

BAB XII

DISKUSI DAN KESIMPULAN

XII.1. Diskusi

Pabrik etilen glikol didirikan karena kebutuhan pasar dalam negeri serta luar negeri yang meningkat dari tahun ke tahun. Etilen glikol di Indonesia mayoritas masih impor dari negara lain untuk memenuhi kebutuhan sedangkan pabrik etilen glikol hanya ada satu untuk saat ini. Pendirian pabrik etilen glikol ini diharapkan dapat memenuhi kebutuhan dalam negeri serta meningkatkan lapangan pekerjaan bagi masyarakat Indonesia

Kelayakan pabrik etilen glikol dengan bahan baku ampas tebu dapat dilihat dari beberapa faktor sebagai berikut:

- Aspek bahan baku

Ampas tebu digunakan sebagai bahan baku dalam pembuatan etilen glikol dengan katalis nikel tungsten karbit. Ampas tebu dipilih sebagai bahan baku karena jumlah yang melimpah serta masih jarang untuk dimanfaatkan kembali.

- Aspek proses dan produk

Selulosa didapatkan dari ampas tebu yang melalui proses *wet oxidation* dan delignifikasi. Kemudian selulosa dilakukan proses hidrolisis hidrogenasi menjadi etilen glikol dan beberapa produk samping. Etilen glikol didapatkan dengan memurnikan hasil hidrolisis hidrogenasi dengan distilasi dan kristalisasi. Etilen glikol yang dihasilkan memiliki kemurnian sebesar 99,%% yang digunakan pada berbagai industri seperti industri PET, dan tekstil.

- Aspek lokasi

Pabrik etilen glikol ini didirikan di Desa Brumbungan Lor, Kecamatan Gending, Kabupaten Probolinggo, Jawa Timur. Lokasi ini dipilih karena meninjau dari beberapa aspek seperti ketersediaan bahan baku, logistik, sumber daya manusia serta letak geografis.

- Aspek ekonomi

Pabrik etilen glikol ini memiliki kelayakan dari segi ekonomi yang didasari dari hasil perhitungan dengan metode *discounted cash flow*. Hasil analisa tersebut menunjukkan:

- Laju Pengembalian Modal Investasi (ROR) sesudah pajak memiliki nilai sebesar 9,15% yang mana lebih besar dari suku bunga bank yaitu 5,75%
- Laju Pengembalian Modal Ekuitas (ROE) sesudah pajak memiliki nilai sebesar 18,80% yang mana lebih besar rata-rata ROE yaitu 10%
- Waktu Pengembalian Modal (POT) sesudah pajak diperkirakan selama 5 tahun 7 bulan yang mana lebih kecil dari umur pabrik yaitu 10 tahun.
- Titik impas (BEP) memiliki nilai sebesar 41,12% yang mana masuk pada rentang BEP ideal yaitu 40%-60%.

XII.2. Kesimpulan

Pabrik	:	Etilen Glikol (EG)
Kapasitas	:	100.000 ton/tahun
Bahan baku	:	Ampas tebu
Sistem operasi	:	Semi-batch
Utilitas	:	
• Air	:	Air sanitasi = 12 ton/hari Air proses = 48.378,73 ton/hari Air pendingin = 34.937,71 ton/hari Air umpan boiler = 2612,49 ton/hari
• Listrik	:	41.241,9 kW
• Bahan bakar	:	Metana dan etana
Jumlah tenaga kerja	:	200 orang
Lokasi pabrik	:	Desa Brumbungan Lor, Kecamatan Gending, Kabupaten Probolinggo, Jawa Timur

Analisa ekonomi dengan metode *Discounted Cash Flow*:

- *Rate of Return* (ROR) sebelum pajak : 15,03%
- *Rate of Return* (ROR) sesudah pajak : 9,15%
- *Rate of Equity* (ROE) sebelum pajak : 30,11%

- *Rate of Equity* (ROE) sesudah pajak : 18,80%
- *Pay Out Time* (POT) sebelum pajak : 4 tahun 7 bulan
- *Pay Out Time* (POT) sesudah pajak : 5 tahun 7 bulan
- *Break Even Point* (BEP) : 41,12%

