.*, vol. 6, no. June, pp. 1–10, 2022, doi: 10.1016/j.cscee.2022.100239.
- [10] F. Mohamed *et al.*, “Activated carbon derived from sugarcane and modified with natural zeolite for efficient adsorption of methylene blue dye : experimentally and theoretically approaches,” *Sci. Rep.*, pp. 1–18, 2022, doi: 10.1038/s41598-022-22421-8.
- [11] D. Ramutshatsha-makhwedzha, A. Mavhungu, M. L. Moropeng, and R. Mbaya, “Activated carbon derived from waste orange and lemon peels for the adsorption of methyl orange and methylene blue dyes from wastewater,” *Heliyon*, vol. 8, no. July, p. e09930, 2022, doi: 10.1016/j.heliyon.2022.e09930.
- [12] I. Ayouch, I. Kassem, Z. Kassab, I. Barrak, and A. Barhoun, “Crosslinked carboxymethyl cellulose-hydroxyethyl cellulose hydrogel films for adsorption of cadmium and methylene blue from aqueous solutions,” *Surfaces and Interfaces*, vol. 24, no. March, p. 101124, 2021, doi: 10.1016/j.surfin.2021.101124.

- [13] M. N. Nimbalkar and 2019.pdf Bhat, Badekai Ramachandrafile A. EL-Sharkaway, "Simultaneous adsorption of methylene blue and heavy metals from water using Zr-MOF having free carboxylic group," *J. Environ. Chem. Eng.*, vol. 9, no. 5, p. 106216, 2021, doi: 10.1016/j.jece.2021.106216.
- [14] Y. Zhang *et al.*, "The utilization of a three-dimensional reduced graphene oxide and montmorillonite composite aerogel as a multifunctional agent for wastewater treatment," *RSC Adv.*, vol. 8, no. 8, pp. 4239–4248, 2018, doi: 10.1039/c7ra13103h.
- [15] N. Morimoto, T. Kubo, and Y. Nishina, "Tailoring the oxygen content of graphite and reduced graphene oxide for specific applications," *Sci. Rep.*, vol. 6, pp. 4–11, 2016, doi: 10.1038/srep21715.
- [16] S. Ramanathan *et al.*, "Grape Seed Extract Assisted Synthesis of Dual-Functional Anatase TiO₂Decorated Reduced Graphene Oxide Composite for Supercapacitor Electrode Material and Visible Light Photocatalytic Degradation of Bromophenol Blue Dye," *ACS Omega*, vol. 6, no. 23, pp. 14734–14747, 2021, doi: 10.1021/acsomega.0c02325.
- [17] P. Wang, X. Zhou, Y. Zhang, L. Wang, K. Zhi, and Y. Jiang, "Synthesis and application of magnetic reduced graphene oxide composites for the removal of bisphenol A in aqueous solution - A mechanistic study," *RSC Adv.*, vol. 6, no. 104, pp. 102348–102358, 2016, doi: 10.1039/c6ra23542e.
- [18] J. Qi, Y. Hou, J. Hu, W. Ruan, Y. Xiang, and X. Wei, "Decontamination of methylene Blue from simulated wastewater by the mesoporous rGO/Fe/Co nanohybrids: Artificial intelligence

