# **BUKTI KORESPONDENSI**

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Effect of Processing on Bioactive Compounds, Antioxidant Activity, Physicochemical, and Sensory Properties of Orange Sweet Potato, Red Rice, and Their Application for Flake Products

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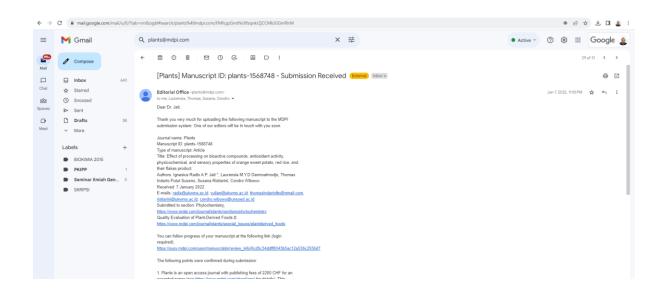
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# Effect of processing on bioactive compounds, antioxidant activity, physicochemical, and sensory properties of orange sweet potato, red rice, and their application for flakes product

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Abstract: Orange sweet potato (OSP) and red rice (RR) are rich source of health benefits associated 13 substances and can be conventionally cooked or developed into food product. This research ap-14 proach was to closely monitor the changes of bioactive compounds and their ability as antioxidants 15 from the native form to the food products which are ready to be consumed. Moreover, this research 16 explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their de-17 veloped flake product, and also investigated their antioxidant activity, physicochemical properties, 18 and sensory properties. Simultaneous identification using liquid chromatographic method show 19 that OSP, RR, and their flake product have significant amount of  $\beta$  and  $\alpha$  carotene,  $\beta$ -cryptoxan-20 thine, and also  $\alpha$  and  $\gamma$  -tocopherol. Different response was observed on the bioactive compound 21 and antioxidant activity affected yg heating process. Meanwhile, OSP and RR can be combined to 22 formed a promising flake breakfast cereal products as shown from the physicochemical analysis 23 such as moisture and dietary fiber contents, water absorption index, fracturability, crispness, and 24 color. Those quality parameters were affected by the proportions of OSP and RR in the flake prod-25 ucts. Moreover, the preference scores (n=120 panelists) for the flakes ranged from slightly liked to 26 indifferent. It can be concluded that OSP and RR are potential sources of bioactive compounds 27 which could act as antioxidant and could be developed into flake product that meet the dietary and 28 sensory needs of consumers. 29

**Keywords:** orange sweet potato; red rice; flakes; breakfast cereal; bioactive compound; antioxidant 30 activity; physicochemical; sensory properties 31

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# 1. Introduction

Modern food trends and lifestyle changes have strongly influenced the dietary habits 34 of society. The demands for ready-to-eat and simple-to-prepare foods are increasing rap-35 idly, providing an excellent opportunity for food industries to play a significant role in 36 supplying such food products. Flakes, one of the most popular foods made from cereals, 37 typically oat, corn, and barley, are commonly served for breakfast with milk in Europe 38 and the USA, and their global appeal is gaining traction. Besides, healthy eating has be-39 come a new trend in modern culture, with consumers increasingly opting for healthy 40 foods options. Secondary metabolites found in have been shown to decrease the risks of 41 degenerative diseases such as coronary heart disease, diabetes, cancer, and stroke [1-4]. 42 Thus, innovative functional food products rich in bioactive compounds are needed to help 43

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promote optimal health and reduce disease risks. Asia has the most rapidly growing food44product market, so the use of local ingredients can help open a new market for flakes and45decrease the reliance on imported foods such as oat and barley. Commodities that can46potentially be developed into flakes are red rice and orange sweet potato.47

Red rice (Oryza nivara L.) is a variety of rice with red pericarp caused by anthocyanin 48 in the aleurone layer. It is a rich source of anthocyanins such as cyanidin 3-O-glucoside 49 and peonidin 3-O-glucoside [5]. Anthocyanin has also been reported to inhibit plaque for-50 mation [6] and to exhibit hypocholesterolemic [7] and anticancer effects [8] in red rice. The 51 health benefits of red rice have also been linked to bioactive compounds such as tocoph-52 erol and tocotrienols [9,10], and dietary fiber [11]. Red rice also contains higher essential 53 minerals, including iron, zinc, and vitamins, which are especially important for babies and 54 toddlers, than white rice [12]. Despite the touted health benefits of red rice, its consump-55 tion remains low. It is generally considered inferior to white rice due to the hard texture 56 and unpleasant aroma when cooked. Traditionally, red rice is consumed steamed or 57 boiled, and its sensory properties are inferior to white rice. The most popular red rice-58 based product is red rice baby porridge, but there are reports on the development of red 59 rice-based products such as pasta [13], noodles [14], flakes [15], rice milk [16], and fer-60 mented beverages [17]. However, these have not been scaled up or commercially estab-61 lished. 62

Orange sweet potato (*Ipomoea batatas*) is one variety of sweet potato with a bright 63 orange flesh color caused by carotenoids, of which high amounts of  $\beta$  carotene,  $\alpha$  carotene, 64 and  $\beta$  cryptoxanthin have been reported [18–20]. In many countries, orange sweet potato 65 has been used to eradicate vitamin A deficiency due to its high content of beta carotene, a 66 pro-vitamin A carotenoid. Sweet potato was extensively promoted in Africa and some 67 Asian countries with remarkable results [21–23]. However, there were difficulties in en-68 suring its sustainability due to the monotonous way of its preparation, mainly boiled and 69 baked, even though the essential nutritional components were reportedly retained after 70 processing [24]. Processing orange sweet potato could decrease the beta carotene, but not 71 below the recommended dietary level [25]. Moreover, food prepared from orange sweet 72 potato by the traditional method was not attractive to children, who were the main target 73 of the vitamin A intake enhancement. Therefore, innovative food products need to be de-74 veloped to promote the consumption of red rice and orange sweet potato. Numerous re-75 search has been published to explore the potency of various plant sources as functional 76 foods. However, the approach mainly investigates the plant materials in the native or raw 77 form. In contrast, the consumer will generally consume after the materials undergo trans-78 formations which could affect the characteristics of the products. Moreover, besides the 79 processing, the bioactive compounds and antioxidant activity will be further affected by 80 the in vivo digestion and the intestinal absorption rate of the body metabolism before 81 providing bioavailable compounds that can be used. This research approach is to closely 82 monitor the changes of the bioactive compound and antioxidant activity from the raw 83 materials to the ready to be consumed food products and aimed to investigate the bioac-84 tive compounds and antioxidant activity of raw and cooked red rice and orange sweet 85 potato and the physicochemical and sensory properties of their developed flake products. 86

# 2. Materials and Methods

#### 2.1. Plant materials and Chemicals

A local variety of red rice (RR) (*Oryza nivara* L.), "Cempo abang," and orange sweet potato 89 (OSP) (*Ipomoea batatas* L.), "Mendut," were collected from farmers in Yogyakarta province, 90 Indonesia. Chemicals used for analysis, including aquadest, Folin Ciocalteu, 1,1-diphenyl-2-picrylhydrazyl (DPPH), gallic acid, butylated hydroxyl toluene (BHT), enzymes 92 (thermamyl, pancreatin, pepsin), riboflavin, methionine, and nitroblue tetrazolium (NBT), 93 were purchased from Sigma Chemical. Methanol, Whatman 40 filter paper, n-hexane, 94 NaOH, HCl, ethanol, and phosphate buffer (pH 6) were purchased from Merck, Germany.

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#### The red rice samples were washed, drained, and blended (Philips food processor). 100 The sweet potato samples were chopped into small pieces. All samples were then freeze-101 dried, refined, and sieved (30 mesh). Finally, the powdered samples were placed in dark 102 bottles and stored in a refrigerator (4°C) for further usage. 103

Carotenoid standards, including  $\beta$  and  $\alpha$ -carotene,  $\beta$ -cryptoxanthin, lycopene, and lutein,

and  $\alpha$  and  $\gamma$ -tocopherol standards were obtained from Sigma Aldrich.