DAFTAR PUSTAKA

- Ag Eng, M. J., Shal, E., Tawfik, M., El Shal, A., & Metwally, K. A. (2010). FARM MACHINERY AND POWER STUDY THE EFFECT OF SOME OPERATIONAL FACTORS ON HAMMER MILL (Vol. 27, Issue 1).
- Badan Pusat Statistik. (2023). Impor Etilen Glikol. <https://www.bps.go.id/>
- Chemeo. (2023). Chemeo: High Quality Chemical Properties. chemeo.com
- Chen, Y.-W., & Lee, D.-S. (2021). One Pot Synthesis of Bio-Ethylene Glycol from Cellulose. *Modern Research in Catalysis*, 10, 93–101. <https://doi.org/10.4236/mrc.2021.104006>
- Chindaprasirt, P., & Rattanasak, U. (2020). Eco-production of silica from sugarcane bagasse ash for use as a photochromic pigment filler. *Scientific Reports*, 10(1). <https://doi.org/10.1038/s41598-020-66885-y>
- Couper, J. R., Penney, W. R., & Fair, J. R. (2012). *Chemical Process Equipment: Selection and Design* (3rd ed.). Elsevier.
- Fuller, H. C. (1924). Ethylene Glycol-Its Properties and Uses' (Vol. 16, Issue 6).
- Garg, U., Lowry, J., & Adam Algren, D. (2019). Ethylene glycol and other glycols: Analytical and interpretation issues. In *Critical Issues in Alcohol and Drugs of Abuse Testing* (pp. 59–69). Elsevier. <https://doi.org/10.1016/B978-0-12-815607-0.00005-8>
- Geankoplis, C. J. (2018). *Transport Processes and Separation Process Principles* (5th ed., pp. 1082–1085). Pearson Education, Inc.
- Halperin, M. L., Kamel, K. S., & Goldstein, M. B. (2010). Metabolic Acidosis: Acid Gain Types. In *Fluid, Electrolyte and Acid-Base Physiology* (pp. 160–192). Elsevier. <https://doi.org/10.1016/b978-1-4160-2442-2.00006-x>
- Han, Y., Bai, Y., Zhang, J., Liu, D., & Zhao, X. (2020). A comparison of different oxidative pretreatments on polysaccharide hydrolyzability and cell wall structure for interpreting the greatly improved enzymatic digestibility of sugarcane bagasse by delignification. *Bioresources and Bioprocessing*, 7(1). <https://doi.org/10.1186/s40643-020-00312-y>
- Ho, N. W. Y., Ladisch, M. R., Sedlak, M., Mosier, N., & Casey, E. (2011). Biofuels from Cellulosic Feedstocks. In *Comprehensive Biotechnology, Second Edition* (Vol. 3, pp. 51–62). Elsevier Inc. <https://doi.org/10.1016/B978-0-08-088504-9.00155-0>
- Hua, T.-C., Liu, B.-L., & Zhang, H. (2010a). Protective Agents and Additives for Freeze-drying of Pharmaceutical Products. In *Freeze-Drying of Pharmaceutical and Food Products* (pp. 170–186). Elsevier.