- modeling and optimization,” *Mater. Today Commun.*, vol. 24, no. October 2019, pp. 1–15, 2020, doi: 10.1016/j.mtcomm.2019.100709.
- [19] J. Li *et al.*, “Highly efficient adsorption and mechanism of alkylphenols on magnetic reduced graphene oxide,” *Chemosphere*, vol. 283, no. June, pp. 1–10, 2021, doi: 10.1016/j.chemosphere.2021.131232.
- [20] A. M. Huízar-Félix, C. Aguilar-Flores, A. Martínez-De-La Cruz, J. M. Barandiarán, S. Sepúlveda-Guzmán, and R. Cruz-Silva, “Removal of tetracycline pollutants by adsorption and magnetic separation using reduced graphene oxide decorated with α -Fe₂O₃ nanoparticles,” *Nanomaterials*, vol. 9, no. 3, pp. 1–14, 2019, doi: 10.3390/nano9030313.
- [21] A. I. A. Sherlala, A. A. A. Raman, M. M. Bello, and A. Buthiyappan, “Adsorption of arsenic using chitosan magnetic graphene oxide nanocomposite,” *J. Environ. Manage.*, vol. 246, no. January, pp. 547–556, 2019, doi: 10.1016/j.jenvman.2019.05.117.
- [22] N. C. Joshi and P. Gururani, “Advances of graphene oxide based nanocomposite materials in the treatment of wastewater containing heavy metal ions and dyes,” *Curr. Res. Green Sustain. Chem.*, vol. 5, no. March, pp. 1–11, 2022, doi: 10.1016/j.crgsc.2022.100306.
- [23] Y. A. Wijaya, D. Widyadinata, W. Irawaty, And A. Ayucitra, “Fractionation Of Phenolic Compounds From Kaffir Lime (*Citrus Hystrix*) Peel Extract And Evaluation Of Antioxidant Activity,” *Reaktor*, Vol. 17, No. 3, Pp. 111–117, 2017, Doi: 10.14710/Reaktor.17.3.111-117.
- [24] S. Srifuengfung *et al.*, “Antibacterial oral sprays from kaffir lime (*Citrus hystrix* DC.) fruit peel oil and leaf oil and their activities

- against respiratory tract pathogens,” *J. Tradit. Complement. Med.*, vol. 10, no. 6, pp. 594–598, 2020, doi: 10.1016/j.jtcme.2019.09.003.
- [25] V. Priliana *et al.*, “Reduction of Graphene Oxide Using Citrus hystrix Peels Extract for Methylene Blue Adsorption,” *Sustain.*, vol. 14, no. 19, pp. 1–15, 2022, doi: 10.3390/su141912172.
- [26] J. Chen, B. Yao, C. Li, and G. Shi, “An improved Hummers method for eco-friendly synthesis of graphene oxide,” *Carbon N. Y.*, vol. 64, no. 1, pp. 225–229, 2013, doi: 10.1016/j.carbon.2013.07.055.
- [27] A. Jiříčková, O. Jankovský, Z. Sofer, and D. Sedmidubský, “Synthesis and Applications of Graphene Oxide,” *Materials*, vol. 15, no. 3, pp. 1–21, 2022. doi: 10.3390/ma15030920.
- [28] S. Thangavel and G. Venugopal, “Understanding the adsorption property of graphene-oxide with different degrees of oxidation levels,” *Powder Technol.*, vol. 257, pp. 141–148, 2014, doi: 10.1016/j.powtec.2014.02.046.
- [29] J. Guerrero-Contreras and F. Caballero-Briones, “Graphene oxide powders with different oxidation degree, prepared by synthesis variations of the Hummers method,” *Mater. Chem. Phys.*, vol. 153, pp. 209–220, 2015, doi: 10.1016/j.matchemphys.2015.01.005.
- [30] L. Shen *et al.*, “Analysis of oxidation degree of graphite oxide and chemical structure of corresponding reduced graphite oxide by selecting different-sized original graphite,” *RSC Adv.*, vol. 8, no. 31, pp. 17209–17217, 2018, doi: 10.1039/c8ra01486h.
- [31] J. Aixart, F. Díaz, J. Llorca, and J. Rosell-Llompart, “Increasing reaction time in Hummers’ method towards well exfoliated graphene oxide of low oxidation degree,” *Ceram. Int.*, vol. 47, no. 15, pp. 22130–22137, 2021, doi: 10.1016/j.ceramint.2021.04.235.