# 2.2.2. Boiled samples

2.2. Sample preparation

2.2.1. Raw samples

Red rice (75 g) was cooked with 135 g of tap water (1:1.8; w/w) in a rice cooker (Pana-105 sonic) for approximately 45 min. After, the cooked rice was cooled for 10 min. Meanwhile, 106 150 g of orange sweet potato was boiled in an aluminum pot using tap water for 20 min, 107 cooled for 10 min, and then mashed. Both samples were freeze-dried, refined, sieved (30 108 mm mesh), and stored (4°C) in a refrigerator. 109

# 2.2.3. Flakes production

Red rice was placed in a cabinet dryer (60°C) for 1 h. The orange sweet potato was 111 peeled, sliced and placed in a cabinet dryer (60°C) for 6 h. The dried orange sweet potato 112 and red rice were mashed using a blender. The flour was passed through an 80 mm mesh. 113 Flakes were produced using six different proportions of OSP and RR, namely 100:0, 80:20, 114 60:40, 40:60, 20:80, and 0:100. Salt (3% w/w), sugar (30% w/w), and water (150% w/w) were 115 mixed in as additional ingredients. The mixture was heated at 75°C for 1 min and pressed 116 at 170°C for 1 min using a customized flake pressing tool. The pressed flakes were cut at 117 2x2 cm and dried using an oven at 125°C for 5 min. 118

# 2.3. Bioactive compounds and antioxidant activity analysis

# 2.3.1. Methanolic extract of samples

The extraction of samples (raw, boiled, and flakes) was done according to a previ-121 ously published procedure [26]. Briefly, 1 g of sample was weighed, ground, placed in a 122 centrifuge tube, and then extracted with 10 mL of 1% methanol-HCl solution. The mixture 123 was then vortexed for 15 min, centrifuged at 5000 rpm for 15 min, filtered (Whatman No. 124 40), and then used for antioxidant activity analysis, performed in triplicate. 125

# 2.3.2. Phenolic content

The total phenolic content was determined using the Folin Ciocalteau method by Sin-127 gleton and Rossi as described in other published work [27]. In brief, 0.1 ml of extract was 128 mixed with 0.5 ml 1:1 Folin Ciocalteu reagent and aquadest. After 10 min, 4.5 ml of 2% 129 sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>) was added, and the mixture was then vortexed and kept in the dark for 1 h. The blue complex formed was measured using a spectrophotometer at 131 765 nm. Methanol and catechin were used as the blank and standard, respectively. The 132 results were calculated as g catechin equivalents/100 g dry weight. 133

# 2.3.3. Anthocyanin Content

The total anthocyanin in red rice and flake samples was determined spectrophoto-135 metrically using the pH differential method [28]. In brief, 1 mL of extract was diluted in 136 pH 1.0 and pH 4.5 buffers. The absorbance was measured at 510 and 710 nm. The final 137 absorbance was calculated using the formula: 138

#### A = [(A513-A700)pH 1.0 - (A513-A700)pH 4.5)] (1)139

and converted to total g of anthocyanins per 100 g dry weight, with a molar extinction 140coefficient of 26,900 and a molecular weight of 445. 141

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# 2.3.4. Carotenoid and tocopherol analysis

The carotenoids and tocopherols in the raw and boiled samples of red rice and orange 143 sweet potato were determined simultaneously using High-Performance Liquid Chroma-144 tography (HPLC) based on previously published research [24]. In brief, 0.3 g of finely 145 ground samples were extracted with 0.5 mL of 70% ethanol and 0.4mL of n-hexane under 146 yellow lights. In addition, 0.4 mL  $\beta$ -Apo-8'carotenal-O-methyloxim and 0.4 mL  $\alpha$ ,  $\gamma$  to-147 copherol were used as an internal standard for carotenoids and tocopherols, respectively. 148 The mixture was shaken for 20 min, centrifuged at 5000 rpm, 4°C for 20min. Next, the 149 upper layer of extract containing a hexane fraction was collected using a micropipette. 150 The extraction was repeated four times using only n-hexane as a solvent. Finally, the hex-151 ane fractions were pooled and completely dried using nitrogen gas. 152

Before injection, the extracts were reconstituted in 200µl of ethanol containing 153 30µg/ml BHT, then 20µl of the reconstituted extracts were injected into the HPLC (Varian 154 Pro Star 410, Spark, Holland). A mixture of 82% acetonitrile, 15% dioxan, 3% methanol, 155 0,1M ammonium acetate, and 0.1% triethylamine was assigned as the mobile phase and 156 was pumped at a rate of 1.6 ml/min. The solvent was pre-mixed to avoid dependency on 157 reproducible mixing by the pump. For separation, a C18 Spherisorb ODS 2 column (3 µm, 158 250 × 4.6 mm) was applied. In addition, a UV Vis detector at 450nm and a Scanning Fluo-159 rescence detector using an excitation wavelength of 295nm and an emission wavelength 160 of 328nm were used to monitor the carotenoids and tocopherols, respectively. 161

# 2.3.5. DPPH radical scavenging activity

The radical scavenging activity of the extract was examined by the DPPH method 163 [25]. In brief, 1 mL of extract was mixed with 2 mL of 0,2 M DPPH and 2 mL of methanol 164 in centrifuge tubes, vortexed, and kept in the dark for 1 h. The absorbance was measured 165 spectrophotometrically at 517 nm. As a control, 150 ppm BHT solution was used. The 166 DPPH radical scavenging activity of the extract was expressed as a percentage calculated 167 as follows: % radical scavenging capacity = ((Absorbance of control – Absorbance of sample)/Absorbance of control) \* 100% 169

# 2.3.6. FRAP Assay

The ferric reducing antioxidant power (FRAP) was examined based on a previously 171 published report [29]. In brief, a mixture of 60  $\mu$ l extract, 180  $\mu$ l distilled water, and 1.8 ml 172 FRAP reagent was vortexed and incubated at 37°C for 30 min. The spectrophotometer was 173 used to read the absorbance of the mixture at 593 nm. A standard curve was prepared 174 with Fe [II] (FeSO4.7H2O, 100–2000 mM) to calculate the reducing power. The result was 175 expressed as mmol Fe[II]/g. In addition, methanol was used for the reagent blank. 176

# 2.3.7. Superoxide Radical Scavenging Capacity

A previous report [30] was followed to examine the superoxide radical scavenging 178 capacity. Firstly, a reagent containing riboflavin, methionine, and NBT in 0.05 M phosphate buffer pH 7.8 was prepared. Then, 100  $\mu$ l of the extract was mixed with 4.9 ml of reagent and illuminated (20 W fluorescent lamp) at 25°C for 25 min. The absorbance was measured at 560 nm. 182

# 2.4. Physicochemical properties of flake products

### 2.4.1. Moisture and dietary fiber contents

The moisture content was measured thermogravimetrically [31]. In brief, 1 g of each 185 sample was dried at  $105 \pm 0.2$  °C to establish a constant mass. The analysis was performed 186 in triplicate. 187

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The water absorption index was examined according to previously published 189 method [32]. In brief, 5 g of flakes were placed in a 100 mL beaker, 30 mL of water at 30°C 190 was added. After 10 s of immersion, the flakes were dried. The water absorption index 191 was calculated using the formula: WAI = (wf -wi)/wi, where wi and wf are the initial and 192 final weight of the sample, respectively. 193

# 2.4.3. Fracturability and crispness

The fracturability and crispness of flakes were measured using TA-XT Plus Texture 195 Analyzer (Stable Micro Systems, UK) [33]. The probe used was a ¼ inch spherical stain-196 less-steel probe (P0.25S). The sample was placed on the sample holder, and then the probe 197 was moved down to press the sample. The results were obtained in the form of a graph 198 (force vs. time). The value of the y-axis at the graph's highest point is the maximum force 199 value that can be held by the sample, called the value of fracturability. Crispness can be 200 measured through changes in displacement distance during a drastic decline in the graph 201 pattern from the highest peak to the next peak point. 202

### 2.5. Sensory analysis

The sensory evaluation was conducted by 120 untrained panelists to determine the 204 level of consumer preference for the flakes with various proportions of OSP and RR. The 205 parameters tested were preferences for color, taste, crispness of flakes, and mouthfeel 206 when served with milk. The Hedonic Scale Scoring method (preference test) with a scale 207ranging from 1 (strongly disliked) to 7 (strongly liked) was used for the sensory test.

Samples of flakes for the color preference test were prepared in open white plastic 209 containers. Panelists were asked first to assess aspects of flakes' taste, color, and crispness 210 before serving with milk. The crispness was evaluated based on the panelist's preference 211 for the sound of flakes during biting. For the mouthfeel test, 5 g of flakes were prepared 212 in a small plastic container. Panelists were instructed to pour 10 mL of milk into the con-213 tainer and wait for 1 min. Then, they were asked to assess the mouthfeel of the flakes 214 based on preference level by filling the questionnaire sheet provided. 215

# 2.6. Statistical analysis

The data were statistically analyzed using Anova ( $\alpha = 5\%$ ) followed by Duncan's 217 Multiple Range Test (DMRT) on SPSS software version 19. Spider web chart analysis us-218 ing Microsoft Excel was used to determine the best proportion of OSPF and RRF in flakes 219 based on the panelists' preferences. 220

#### 3. Results

# 3.1. Bioactive compounds of orange sweet potato (OSP), red rice (RR), and their flake products

Table 1 shows the phenolic compound, anthocyanin, carotenoid, and tocopherol contents of raw and cooked OSP and RR and their flakes containing different proportions of 224 OSP and RR. The raw RR had a higher phenolic content than OSP with values of 225 301.89±24.86 mg GAE/100 g DW and 110.68±18.37 mg GAE/100 g DW, respectively. As a 226 result, the higher the proportion of RR, the higher the phenolic content of the developed 227 flakes. It was also found that cooking decreased the phenolic content of RR and OSP by approximately 49.34% and 41.08%, respectively. 229