- Hua, T.-C., Liu, B.-L., & Zhang, H. (2010b). Protective Agents and Additives for Freeze-drying of Pharmaceutical Products. In *Freeze-Drying of Pharmaceutical and Food Products* (pp. 170–186). Elsevier. <https://doi.org/10.1533/9781845697471.170>
- Ji, N., Zhang, T., Zheng, M., Wang, A., Wang, H., Wang, X., & Chen, J. G. (2008). Direct catalytic conversion of cellulose into ethylene glycol using nickel-promoted tungsten carbide catalysts. *Angewandte Chemie - International Edition*, 47(44), 8510–8513. <https://doi.org/10.1002/anie.200803233>
- Kalász, H., Báthori, M., & Valkó, K. L. (2020). Basis and pharmaceutical applications of thin-layer chromatography. In *Handbook of Analytical Separations* (Vol. 8, pp. 523–585). Elsevier B.V. <https://doi.org/10.1016/B978-0-444-64070-3.00010-2>
- Karp, S. G., Woiciechowski, A. L., Soccol, V. T., & Soccol, C. R. (2013). Pretreatment strategies for delignification of sugarcane bagasse: A Review. *Brazilian Archives of Biology and Technology*, 56(4), 679–689. <https://doi.org/10.1590/S1516-89132013000400019>
- Kementerian Perindustrian. (2016). Produksi Etilen Terus Didorong. <https://kemenperin.go.id/artikel/18671/Produksi-Etilen-Terus-Didorong>
- Keskin, T., Nalakath Abubackar, H., Arslan, K., & Azbar, N. (2019). Biohydrogen Production From Solid Wastes. In *Biomass, Biofuels, Biochemicals: Biohydrogen, Second Edition* (pp. 321–346). Elsevier. <https://doi.org/10.1016/B978-0-444-64203-5.00012-5>
- Kim, J. S., Lee, Y. Y., & Kim, T. H. (2016). A review on alkaline pretreatment technology for bioconversion of lignocellulosic biomass. *Bioresource Technology*, 199, 42–48. <https://doi.org/10.1016/J.BIORTECH.2015.08.085>
- Manickavasagan, A., Thangavel, K., Dev, S. R. S., Delfiya, D. S. A., Nambi, E., Orsat, V., & Raghavan, G. S. V. (2015). Physicochemical Characteristics of Date Powder Produced in a Pilot-Scale Spray Dryer. *Drying Technology*, 33(9), 1114–1123. <https://doi.org/10.1080/07373937.2015.1014045>
- Marques, C., Tarek, R., Sara, M., & Brar, S. K. (2016). Sorbitol Production From Biomass and Its Global Market. In *Platform Chemical Biorefinery: Future Green Chemistry* (pp. 217–227). Elsevier. <https://doi.org/10.1016/B978-0-12-802980-0.00012-2>
- Martín, C., Klinke, H. B., & Thomsen, A. B. (2007). Wet oxidation as a pretreatment method for enhancing the enzymatic convertibility of sugarcane bagasse. *Enzyme and Microbial Technology*, 40(3), 426–432. <https://doi.org/10.1016/j.enzmictec.2006.07.015>
- McCabe, W. L., Smith, J. C., & Harriott, P. (1993). *Unit Operations of Chemical Engineering* (5th ed.). McGraw-Hill.

- Mkhize, T., Debra Mthembu, L., Gupta, R., Kaur, A., Chander Kuhad, R., Reddy, P., & Deenadayalu, N. (2016). Saccharification of bagasse. In *BioResources* (Vol. 11, Issue 3).
- Mudgil, D. (2017). The Interaction Between Insoluble and Soluble Fiber. In *Dietary Fiber for the Prevention of Cardiovascular Disease: Fiber's Interaction between Gut Microflora, Sugar Metabolism, Weight Control and Cardiovascular Health* (pp. 35–59). Elsevier Inc. <https://doi.org/10.1016/B978-0-12-805130-6.00003-3>
- Pang, J., Zheng, M., Wang, A., & Zhang, T. (2011). Catalytic hydrogenation of corn stalk to ethylene glycol and 1,2-propylene glycol. *Industrial and Engineering Chemistry Research*, 50(11), 6601–6608. <https://doi.org/10.1021/ie102505y>
- Perry, S., Perry, R. H., Green, D. W., & Maloney, J. O. (2000). Perry's chemical engineers' handbook. In *Choice Reviews Online* (Vol. 38, Issue 02). <https://doi.org/10.5860/choice.38-0966>
- Peters, M. S., Timmerhaus, K. D., & West, R. E. (2003). *Plant Design and Economics for Chemical Engineers*. McGraw-Hill Education.
- Polychem Indonesia. (2020). Produk. <https://polychemindo.com/id/produk>
- Rodella, C. B., Barrett, D. H., Moya, S. F., Figueroa, S. J. A., Pimenta, M. T. B., Aprígio, A., Curvelo, S., & Teixeira Da Silva, V. (2013). Physical and Chemical Studies of Tungsten Carbide Catalysts: Effects of Ni Promotion and Sulphonated Carbon. www.rsc.org/advances
- Scapini, T., dos Santos, M. S. N., Bonatto, C., Wancura, J. H. C., Mulinari, J., Camargo, A. F., Klanovicz, N., Zobot, G. L., Tres, M. v., Fongaro, G., & Treichel, H. (2021). Hydrothermal pretreatment of lignocellulosic biomass for hemicellulose recovery. *Bioresource Technology*, 342, 126033. <https://doi.org/10.1016/J.BIORTECH.2021.126033>
- Sinnot, R. K., Coulson, J. M., & Richardson, J. F. (2005). *Chemical Engineering Design* (4th ed., Vol. 6). Elsevier Butterworth-Heinemann.
- Solarte-Toro, J. C., Romero-García, J. M., Martínez-Patiño, J. C., Ruiz-Ramos, E., Castro-Galiano, E., & Cardona-Alzate, C. A. (2019). Acid pretreatment of lignocellulosic biomass for energy vectors production: A review focused on operational conditions and techno-economic assessment for bioethanol production. *Renewable and Sustainable Energy Reviews*, 107, 587–601. <https://doi.org/10.1016/J.RSER.2019.02.024>
- Statista. (2021). Production Capacity of Ethylene Glycol Worldwide from 2014 to 2025. <https://www.statista.com/statistics/1067418/global-ethylene-glycol-production-capacity/>
- Te Molder, T. D. J., Kersten, S. R. A., Lange, J. P., & Ruiz, M. P. (2021). Ethylene Glycol from Lignocellulosic Biomass: Impact of Lignin on Catalytic Hydrogenolysis. *Industrial and Engineering Chemistry Research*, 60(19), 7043–7049. <https://doi.org/10.1021/acs.iecr.1c01063>