- [32] N. I. Zaaba, K. L. Foo, U. Hashim, S. J. Tan, W. W. Liu, and C. H. Voon, "Synthesis of Graphene Oxide using Modified Hummers Method: Solvent Influence," *Procedia Eng.*, vol. 184, pp. 469–477, 2017, doi: 10.1016/j.proeng.2017.04.118.
- [33] M. Vazquez-Jaime *et al.*, "Effective removal of arsenic from an aqueous solution by ferrihydrite/goethite graphene oxide composites using the modified Hummers method," *J. Environ. Chem. Eng.*, vol. 8, no. 6, pp. 1–12, 2020, doi: 10.1016/j.jece.2020.104416.
- [34] W. Ahlawat, N. Dilbaghi, and S. Kumar, "Evaluation of graphene oxide and its composite as potential sorbent for removal of cationic and anionic dyes," *Mater. Today Proc.*, vol. 45, pp. 5500–5505, 2021, doi: 10.1016/j.matpr.2021.02.215.
- [35] R. Tarcan, O. Todor-Boer, I. Petrovai, C. Leordean, S. Astilean, and I. Botiz, "Reduced graphene oxide today," *J. Mater. Chem. C*, vol. 8, no. 4, pp. 1198–1224, 2020, doi: 10.1039/c9tc04916a.
- [36] A. Razaq, F. Bibi, X. Zheng, R. Papadakis, S. H. M. Jafri, and H. Li, "Review on Graphene-, Graphene Oxide-, Reduced Graphene Oxide-Based Flexible Composites: From Fabrication to Applications," *Materials (Basel)*, vol. 15, no. 3, 2022, doi: 10.3390/ma15031012.
- [37] M. Q. Jian, H. H. Xie, K. L. Xia, and Y. Y. Zhang, "Challenge and Opportunities of Carbon Nanotubes," in *Industrial Applications of Carbon Nanotubes*, Elsevier Inc., 2017, pp. 433–476. doi: 10.1016/B978-0-323-41481-4.00015-0.
- [38] J. Xiao, W. Lv, Z. Xie, Y. Tan, Y. Song, and Q. Zheng, "Environmentally friendly reduced graphene oxide as a broad-spectrum adsorbent for anionic and cationic dyes: Via π - π Interactions," *J. Mater. Chem. A*, vol. 4, no. 31, pp. 12126–12135,

- 2016, doi: 10.1039/c6ta04119a.
- [39] L. The Vinh, T. Ngoc Khiem, H. Dang Chinh, P. Van Tuan, and V. T. Tan, “Adsorption capacities of reduced graphene oxide: Effect of reductants,” *Mater. Res. Express*, vol. 6, no. 7, 2019, doi: 10.1088/2053-1591/ab1862.
- [40] D. Zhao, G. Zhu, Y. Ding, and J. Zheng, “Construction of a different polymer chain structure to study π - π interaction between polymer and reduced graphene oxide,” *Polymers (Basel)*, vol. 10, no. 7, 2018, doi: 10.3390/polym10070716.
- [41] W. Fu and Z. Huang, “Magnetic dithiocarbamate functionalized reduced graphene oxide for the removal of Cu(II), Cd(II), Pb(II), and Hg(II) ions from aqueous solution: Synthesis, adsorption, and regeneration,” *Chemosphere*, vol. 209, no. Ii, pp. 449–456, 2018, doi: 10.1016/j.chemosphere.2018.06.087.
- [42] Z. Liu, Z. Gao, L. Xu, and F. Hu, “Polypyrrole modified magnetic reduced graphene oxide composites: Synthesis, characterization and application for selective lead adsorption,” *RSC Adv.*, vol. 10, no. 30, pp. 17524–17533, 2020, doi: 10.1039/d0ra01546f.
- [43] S. Zhao *et al.*, “Fabrication of rGO/Fe₃O₄ magnetic composite for the adsorption of anthraquinone-2-sulfonate in water phase,” *Water (Switzerland)*, vol. 13, no. 17, pp. 1–12, 2021, doi: 10.3390/w13172315.
- [44] M. Adel, M. A. Ahmed, and A. A. Mohamed, “Synthesis and characterization of magnetically separable and recyclable crumbled MgFe₂O₄/reduced graphene oxide nanoparticles for removal of methylene blue dye from aqueous solutions,” *J. Phys. Chem. Solids*, vol. 149, no. May 2020, pp. 1–12, 2021, doi:

- 10.1016/j.jpccs.2020.109760.
- [45] F. Habibi, M. Seyyedi, and B. Ayati, “Synthesis and Application of Reusable and Magnetic RGO / Fe₃O₄ Nanocomposites in BR46 Removal from an Aqueous Solution ; Future Prospects of an Efficient Adsorption Platform,” vol. 13, no. 8, pp. 900–913, 2022.
- [46] J. Z. Sun *et al.*, “Adsorption and removal of triphenylmethane dyes from water by magnetic reduced graphene oxide,” *Water Sci. Technol.*, vol. 70, no. 10, pp. 1663–1669, 2014, doi: 10.2166/wst.2014.427.
- [47] S. Park, J. An, J. R. Potts, A. Velamakanni, S. Murali, and R. S. Ruoff, “Hydrazine-reduction of graphite- and graphene oxide,” *Carbon N. Y.*, vol. 49, no. 9, pp. 3019–3023, 2011, doi: 10.1016/j.carbon.2011.02.071.
- [48] L. Chit-aree, Y. Unpaprom, R. Ramaraj, and M. Thirabunyanon, “Valorization and biorefinery of kaffir lime peels waste for antifungal activity and sustainable control of mango fruit anthracnose,” *Biomass Convers. Biorefinery*, no. 0123456789, 2021, doi: 10.1007/s13399-021-01768-4.
- [49] O. Ullah Shirazi, M. Muzaffar Ali Khan Khattak, N. Azwani Mohd Shukri, M. A. Mohd Nur Nasyriq, O. Shirazi, and M. A. Nur Nasyriq, “Determination of total phenolic, flavonoid content and free radical scavenging activities of common herbs and spices,” *J. Pharmacogn. Phytochem. JPP*, vol. 104, no. 33, pp. 104–108, 2014.
- [50] S. Kupina, C. Fields, M. C. Roman, and S. L. Brunelle, “Determination of total phenolic content using the Folin-C assay: Single-laboratory validation, first action 2017.13,” *J. AOAC Int.*, vol. 102, no. 1, pp. 320–321, 2019, doi: 10.5740/jaoacint.2017.13.

- [51] N. Phuyal, P. K. Jha, P. P. Raturi, and S. Rajbhandary, "In Vitro Antibacterial Activities of Methanolic Extracts of Fruits, Seeds, and Bark of *Zanthoxylum armatum* DC," *J. Trop. Med.*, vol. 2020, 2020, doi: 10.1155/2020/2803063.
- [52] H. Noreen, N. Semmar, M. Farman, and J. S. O. McCullagh, "Measurement of total phenolic content and antioxidant activity of aerial parts of medicinal plant *Coronopus didymus*," *Asian Pac. J. Trop. Med.*, vol. 10, no. 8, pp. 792–801, 2017, doi: 10.1016/j.apjtm.2017.07.024.
- [53] E. A. Al-Maliky, H. A. Gzar, and M. G. Al-Azawy, "Determination of Point of Zero Charge (PZC) of Concrete Particles Adsorbents," *IOP Conf. Ser. Mater. Sci. Eng.*, vol. 1184, no. 1, p. 012004, 2021, doi: 10.1088/1757-899x/1184/1/012004.
- [54] S. N. Rao, "Adsorption," in *Interface Science and Technology*, vol. 21, 2017, pp. 251–331. doi: 10.1016/B978-0-12-801970-2.00005-7.
- [55] M. El-Dairi *et al.*, "Introduction," in *Alliances: Re/envisioning Indigenous-non-indigenous Relationships*, vol. 25, no. Ii, 2014, pp. 11–28.
- [56] M. T. Yagub, T. K. Sen, S. Afroze, and H. M. Ang, "Dye and its removal from aqueous solution by adsorption: A review," *Adv. Colloid Interface Sci.*, vol. 209, pp. 172–184, 2014, doi: 10.1016/j.cis.2014.04.002.
- [57] E. D. Revellame, D. L. Fortela, W. Sharp, R. Hernandez, and M. E. Zappi, "Adsorption kinetic modeling using pseudo-first order and pseudo-second order rate laws: A review," *Clean. Eng. Technol.*, vol. 1, no. October, p. 100032, 2020, doi: 10.1016/j.clet.2020.100032.
- [58] S. Ersali, V. Hadadi, O. Moradi, and A. Fakhri, "Pseudo-second-