Interestingly, the flakes with 100% RR showed a higher phenolic content 230 (162.40±21.54 mg GAE/100 g) than the cooked RR. Anthocyanin was only observed in RR, 231 8.81±0.05 mg/100 g DW for raw RR and 8.64±0.08 mg/100 g DW for cooked RR, and the 232 difference was not significant. Furthermore, flakes containing 100% RR had a lower an-233 thocyanin content (5.81±0.04 mg/100 g DW) than the conventionally cooked RR. 234

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# 222 223

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13,23±

1.1 a

 $2.40\pm$ 

0,2<sup>a</sup>

 $\alpha$ -tocopherol (µg/g)

 $\gamma$ -tocopherol ( $\mu$ g/g)

 $15.11 \pm$ 

0,5<sup>b</sup>

5.38±

0.05<sup>b</sup>

 $34.08 \pm$ 

2.2<sup>a</sup>

29.27 ±

2.4 a

57.65 ±

2.1<sup>b</sup>

 $40,11 \pm$ 

1.8<sup>b</sup>

	0	SP	R	R		Propo	ortions of O	SP and RR	in flakes	
	Raw	Cook	Raw	Cook	100:0	80:20	60:40	40:60	20:80	0:100
Phenolic (mg	110.68	65.21±	301.89±	152.91±	77.46±	97.34±	102.03±	131.79 ±	146.09±	162.40±
GAE/100 g DW)	±18.3ª	7.3 <sup>b</sup>	24.86 ª	28.92 <sup>b</sup>	8,28 ª	8.79 <sup>b</sup>	11.65 °	10.93 <sup>d</sup>	15.64 <sup>e</sup>	$21.54^{f}$
Anthocyanin			8,81±	8.64±			1.74±	2.09±	3.73±	5.81±
(mg/100g DW)	-	-	0.05 ª	0.08 a	-	-	0.06 c	0.05 <sup>d</sup>	0.07 <sup>e</sup>	$0.04{}^{\rm f}$
	278.58	134.17±	13.17 ±	7.37 ±	48,83±	36.27±	25.77±	27.23±	15.69±	3.12±
β-carotene (µg/g)	± 31.5 ª	17.2 <sup>ь</sup>	2.62 ª	0.5 <sup>b</sup>	3,31 ª	3.01 <sup>b</sup>	3.45 °	2.72 <sup>d</sup>	2.21 <sup>e</sup>	$0.66^{\text{ f}}$
	19.57 ±	23.83 ±	5.53 ±	11.66 ±	15.61±	11.82±	5.31±	2.53±	. 1	. 1
α-carotene (µg/g)	1.8 ª	1.6 <sup>b</sup>	1.4ª	1.5 <sup>b</sup>	1.44 ª	3.11 <sup>b</sup>	1.59 °	0.87 <sup>d</sup>	nd	nd
β-cryptoxanthine	4.83±	$4.48\pm$	3.67 ±	3.96 ±	2.64±	2.81±	2.97±	2.78±	2.77±	2.81±
(µg/g)	0.2 <sup>a</sup>	0.8 ª	2.15 ª	1.9ª	0,05 ª	0.13 <sup>b</sup>	0.08	0.12 <sup>c</sup>	0.25 °	0.16 <sup>b,c</sup>
To take ( a /a)	3.77±	3.81 ±	2.16 ±	1.82 ±	. 1	. 1	. 1	. 1	. 1	. 1
Lutein (µg/g)	0.8 <sup>a</sup>	0.7 ª	0.8 ª	0.5 <sup>b</sup>	nd	nd	nd nd	nd	nd	nd

. Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–f) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each category (OSP, RR, and Flakes)

 $4.58 \pm$ 

0.73<sup>a</sup>

nd

Of the carotenoids, OSF had a higher content of  $\beta$  carotene (278.58 ± 31.5 µg/g),  $\alpha$ -240 carotene,  $\beta$ -cryptoxanthin, and lutein.  $\beta$  carotene and  $\beta$ -cryptoxanthin were the dominant 241 carotenoids observed in RR. The cooking process significantly decreased the content of  $\beta$ 242 carotene in OSP and RR roughly by 48% and 56%, respectively. Overall, flakes containing 243 higher amounts of OSP showed higher carotenoid contents. The processing significantly 244 decreased the carotenoids in the flake products when considering the raw forms and the 245 proportions of OSP and RR, and  $\beta$  carotene was the major carotenoid remaining in the 246 flake products. 247

7.34±

1.49<sup>b</sup>

nd

10.51±

1.27 °

3.38±

1.22 °

12.45±

1.21 °

6.71±

1.19<sup>d</sup>

16.82±

0.52 e

 $8.06 \pm$ 

0.98 e

18.31±

 $0.77^{f}$ 

12.15±

0.73<sup>f</sup>

Moreover, raw contained higher  $\alpha$ -tocopherol (34,08 ± 2,2 µg/g) than OSP (13,23± 1,1 248 µg/g) in raw forms. Unlike the decreasing trend observed in other bioactive compounds 249 due to processing, the tocopherol content of both RR and OSP increased after conventional 250 cooking. On the other hand, the two-step thermal processing decreased the tocopherol 251 content of flakes. 252

# 3.2. Antioxidant activity of raw and cooked OSP and RR and their flake products

The antioxidant activity of raw and cooked OSP and RR and their flake products 254 were examined using DPPH, FRAP, and Superoxide radical scavenging activity methods. 255 Figure 1a shows the DPPH scavenging activity of methanolic extract of RR, OSP, and the 256 flake products. Boiling affected the ability of the methanolic extract to scavenge DPPH 257 radicals. Approximately 16% and 23% decreases were observed in cooked RR and OSP, 258 respectively. The combination of OSP and RR in the ratio of 60:40 resulted in flakes with 259 the highest antioxidant activity (84%). The results trend indicated that the higher propor-260 tion of OSP contributed to the more robust antioxidant capacity of the extract. Further-261 more, the DPPH result was in agreement with FRAP (Figure 1b) and Superoxide scaveng-262 ing capacity (Figure 1c). Thus, conventional cooking and flake processing methods reduce 263 the antioxidant activity of extracts of OSP and RR compared to their raw forms, and the 264

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activity.

was made within each category (OSP, RR, and Flakes).

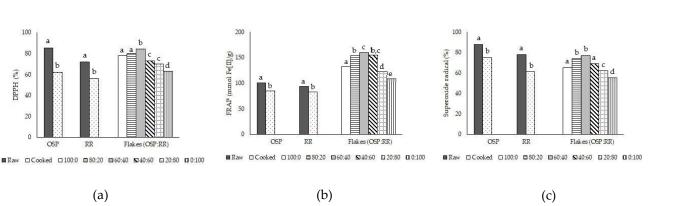


 Figure 1. Antioxidant activity of extract determined by (a) DPPH, (b) FRAP, and (c) Superoxide
 radical assays.
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 Different superscript letters (a–f) denote significantly different values according to Duncan's test (p < 0.05). Comparison</td>
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# 3.3. Physicochemical properties of OSP and RR-based flake products

The proportion of OSP and RR in the flake formulation affected the moisture content276of flakes. A lower OSP proportion, resulted in a lower moisture content of flakes (Table2772). The dietary fiber content increased with a higher proportion of RR. The dietary fiber278content of flake products ranged from  $13.86 \pm 0.73\%$  to  $9.47 \pm 0.01\%$ .279

right combination of OSP and RR in the flake formulation is critical for higher antioxidant

Table 2. The physicochemical properties of flakes produced from different ratios of OSP and RR.