- Trisnani, I. A. (2015). Prarancangan Pabrik Etilen Glikol dari Etilen Oksida dan Air dengan Proses Hidrasi Non Katalitik Kapasitas 230.000 Ton/Tahun.
- Ulrich, G. D. (1984). *A Guide to Chemical Engineering Process Design and Economics*. Wiley.
- Vale, J. A., Bluett, N. H., & Widdop, B. (1976). Ethylene glycol poisoning. *Postgraduate Medical Journal*, 52(611), 598–602. <https://doi.org/10.1136/pgmj.52.611.598>
- Wang, T., Li, X., & Dong, J. (2020). Ethylene Glycol Purification by Melt Crystallization: Removal of Short-Chain Glycol Impurities. *Industrial and Engineering Chemistry Research*, 59(18), 8805–8812. <https://doi.org/10.1021/acs.iecr.0c00347>
- Yaws, C. L. (2006). *Yaws Handbook of Thermodynamic Properties*. Elsevier Science.
- Yue, H., Zhao, Y., Ma, X., & Gong, J. (2012). Ethylene glycol: Properties, synthesis, and applications. *Chemical Society Reviews*, 41(11), 4218–4244. <https://doi.org/10.1039/c2cs15359a>
- Zevilla, M. F., Nugroho, W. A., Djojowasito, G., Keteknikan, J., Teknologi, P.-F., Brawijaya, P.-U., Veteran, J., & Korespondensi, P. (2015). Pengukuran Efektivitas Mesin Rotary Vacuum Filter dengan Metode Overall Equipment Effectiveness. In *Jurnal Keteknikan Pertanian Tropis dan Biosistem* (Vol. 3, Issue 3).
- Zhang, T. (2010). Tungsten Carbide Catalyst.
- Zheng, M., Pang, J., Wang, A., & Zhang, T. (2014). One-pot catalytic conversion of cellulose to ethylene glycol and other chemicals: From fundamental discovery to potential commercialization. In *Cuihua Xuebao/Chinese Journal of Catalysis* (Vol. 35, Issue 5, pp. 602–613). Science Press. [https://doi.org/10.1016/s1872-2067\(14\)60013-9](https://doi.org/10.1016/s1872-2067(14)60013-9)
- Zhou, Z., Ouyang, D., Liu, D., & Zhao, X. (2023). Oxidative pretreatment of lignocellulosic biomass for enzymatic hydrolysis: Progress and challenges. *Bioresource Technology*, 367, 128208. <https://doi.org/10.1016/J.BIORTECH.2022.128208>
- Zumbé, A., Lee, A., & Storey, D. (2001). Polyols in confectionery: the route to sugar-free, reduced sugar and reduced calorie confectionery. *British Journal of Nutrition*, 85(S1), S31–S45. <https://doi.org/10.1079/bjn2000260>