- order kinetic equations for modeling adsorption systems for removal of ammonium ions using multi-walled carbon nanotube,” *Fullerenes, Nanotub. Carbon Nanostructures*, p. 150527104639002, 2013, doi: 10.1080/1536383x.2013.787610.
- [59] L. Liu, X. B. Luo, L. Ding, and S. L. Luo, “Application of Nanotechnology in the Removal of Heavy Metal From Water,” in *Nanomaterials for the Removal of Pollutants and Resource Reutilization*, 2018, pp. 83–147. doi: 10.1016/B978-0-12-814837-2.00004-4.
- [60] H. Tun and C. C. Chen, “Isosteric heat of adsorption from thermodynamic Langmuir isotherm,” *Adsorption*, vol. 27, no. 6, pp. 979–989, 2021, doi: 10.1007/s10450-020-00296-3.
- [61] A. K. Singh, “Nanoparticle Ecotoxicology,” in *Engineered Nanoparticles*, 2016, pp. 343–450. doi: 10.1016/b978-0-12-801406-6.00008-x.
- [62] M. Vigdorowitsch, A. Pchelintsev, L. Tsygankova, and E. Tanygina, “Freundlich Isotherm : An Adsorption Model Complete Framework applied sciences Freundlich Isotherm : An Adsorption Model Complete Framework,” no. August, 2021, doi: 10.3390/app11178078.
- [63] J. Wang and X. Guo, “Adsorption isotherm models: Classification, physical meaning, application and solving method,” *Chemosphere*, vol. 258, p. 127279, 2020, doi: 10.1016/j.chemosphere.2020.127279.
- [64] N. Tzabar and H. J. M. ter Brake, “Adsorption isotherms and Sips models of nitrogen, methane, ethane, and propane on commercial activated carbons and polyvinylidene chloride,” *Adsorption*, vol. 22, no. 7, pp. 901–914, 2016, doi: 10.1007/s10450-016-9794-9.

- [65] P. F. Zito, A. Caravella, A. Brunetti, E. Drioli, and G. Barbieri, “Estimation of Langmuir and Sips Models Adsorption Parameters for NaX and NaY FAU Zeolites,” 2015, doi: 10.1021/acs.jced.5b00215.
- [66] N. A. Oladoja, “Fundamentals and Applications,” in *Composite Nanoadsorbents*, Elsevier Inc., 2019, pp. 85–118. doi: 10.1016/B978-0-12-814132-8.00005-8.
- [67] J. Yeon, Y. Jung, and W. H. Kim, “Adsorption of Carboxymethylated Polyethyleneimine (CM-PEI) on a Microporous Activated Carbon,” vol.119, pp. 103–106, 2007, doi: 10.4028/www.scientific.net/SSP.119.103.
- [68] P. Saha and S. Chowdhury, “Insight Into Adsorption Thermodynamics,” *Thermodynamics*, 2011, doi: 10.5772/13474.
- [69] Z. Zhou, Y. Li, S. Yao, and H. Yan, “Preparation of calcium carbonate@graphene oxide core-shell microspheres in ethylene glycol for drug delivery,” *Ceram. Int.*, vol. 42, no. 2, pp. 2281–2288, 2016, doi: 10.1016/j.ceramint.2015.10.022.
- [70] I. O. Faniyi *et al.*, “The comparative analyses of reduced graphene oxide (RGO) prepared via green, mild and chemical approaches,” *SN Appl. Sci.*, vol. 1, no. 10, 2019, doi: 10.1007/s42452-019-1188-7.
- [71] G. Lujanienė *et al.*, “Magnetic graphene oxide based nanocomposites for removal of radionuclides and metals from contaminated solutions,” *J. Environ. Radioact.*, vol. 166, pp. 166–174, 2017, doi: 10.1016/j.jenvrad.2016.02.014.
- [72] A. A. Radhakrishnan and B. B. Beena, “Structural and Optical Absorption Analysis of CuO Nanoparticles,” *Indian J. Adv. Chem. Sci.*, vol. 2, no. 2, pp. 158–161, 2014.
- [73] H. Zhang *et al.*, “Kinetics and equilibrium studies of the adsorption