			Proportions of C	SP and RR in fla	akes	
	100:0	80:20	60:40	40:60	20:80	0:100
Moisture content (%)	5.71 ± 0,07 <sup>a</sup>	$5.31 \pm 0.10^{\mathrm{b}}$	5.09 ± 0.06 °	$4.87 \pm 0.01$ <sup>d</sup>	$4.43 \pm 0.03$ <sup>e</sup>	$4.25 \pm 0.03$ f
Dietary fiber (%)	<b>9.47 ± 0,01</b> <sup>a</sup>	9.9 ± 0.02 <sup>b</sup>	10.9 ± 0.05 °	$11.63 \pm 0.34$ <sup>d</sup>	12,73 ± 0.26 <sup>e</sup>	$13.86 \pm 0.73$ f
Water absorption index	$1.69 \pm 0.03^{a}$	$1.14 \pm 0.02^{\mathrm{b}}$	$1.06 \pm 0.03$ <sup>c</sup>	$0.96 \pm 0.03$ <sup>d</sup>	$1.09 \pm 0.03$ b,c	$1.12 \pm 0.02^{b,c}$
Fracturability	$8.48 \pm 0.09$ a	$5.35 \pm 0.85$ b	$3.34 \pm 0.34$ c	$2.27 \pm 0.04$ d	$3.17 \pm 0.09$ °	$4.64 \pm 0.12^{\mathrm{f}}$
Crispness	$3.9 \pm 0.03^{a}$	$2.4\pm0.02^{\mathrm{b}}$	$1.5\pm0.02$ c	$1.9 \pm 0.03^{d}$	$3.21 \pm 0.05^{\text{ e}}$	$3.7 \pm 0.03$ f

Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a-f) denote significantly different values281according to Duncan's test (p < 0.05). Comparison was made within each row</td>282

The water absorption index of flakes was lowest at an OSP to RR ratio of 40:60, with  $0.96 \pm 0.03$  %, and was generally higher at combination ratios of 100:0, 80:20, 0:100, and 285 20:80. In addition, the texture characteristic of flakes was determined by the fracturability 286 and crispness. The highest fracturability value was found in flakes made from 100% OSP 287  $(8.48 \pm 0.09)$ . The reduction of the OSP proportion in flakes caused a decrease in fractura-288 bility until the proportion of 40:60 (2.27  $\pm$  0.04), beyond which the fracturability of flakes 289 increased. A similar trend was observed in the crispness value of flakes. The lowest crisp-290 ness value was detected in flakes with an OSP to RR ratio of 60:40, while higher values 291 were obtained at ratios of 100:0, 0:100, 20:80, and 80:20. 292

Furthermore, the color of flakes was affected by the color of OSP and RR. The color293profile of the flakes is shown in Table 3. Based on the results, *lightness* ranged between29444.0 and 52.7, *redness* was between 8.2 and 10.8, *yellowness* was between 5.9 and 16.5, the295*hue* was between 28.6476 and 63.5740, and *chroma* was between 12.3145 and 18.4700.296

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		Propor	tions of OSP a	nd RR in flakes	6	
	100:0	80:20	60:40	40:60	20:80	0:100
L	44.0±0.1	47.3±0.2	51.5±0.1	52.7±0.4	51.8±0.2	51.8±0.7
a*	8.2±0.3	8,5±0.3	8.9±0.4	9.4±0.4	10.2±0.6	$10.8 \pm 0.4$
b*	16.5±0.3	14.7±0.2	13.4±0.2	10.3±0.2	9.0±0.4	$5.9 \pm 0.4$
٥h	63.574	59.9622	56.4087	47.6158	41.4237	28.6476
С	18.47	16.9685	16.0703	13.898	13.56	12.3145

Table 3. Color profiles of flakes produced from different ratios of OSP and RR.

# 3.4. Sensory characteristics of flakes

A preference test was conducted to determine the sensory characteristics of the 304 flakes. The results are presented in Table 4. The highest level of color preference was found in flakes with OSP to RR ratios of 60:40, 40:60, and 20:80, while flakes containing 100% 306 OSP had the lowest level of acceptance for color. The preference scores for taste and crispness of the flakes with various proportions of OSP and RR were comparable. Flakes containing 100% OSP and 100% RR received the highest preference score for mouthfeel, with 309 the former having a significant edge. Therefore, mixing the OSP and RR lowers the mouthfeel acceptance of the flake products. 311

Table 4. Preference test of flakes produced from different ratios of OSP and RR.

	Proportions of OSP and RR in flakes					
	100:0	80:20	60:40	40:60	20:80	0:100
Color	3.35±1.42ª	$4.40 \pm 1.22^{b}$	5.14±1.12 <sup>c</sup>	4.93±1.21°	4.89±1.30°	4.30±1.12 <sup>b</sup>
Taste	$4.43 \pm 1.34^{a}$	$4.69 \pm 1.28^{ab}$	$5.08 \pm 1.18^{\circ}$	5.09±1.20 <sup>c</sup>	$5.05 \pm 1.26^{bc}$	$4.72 \pm 1.29^{\text{abc}}$
Crispness	4.76±1.16 <sup>b</sup>	$4.85 \pm 1.04^{b}$	$4.76 \pm 1.40^{b}$	$4.08 \pm 1.17^{a}$	$4.96 \pm 1.07^{b}$	$5.00 \pm 1.04^{b}$
Mouthfeel	5.41±0.91 <sup>d</sup>	5.04±1.18°	$4.84 \pm 1.31^{bc}$	3.91±1.59 <sup>a</sup>	4.66±1.32 <sup>b</sup>	5.05±1.03°

Data are presented as mean ± standard deviation. Different superscript letters (a-d) denote signifi-314 cantly different values according to Duncan's test (p < 0.05). Comparison was made within each 315 row 316

# 4. Discussion

Phenolic compounds are the most widely found bioactive compounds in plants. 318 Some are produced in response to stress conditions as a defense mechanism of the plant. 319 They have been extensively investigated due to their antioxidant activity and anti-degen-320 erative disease effects [34]. In this research, methanol-HCl (1%) was used because acidic 321 methanol can penetrate deeply into cells, disrupting the cell membrane. In addition, acidic 322 methanol can dissolve and stabilize polar compounds such as phenolics and anthocyanins 323 [35]. This research shows that red rice (RR) and orange sweet potato (OSP) are rich sources 324 of phenolic compounds. It has previously been reported that RR has a high content of 325 phenolic compounds [36], and that these compounds are primarily accumulated in the 326 aleurone layer and bran of RR [37]. Thus, the rice milling process to remove the husk plays 327 a vital role in preventing the loss of various beneficial compounds. Previously, it was 328

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suggested that ferulic acid, p-coumaric acid, and procatechuic acid are the most abundantly found phenolic compounds in RR [37]. 330

Here, cooking led to a 49% decrease in the phenolic compounds of RR. Heating of RR 331 generally destroys the structure of phenolic compounds by breaking the esterified and 332 glycosylated bonds, thus decreasing the quantified content of phenolic compounds in 333 cooked RR [38]. Our results agree with previous findings [39,40], which reported high 334 levels of total phenolic compounds, mostly gallic acid, chlorogenic acid, pro-catechuic 4-335 hydroxybenzoic acid, and salicylic acid, in different varieties of OSP. However, phenolic 336 contents were broken down by the heating process. Regarding the flake products, flakes 337 containing a higher proportion of RR showed a higher content of phenolic compounds. 338 Nevertheless, the processing lowered the phenolic compounds in flakes when compared 339 to raw OSP and RR. Boiling and baking have previously been reported to be responsible 340 for the loss of phenolic compounds of red rice-based products [41] 341

A similar trend was observed in the anthocyanin content of RR. The cooking process 342 resulted in a significant decrease in the anthocyanin content due to the unstable property 343 of anthocyanins when exposed to high temperatures [42]. Anthocyanins were only de-344 tected in RR in this research. Anthocyanins such as cyanidin 3 glucoside, delphinidin 3 345 glucoside, and peonidin are reported to have health-promoting properties [43]. Thus, ex-346 posure to high temperatures for extended periods should be avoided to reduce the risk of 347 anthocyanin breakdown. The anthocyanin content in flakes was lower than in the raw 348 samples, and increasing the proportion of RR resulted in a higher anthocyanin content of 349 flakes. The decrease in the anthocyanin content of flakes is possibly due to the heat treat-350 ment during flake production, typically involving two high temperature processing steps 351 of pregelatinization and flaking. It has been reported that the high-temperature treatment 352 used for food processing can lead to a reduction of the anthocyanin content [44] 353

Moreover, there is a growing research interest in the conversion of  $\beta$ -carotene and  $\alpha$ -354 carotene absorbed in the duodenum to retinol by intestinal enzymes [45]. The high rate of 355 vitamin A deficiency in the world and the detrimental effects caused by the condition 356 have necessitated the search for foods that can supply sufficient amounts of daily vitamin 357 A requirement. In this research, OSP had the highest content of  $\beta$ -carotene. However, boil-358 ing of OSP decreased the  $\beta$ -carotene by approximately 41% of the  $\beta$ -carotene available. 359 This phenomenon could be due to the thermal breakdown of  $\beta$ -carotene. Moreover, carot-360 enoids are well known as substances that are sensitive to light and high temperatures [47]. 361 Red rice contains different types of carotenoids [46]. Here, an increase in carotenoids after 362 boiling was found, which could be related to the thermal disruption of the protein-carot-363 enoid complex and the consequent release of carotenoids from the matrix. Similar findings 364 have been published [48]. Similar trends were also observed in the OSP and RR-based 365 flake products. The β-carotene was significantly lower the cooked sample compared to 366 the raw sample. The OSP proportion in the flake formulation affected the carotenoid con-367 tent. The higher the OSP proportion, the greater the carotenoid content of the flakes. Also, 368 the carotenoid contents of the flakes were significantly lower than the raw material. The 369 simultaneous heating process from pre-gelatinization to flake pressing could further 370 break down the carotenoids. This finding is supported by previously published work, 371 which shows that heat treatment during food processing is responsible for the loss of ca-372 rotenoids to degradation[49] 373

Vitamin E deficiency could lead to severe neurological problems. The main vitamin 374 E compounds are tocopherols, with both  $\alpha$  and  $\gamma$  tocopherol providing most vitamin E 375 activity. Both samples had high contents of  $\alpha$  and  $\gamma$  tocopherol. A significant increase 376 (74%) in  $\alpha$  tocopherol was found in OSP after boiling. This result indicates that heat treat-377 ment could be beneficial for the bioaccessibility of tocopherol. Furthermore, heat treat-378 ment can assist in the breakdown of complex foods. Thus, tocopherol can be quickly re-379 leased from its binding site. On the contrary, heat treatment was reported to reduce the 380 tocopherol content in corn [50]. The increase in tocopherol after boiling indicates that to-381 copherols in the sample are more stable to heat treatment than other foods. In this 382 research, RR had a considerably high tocopherol content. Therefore, boiling could have 383 released the tocopherols from their binding site, facilitating their extraction and detection. 384 Moreover, the structure of the rice grains could have played a role. Red rice has a compact 385 structure. Therefore, cooking could assist the extraction of tocopherols, increasing the ex-386 tractable tocopherol content, although some of the tocopherols might be lost to high tem-387 perature. The tocopherol contents of the different flake products were lower than the raw 388 and boiled OSP and RR. It could be due to the simultaneous or prolonged exposure to 389 heat treatment. Unlike the conventional method of cooking, which increases the tocoph-390 erol content, extended heating breaks the matrix structure flakes and significantly affects 391 the tocopherols. Thus, prolonged heating should be avoided in the processing of healthy 392 food products to retain their tocopherols. Heat treatment combined with mechanical treat-393 ment in specific conditions could release the tocopherols from the food matrix. However, 394 extended exposure will lead to the breakdown of carotenoids in the sample. 395

This research measured the antioxidant activity of raw and cooked 396 OSP and RR and their flake products using DPPH, FRAP, and Superoxide radical scav-397 enging capacity assays. The result showed that heat treatment was responsible for de-398 creasing the antioxidant activity of OSP, RR, and the flake products. A positive correlation 399 was observed between the reduction of bioactive compounds and antioxidant capacity. 400 Most bioactive compounds are heat sensitive, which influences their antioxidant activity. 401 A previous study reported that an increase in temperature accelerated the initiation of 402 oxidation, preventing antioxidant compounds from working optimally [51]. The bioactive 403 compounds were degraded, experiencing structural changes, and wholly transformed 404 into inactive substances. Nevertheless, due to the complex nature of antioxidant com-405 pounds in plants, their thermal stability varies. Some compounds such as procatechuic 406 acid, p-coumaric acid, and ferulic acid have high thermal stability, which facilitates their 407 extraction and antioxidant activity. The heat treatment process assists in the release of 408 such compounds without affecting their activity. Moreover, the flake products exhibited 409 lower antioxidant activity than the raw or boiled products. Based on the percentage values 410 of DPPH and Superoxide radical scavenging capacity and the content of Fe [II] formed, 411 flakes containing only OSP or RR had lower antioxidant activity. Interestingly, the com-412 bination of OSP and RR increased the antioxidant activity, probably due to the synergistic 413 effect of bioactive compounds from OSP and RR [52]. Even though the processing reduced 414 the antioxidant activity in boiled samples and flake products, the remaining antioxidant 415 activity was still considerably high. Therefore boiled OSP, RR, and their flake products 416 are a good source of bioactive compounds and antioxidants. 417

Regarding the physicochemical and sensory properties of the flakes, it was observed 418 that the moisture content of flake products was mainly influenced by their composition. 419 Starch is the dominant carbohydrate found in OSP and RR, and OSP has a lower amylose 420 content compared to RR. According to Wang et al. [53], the amylose content of OSP is 421 18.71%, while Markus et al. [54] reported that RR has 23% amylose content. Amylose is a 422 linear polymer of glucose, which forms starch. The higher the amylose content, the greater 423 the moisture content of the dough due to a higher capability to absorb water. The absorbed 424 water will promote dough gelatinization during heating. The water absorbed by the 425 dough will then evaporate during the flaking process due to the network's inability to 426 entrap water during the pre-gelatinization heat treatment. The high flaking temperature 427 will detach water from the matrix structure of the flakes, resulting in increased evapora-428 tion and a lower moisture content of the flake products. 429

The dietary fiber content of the flakes ranged between 9.47 and 13.86%, comparable 430 to values commonly found in breakfast cereals such as corn flakes, rice, quinoa, millet, 431 and amaranth flakes [55]. The proportion of RR affected the dietary fiber of flakes. The 432 fiber content of flakes was also influenced by the heating and pressing processes. Heat 433 treatment caused the degradation of the fiber matrix and the glycosidic bond. The degradation affected the solubility level of the fiber, i.e., the ratio between soluble and insoluble 435 fiber, thus resulting in the reduction of total fiber in the product.

Water absorption index (WAI) is a physical property associated with the ability of 437 flakes to absorb water molecules within a particular time. Absorbed water molecules 438 could be bound or detained in matrix pores of flakes. The water absorption index is crucial 439 because it is associated with the quality of the flakes. Consumers can experience the crispy 440 and crunchy sensation of the flakes after soaking in milk or water. In contrast, the higher 441 WAI is interrelated with the unwanted soggy texture of flakes. The result shows that WAI 442 was decreased as the proportion of RR increased. The WAI is influenced by the porosity 443 of the matrix on flakes, thickness, and hygroscopicity of flakes. In addition, the presence 444 of fiber and protein could assist the flakes in absorbing water into their structure. The 445 result shows that increasing the proportion of RR reduces the WAI due to a decrease in 446 hygroscopicity. OSP is a rich source of sugar; thus, it has higher hygroscopicity compared 447 to RR. In addition, the presence of RR affects the network construction of flakes by inhib-448iting the formation of the starch-protein structure, which could entrap gas. The compact 449 structure created by RR starch inhibits water absorption into the matrix of the product. 450 On the other hand, RR has higher fiber and protein contents, which help improve water 451 absorption [56]. Moreover, the suitable fragmentation of the amylose and amylopectin 452 chain in sweet potato could also affect the water absorption capacity. The heating pro-453 cesses such as roasting, flaking, and extrusion will induce starch fragmentation with suf-454 ficient water. The gelatinization process converts starch to a digestible material and plays 455 a vital role in determining the structural properties of flakes and their ability to absorb 456 moisture. The presence of RR in the flake dough disturbs the composition of starch, thus 457 inhibiting the flakes from forming a porous structure and affecting their water absorption 458 capacity. 459

Fracturability is a physical property related to deformation conditions when a spe-460 cific maximum force is applied. A higher fracturability value represents the ability of food 461 products to maintain their structure when force is applied. According to Table 2, flakes 462 with 100% OSP has a higher fracturability value compared to others as the homogenous 463 matrix of starch, protein, and fiber in OSP enables the interaction between the matrix and 464 water molecules, leading to the firm, sturdy and rigid texture of flakes. The rigid texture 465 is related to the evolution phase of starch from amorphous conditions, which indicates 466 complete disorganization of the crystalline structure of starch. Increasing the proportion 467 of RR in the flakes reduces the fracturability value. The mixture of OSP and RR decreases 468 the rigidity of the flakes due to the different structural properties of the two samples, cre-469 ating flakes that are susceptible to fracture. Crispness is a complex texture attribute be-470 cause it comprises a combination of sensory analysis, acoustical procedure, and instru-471 mental analysis. The instrumental analysis revealed that flakes with OSP to RR ratios of 472 100:0 and 0:100 had higher crispness values than others. Similarly to fracturability, the 473 crispness value decreased with an increasing RR proportion in the flakes. The mixture of 474 ingredients with different structural properties can affect the crispness value of flakes. 475

The color profile shown in Table 3 revealed that the color of flakes was affected by 476 the pigments in the raw materials used for their production. The red color of RR is asso-477 ciated with anthocyanins found in its bran layer. The yellowness of OSP is linked to ca-478rotenoids, primarily  $\beta$ -carotene. Previous research reported that carotenoids are easily ox-479 idized and undergo color degradation due to thermal treatment [58]. The hue results 480 showed a flake color range between yellow and red. The color of flakes was also affected 481 by the Maillard reaction product. The higher the Maillard reaction product, the darker the 482 appearance of flakes. 483

The sensory analysis involved a preference test of the color, taste, crispness, and mouthfeel. Flakes containing 100% OSP had the lowest color preference score, associated with the orange appearance, and the lowest brightness score. Most panelists perceived the dark orange color as less fresh, less attractive, and less tasty. The color preference was increased with the addition of RR, which also helped improve the product lightness and redness values. The panelists were mostly in favor of flakes that appeared brighter and reddish. The higher brightness level is attributed to the white endosperm color of red rice, 480