- of methylene blue on Euryale ferox shell-based activated carbon,” *Micro Nano Lett.*, vol. 13, no. 4, pp. 552–557, 2018, doi: 10.1049/mnl.2017.0638.
- [74] L. Mouni *et al.*, “Removal of Methylene Blue from aqueous solutions by adsorption on Kaolin: Kinetic and equilibrium studies,” *Appl. Clay Sci.*, vol. 153, no. March 2017, pp. 38–45, 2018, doi: 10.1016/j.clay.2017.11.034.
- [75] J. P. Vareda, “On validity, physical meaning, mechanism insights and regression of adsorption kinetic models,” *J. Mol. Liq.*, vol. 376, p. 121416, 2023, doi: 10.1016/j.molliq.2023.121416.
- [76] M. A. Nazir *et al.*, “Enhanced adsorption removal of methyl orange from water by porous bimetallic Ni / Co MOF composite: a systematic study of adsorption kinetics,” *Int. J. Environ. Anal. Chem.*, vol. 00, no. 00, pp. 1–16, 2021, doi: 10.1080/03067319.2021.1931855.
- [77] R. Begum *et al.*, “Chemical reduction of methylene blue in the presence of nanocatalysts: A critical review,” *Rev. Chem. Eng.*, vol. 36, no. 6, pp. 749–770, 2020, doi: 10.1515/revce-2018-0047.
- [78] S. Sohrabnezhad, A. Pourahmad, R. Rakhshaei, A. Radaee, and S. Heidarian, “Catalytic reduction of methylene blue by sulfide ions in the presence of nanoAlMCM-41 material,” *Superlattices Microstruct.*, vol. 47, no. 3, pp. 411–421, 2010, doi: 10.1016/j.spmi.2009.12.006.
- [79] L. Fan *et al.*, “Degradation of Methylene Blue by Hot Electrons Transfer in SnSe,” *Adv. Mater. Interfaces*, vol. 10, no. 11, 2023, doi: 10.1002/admi.202202207.
- [80] W. X. 1 Zhicheng Zhu 1, Yaji Huang 2, Mengzhu Yu 1, Haoqiang

- Cheng 1, Zhiyuan Li 1, "Improv. Adsorpt. Capacit. Appl. Temp. gaseous PbCl₂ capture by Modif. montmorillonite with Comb. Therm. Treat. acid Act", vol. 3, 2023, doi: 10.1016/j.chemosphere.2022.137466.
- [81] N. Ayawei, A. N. Ebelegi, and D. Wankasi, "Modelling and Interpretation of Adsorption Isotherms," *J. Chem.*, vol. 2017, 2017, doi: 10.1155/2017/3039817.
- [82] S. P. Santoso *et al.*, "One-step synthesis of nitrogen-grafted copper-gallic acid for enhanced methylene blue removal," *Sci. Rep.*, vol. 11, no. 1, pp. 1–14, 2021, doi: 10.1038/s41598-021-91484-w.
- [83] A. Özcan, A. S. Özcan, S. Tunali, T. Akar, and I. Kiran, "Determination of the equilibrium, kinetic and thermodynamic parameters of adsorption of copper(II) ions onto seeds of *Capsicum annum*," *J. Hazard. Mater.*, vol. 124, no. 1–3, pp. 200–208, 2005, doi: 10.1016/j.jhazmat.2005.05.007.
- [84] M. Kebir, M. Trari, R. Maachi, N. Nasrallah, and A. Amrane, "Valorization of *Inula viscosa* waste extraction, modeling of isotherm, and kinetic for the tartrazine dye adsorption," *Desalin. Water Treat.*, vol. 54, no. 10, pp. 2806–2816, 2015, doi: 10.1080/19443994.2014.905976.