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while the redness is related to the anthocyanins in the bran of red rice. The mouthfeel 491 preference of flakes is associated with the water absorption index. The higher the ability 492 of flakes to absorb water, the greater the plasticizing effect due to the presence of more 493 hydrophilic components such as the phosphate monoester found in sweet potato starch 494 [59]. The panelists generally preferred flakes with soft mouthfeel. The taste and crispness 495 preferences for OSP and RR-based flakes were in the range of "indifferent" and "slightly 496 likes." Increasing the RR proportion in flakes decreases the bitterness intensity and in-497 creases the savory taste. The perception of savory taste is generally influenced by the 498 moisture content and the flavor of red rice. Niu et al. [60] suggests that sweet potato with 499 a low dry matter content has a bitter taste, and increasing the dry matter reduces the bit-500 terness. On the other hand, a decrease in invertase activity may support the bitter after-501 taste of the sweet potato. 502

# 5. Conclusions

Orange sweet potato and red rice are rich sources of bioactive compounds, especially 504  $\beta$  carotene, for OSP, and phenolic compounds and anthocyanins, for RR. The boiling pro-505 cess significantly decreased most of the bioactive compounds, except to copherols and  $\alpha$ -506 carotene. The level of bioactive compounds in the flake products was dependent on the 507 proportion of OSP and RR. Heat treatment resulted in a decrease in antioxidant activity, 508 even though the remaining activity was still considerably high. The mixture of OSP and 509 RR can produce flakes with low moisture and high fiber contents. The optimum flake 510 water absorption index, fracturability, and crispness were obtained by combining 40% 511 OSP and 60% RR. Moreover, the ratio of OSP and RR influenced the color and sensory 512 preferences of the panelists. 513

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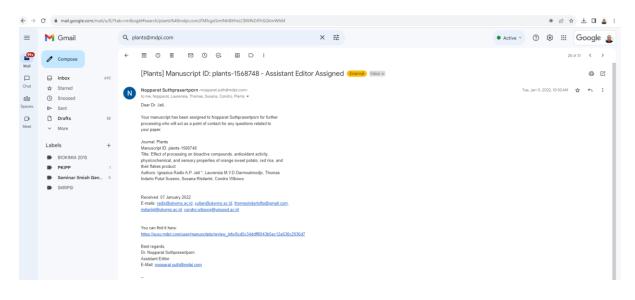
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 Bukti Assistant editor assigned. 11 Januari 2022



 Bukti konfirmasi review dan hasil review 28 januari 2022

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Change Password (/user/chgpwd)	Title	Effect of processing on bioactive compounds, antioxidant activity, physicochemical, and sensory properties of orange sweet potato, red rice, and their flakes product				
Edit Profile (/user/edit)	Authors	Ignasius Radix A.P. Jati * , Laurensia M.Y.D Darmoatmodjo , Thomas Indarto Putut Suseno ,				
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Submissions Menu 2	Special Issue	Quality Evaluation of Plant-Derived Foods II (https://www.mdpi.com/journal/plants/special_issues/plantderived_foods)				
Submit Manuscript (/user/manuscripts/upload)		Orange sweet potato (OSP) and red rice (RR) are rich source of health benefits associated substances and can be conventionally cooked or developed into food product. This research				
Display Submitted Manuscripts (/user/manuscripts/status)		explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their developed flake product, and also investigated their antioxidant activity, physicochemical properties, and sensory properties. Simultaneous identification using liquid chromatographic				
Display Co-Authored Manuscripts (/user/manuscripts/co- authored)		method show that OSP, RR, and their flake product have significant amount of $\beta$ and $\alpha$ cards $\beta$ -cryptoxanthine, and also $\alpha$ and $\gamma$ -tocopherol. Different response was observed on the bioactive compound and antioxidant activity affected yg heating process. Meanwhile, OSP a RR can be combined to formed a promising flake breakfast cereal products as shown from t physicochemical analysis such as moisture and dietary fiber contents, water absorption independent of the statement				
English Editing (/user/pre_english_article/st		fracturability, crispness, and color. Those quality parameters were affected by the proportions OSP and RR in the flake products. Moreover, the preference scores (n=120 panelists) for the flakes ranged from slightly liked to indifferent. It can be concluded that OSP and RR are pote				
Discount Vouchers		sources of bioactive compounds which could act as antioxidant and could be developed into				
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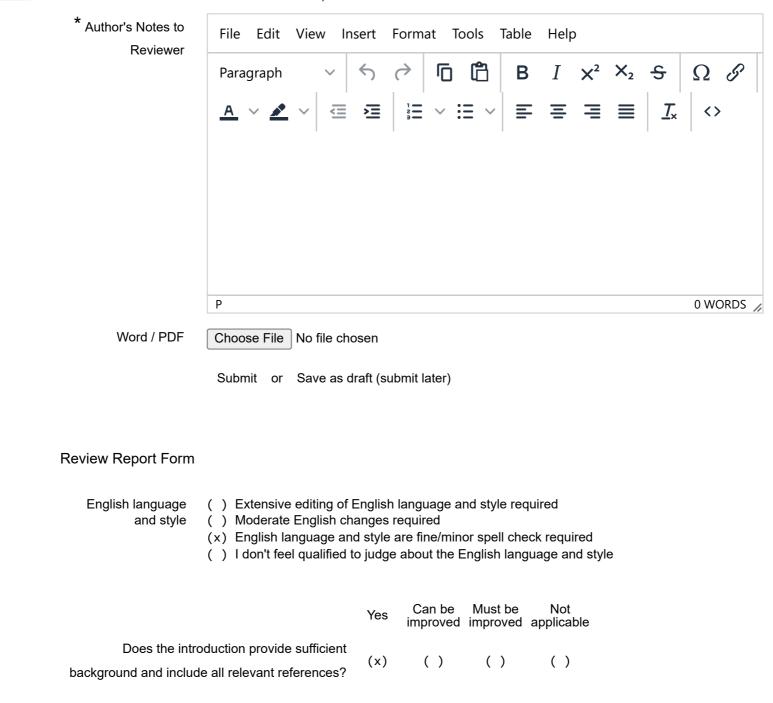
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Is the research design appropriate?	(x)	( )	()	()
Are the methods adequately described?	(x)	()	()	()
Are the results clearly presented?	(x)	()	()	()
Are the conclusions supported by the results?	(x)	()	()	( )

# Comments and Dear Editor,

# Suggestions for

Authors I carefully read the submission title 'Effect of processing on bioactive compounds, antioxidant activity, physicochemical, and sensory properties of orange sweet potato, red rice, and their flakes product'.

In fact more recently there has been an increasing interest to local product.

My first impression that the paper contain new information and title of the manuscript cover its content. The summary is appropriate and the aim of the work clearly established. The methods are used are adequate and used sophisticated techniques and equipment's. I found the results very reliable. Discussion and conclusions are well documented and scientifically coherent.

The language of paper is fluently.

However, I have some corrections and additions on it before acceptance.

ABSTRACT: Should be include more numeric data. I did not see numeric data on abstract.

INTRODUCTION: Very well and logically prepared

M&M. It is OK

Discussion: Line 308. Phenolic compounds are the most widely found bioactive compounds in plants..needs references. I suggested below ones

Zia-UI-Haq, M.; Ahmad, S.; Qayum, M.; Ercisli, S. Compositional studies and antioxidant potential of *Albizia lebbeck* (L.) Benth. Pods and seeds. *Turk. J. Biol.* **2013**, 37, 25-32.

Mollova, S.; Fidan, H.; Antonova, D.; Bozhilov, D.; Stanev, S.; Kostova, I.; Stoyanova, A. Chemical composition and antimicrobial and antioxidant activity of *Helichrysum italicum* (Roth) G. Don subspecies essential oils. *Turk. J. Agric. For*, **2020**, 44, 371-378.

Karatas, N.; Sengul, M. Some important physicochemical and bioactive characteristics of the main apricot cultivars from Turkey. Turk. J. Agric. For. 2020, 44, 651-661.

The rest of parts are well prepared

Submission Date07 January 2022Date of this review13 Jan 2022 05:02:16

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# **Response to Reviewer 1 Comments**

**Point 1:** ABSTRACT: Should be include more numeric data. I did not see numeric data on abstract.

**Response 1**: Thank you for the reviewer suggestion. Changes has been made in abstract by include more numeric data (line 18-19; 22-23)

**Point 2:** Discussion: Line 308. Phenolic compounds are the most widely found bioactive compounds in plants..needs references. I suggested below ones

Response 2: Thank you for the reviewer suggestion. New reference by Karatas and Sengul (Turk. J. Agric. For. 2020, 44, 651-661) as suggested has been added in line 321.



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Edit Profile (/user/edit)	Authors	Ignasius Radix A.P. Jati * , Laurensia M.Y.D Darmoatmodjo , Thomas Indarto Putut Suseno ,			
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Display Submitted Manuscripts (/user/manuscripts/status)	atus)	explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their developed flake product, and also investigated their antioxidant activity, physicochemical properties, and sensory properties. Simultaneous identification using liquid chromatographic			
Display Co-Authored Manuscripts (/user/manuscripts/co- authored)		method show that OSP, RR, and their flake product have significant amount of $\beta$ and $\alpha$ carof $\beta$ -cryptoxanthine, and also $\alpha$ and $\gamma$ -tocopherol. Different response was observed on the bioactive compound and antioxidant activity affected yg heating process. Meanwhile, OSP a RR can be combined to formed a promising flake breakfast cereal products as shown from t physicochemical analysis such as moisture and dietary fiber contents, water absorption independent of the statement of the statemen			
English Editing (/user/pre_english_article/sta		fracturability, crispness, and color. Those quality parameters were affected by the proportions of OSP and RR in the flake products. Moreover, the preference scores (n=120 panelists) for the flakes ranged from slightly liked to indifferent. It can be concluded that OSP and RR are potential			
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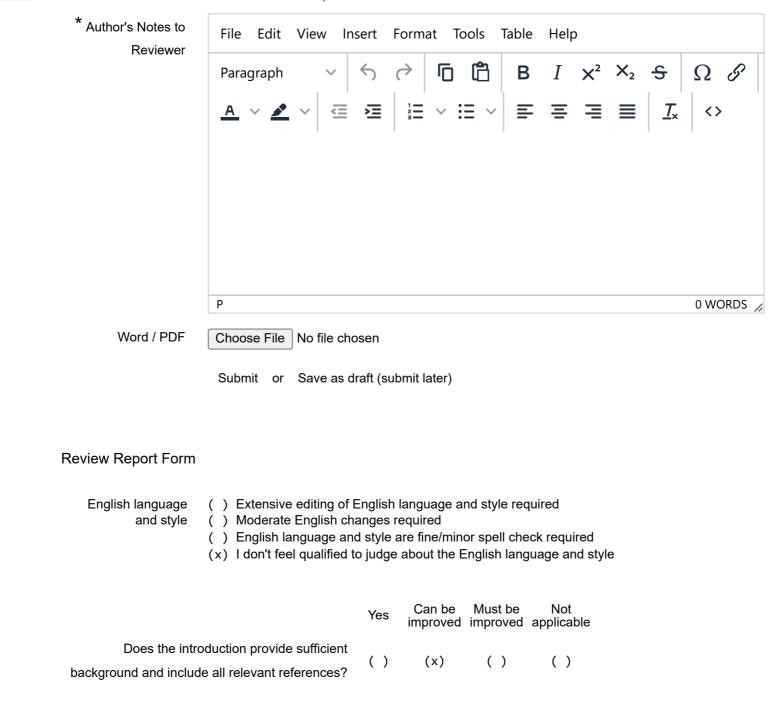
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Is the research design appropriate?	(x)	()	()	()
Are the methods adequately described?	(x)	()	()	()
Are the results clearly presented?	(x)	()	()	()
Are the conclusions supported by the results?	()	(x)	()	()

Comments and Suggestions for Authors

The development and use of innovative functional food products rich in bioactive compounds are clearly an objective for both the agri-food industry and consumers health.

This work addresses the question of new industrial products that are rich in biocactive molecules and acceptable by consumers. The authors describe the development of a range of breakfast flakes produced with red rice and orange sweet potatoes in different proportions. These new products have been largely characterized in terms of their content in phenolic compounds, carotenoids, tocopherols and dietary fibers, analyzed for their antioxidant activity and some physico-chemical properties, *i.e.* water absorption, fracturability, color profile, as well as a sensory analysis.

The methodological approach seems appropriate and the experiments well conducted. The results are correctly presented and appropriately discussed. Briefly, this study evidences that the industrial processing of natural ingredients affects the properties of the final product: *e.g.* in this case, heat treatment and general processing result in a decrease of phenolic compounds and in antioxidant activity, but also decrease or increase the content of carotenoids and tocopherols.

If we consider the primary scope of this study, *i.e.* the development of breakfast flakes form orange sweet potato and red rice, this should appear as very limited and restricted to only few readers specialized in this range of products. Nevertheless, I consider that the methodological approach that has been developed and is described in this paper is really very important and that this paper should therefore include this extended scope. Many studies consider natural products and promote them as functional food only based on the characteristics of the native ingredients although the consumers will generally use them after (ultra)transformation(s), which, as shown by this study considerably modify, negatively or positively, their properties.

I therefore suggest to the journal to ask the authors to modify the abstract, introduction and the end of their paper by emphasizing the conceptual basis of their approach and suggest to extend it to other products. Incidentally, they should also briefly indicate that the bioactive compounds may be further affected by the *in vivo* digestion process as well as by the intestinal absorption, to reach, at the end, the bioavailability of the product and its potential health effects.

Submission Date 07 January 2022

Date of this review 26 Jan 2022 12:11:32

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# **Response to reviewer 2.**

**Point 1:** I therefore suggest to the journal to ask the authors to modify the abstract, introduction and the end of their paper by emphasizing the conceptual basis of their approach and suggest to extend it to other products.

Response 1: Thank you for the reviewer's suggestion. Changes has been made to emphasize the conceptual basis of our approach including the addition of some references in:

**Abstract (line 14-16)** "This research approach was to closely monitor the changes of bioactive compounds and their ability as antioxidants from the native form to the food products which are ready to be consumed"

**Introduction (line 75-84)** "Numerous research has been published to explore the potency of various plant sources as functional foods. However, the approach mainly investigates the plant materials in the native or raw form. In contrast, the consumer will generally consume after the materials undergo transformations which could affect the characteristics of the products [23]. Moreover, besides the processing, the bioactive compounds and antioxidant activity will be further affected by the in vivo digestion and the intestinal absorption rate of the body metabolism before providing bioavailable compounds that can be used [24]. This research approach is to closely monitor the changes of the bioactive compound and antioxidant activity from the raw materials to the ready to be consumed food products"

**Discussion (line 504-513)** "This research has successfully monitored the changes of the bioactive compound and antioxidant activity of OSP and RR in their native form and the flake product. It can be observed that individual compounds were acted differently to processing methods. Therefore, it can be suggested that research on the development of functional foods should address the products ready to be consumed instead of solely focusing on the raw materials due to the changes that have taken place during the transformation. This approach should be implemented for other potential materials rich in bioactive compounds. Moreover, further consideration of the bioaccessibility and bioavailability that are affected by digestion and absorption in the human metabolism system should also be considered [60]"



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Display Submitted Manuscripts (/user/manuscripts/status)		explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their developed flake product, and also investigated their antioxidant activity, physicochemical properties, and sensory properties. Simultaneous identification using liquid chromatographic			
Display Co-Authored Manuscripts (/user/manuscripts/co- authored)		method show that OSP, RR, and their flake product have significant amount of $\beta$ and $\alpha$ carotene $\beta$ -cryptoxanthine, and also $\alpha$ and $\gamma$ -tocopherol. Different response was observed on the bioactive compound and antioxidant activity affected yg heating process. Meanwhile, OSP and RR can be combined to formed a promising flake breakfast cereal products as shown from the physicochemical analysis such as moisture and dietary fiber contents, water absorption index,			
English Editing (/user/pre_english_article/stat	us)	fracturability, crispness, and color. Those quality parameters were affected by the proportions of OSP and RR in the flake products. Moreover, the preference scores (n=120 panelists) for the flakes ranged from slightly liked to indifferent. It can be concluded that OSP and RR are potential			
Discount Vouchers (/user/discount_voucher)		sources of bioactive compounds which could act as antioxidant and could be developed into flake product that meet the dietary and sensory needs of consumers.			
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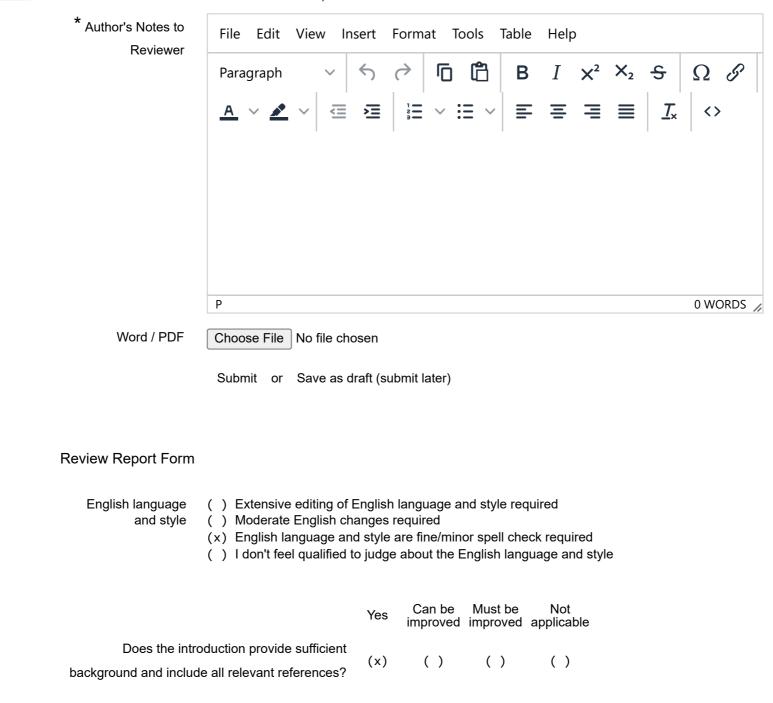
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Are the methods adequately described?		(x)	( )	( )	( )
Are the results clearly presented?		()	(x)	( )	( )
Are the conclusions supported by the results?		(x)	( )	( )	( )
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Antioxidant-enriched dietary products play crucial role in health promotion. Unfortunately not all of the promising plant products have acceptable taste; however, besides other organoleptic parameters, it is one of the most critical point for the consumers. The idea to incorporate orange sweet potato and red rice into flakes could be very beneficial regarding to their antioxidant properties. According to this research, most of the flake products also meet the requirements on the sensory analysis. Even so some questions arose while reading the manuscript.

Affiliation has to be checked.

Line 19-20: "Different response was observed on the bioactive compound and antioxidant activity affected yg heating process." There is a mistype in this sentence.

Line 20-21: "Meanwhile, OSP and RR can be combined to formed a promising flake breakfast cereal products as shown from the physicochemical analysis such as moisture and dietary fiber contents, water absorption index, fracturability, crispness, and color." Grammatically incorrect sentence.

Since they have the same meaning only "flakes" and not "breakfast cereal" should be highlighted as keyword.

After the first appearance abbreviations should be used for red rice and orange sweet potatoes in the main text. In addition abbreviations need to be defined the first time used in figure/ table caption.

Line 39-40: "Secondary metabolites found in have been shown to decrease the risks of degenerative diseases such as coronary heart disease, diabetes, cancer, and stroke [1-4]." Incomplete sentence.

Line 41-44: Grammatically incorrect sentences.

It is highly advised to use the plural form of anthocyanin in the Introduction.

Line: 53-54: "Despite the touted health benefits of red rice, its consumption remains low." Touted has different meaning, it has to be replaced.

Aquadest has to be replaced by distilled water in Materials and Methods.

Line 98-99 and 103: Are 30 mm and 80 mm mesh units correct?

Line 122-123: "The results were calculated as g catechin equivalents/100 g dry weight." In contrast the phenolic content of the samples are given in mg GAE/100g DW in Table 1, explanation is needed in the text.

Line 127-131: This part has to be edited and unified.

Line 143-144: The word "reconstituted" is not appropriate here.

How many repetitions were made for HPLC analysis?

Line 165: Edit the formula of  $FeSO_4 \cdot 7H_2O$ .

Line 188-189: "The results were obtained in the form of a graph (force vs. time)." Indicate that this data are not shown in the manuscript.

Line 209: The meaning of OSPF and RRF were not explained in the manuscript.

Table 1: The interpretation of the data is not clear enough. The data should appear on the same page and the corresponding bioactive molecules and their values in the same rows. Use nd for all of the biomolecules which cannot be detected.

The values of the different bioactive molecules are presented in Table 1; therefore their appearance in the text is redundant.

Line 238-239: It was not specified that RR has higher  $\alpha$ -tocopherol content, only "raw" was mentioned in the text.

Line 239-241: "Unlike the decreasing trend observed in other bioactive compounds due to processing, the tocopherol content of both RR and OSP increased after conventional cooking." The  $\alpha$ -carotene content of the samples was also elevated after cooking which was not described in the text.

Figure 1: Size and resolution of Figure 1 has to be increased. Check Figure legend.

Line 268-269 "The dietary fiber content of flake products ranged from  $13.86 \pm 0.73\%$  to  $9.47 \pm 0.01\%$ ." The order of numbers should be changed.

The values of color profiles are presented in Table 3; therefore their appearance in the text is redundant.

Table 3: The description of the abbreviations is missing.

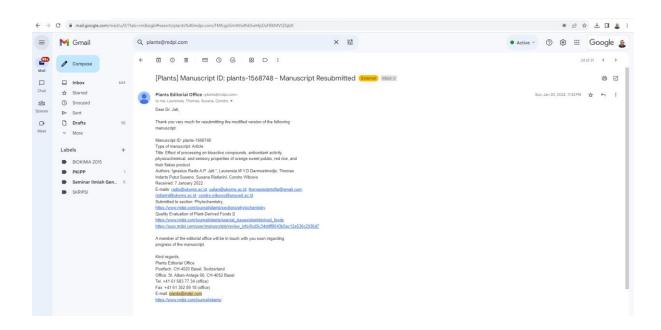
Line 356-357: "The  $\beta$ -carotene was significantly lower the cooked sample compared to the raw sample." Grammatically incorrect sentence.

Line 396-398: "Some compounds such as procatechuic acid, p-coumaric acid, and ferulic acid have high thermal stability, which facilitates their extraction and antioxidant activity." Reference should be added.

Line 432-433 and 435-436: These sentences have the same meaning; one of them has to be cancelled.

Nevertheless the manuscript provides a novel way to improve the health impact of flake products, by increasing the amount of bioactive compounds accompanied with promising antioxidant activity. I suggest submitting the manuscript to a minor revision.

# 4. Bukti konfirmasi submit revisi artikel dan artikel yang di-resubmit 30 Januari 2022







## Effect of processing on bioactive compounds, antioxidant activity, physicochemical, and sensory properties of orange sweet potato, red rice, and their application for flakes product

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Abstract: Orange sweet potato (OSP) and red rice (RR) are rich source of health benefits associated 13 substances and can be conventionally cooked or developed into food product. This research ap-14 proach was to closely monitor the changes of bioactive compounds and their ability as antioxidants 15 from the native form to the food products which are ready to be consumed. Moreover, this research 16 explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their de-17 veloped flake product, and also investigated their antioxidant activity, physicochemical properties, 18 and sensory properties. Simultaneous identification using liquid chromatographic method show 19 that OSP, RR, and their flake product have significant amount ( $\mu$ g/g) of  $\beta$  carotene (278.58-48.83),  $\alpha$ 20 carotene (19.57-15.66),  $\beta$ -cryptoxanthine (4.83-2.97),  $\alpha$  -tocopherol (57.65-18.31), and also  $\gamma$  -tocoph-21 erol (40.11-12.15). Different response was observed on the bioactive compound and antioxidant ac-22 tivity affected by heating process. Meanwhile, OSP and RR can be combined to form a promising 23 flake products as shown from the physicochemical analysis such as moisture (5.71-4.25%) and die-24 tary fiber (13.86-9.47%) contents, water absorption index (1.69-1.06), fracturability (8.48-2.27), crisp-25 ness (3.9-1.5), and color. Those quality parameters were affected by the proportions of OSP and RR 26 in the flake products. Moreover, the preference scores (n=120 panelists) for the flakes ranged from 27 slightly liked to indifferent. It can be concluded that OSP and RR are potential sources of bioactive 28 compounds which could act as antioxidant and could be developed into flake product that meet the 29 dietary and sensory needs of consumers. 30

Keywords: orange sweet potato; red rice; flakes; bioactive compound; antioxidant activity; physi-31cochemical; sensory properties32

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## 1. Introduction

Modern food trends and lifestyle changes have strongly influenced the dietary habits 35 of society. The demands for ready-to-eat and simple-to-prepare foods are increasing rap-36 idly, providing an excellent opportunity for food industries to play a significant role in 37 supplying such food products. Flakes, one of the most popular foods made from cereals, 38 typically oat, corn, and barley, are commonly served for breakfast with milk in Europe 39 and the USA, and their global appeal is gaining traction. Besides, healthy eating has be-40 come a new trend in modern culture, with consumers increasingly opting for healthy 41 foods options. Secondary metabolites found in plants have been shown to decrease the 42 risks of degenerative diseases such as coronary heart disease, diabetes, cancer, and stroke 43

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[1-4]. Thus, innovative functional food products rich in bioactive compounds are needed
to promote a healthy diet and reduce disease risks. Asia has the most rapidly growing
food product market, so using local ingredients can help open a new market for flakes
and decrease the reliance on imported foods such as oat and barley. Commodities that can
potentially be developed into flakes are red rice (RR) and orange sweet potato (OSP).

RR (Oryza nivara L.) is a variety of rice with red pericarp caused by anthocyanins in 49 the aleurone layer. It is a rich source of anthocyanins such as cyanidin 3-O-glucoside and 50 peonidin 3-O-glucoside [2]. Anthocyanins has also been reported to inhibit plaque for-51 mation [3] and to exhibit hypocholesterolemic [4] and anticancer effects [5] in RR. The 52 health benefits of RR have also been linked to bioactive compounds such as tocopherol 53 and tocotrienols [6,7], and dietary fiber [8]. RR also contains higher essential minerals, 54 including iron, zinc, and vitamins, which are especially important for babies and toddlers, 55 than white rice [9]. Despite the various health benefits of RR, its consumption remains 56 low. It is generally considered inferior to white rice due to the hard texture and unpleasant 57 aroma when cooked. Traditionally, RR is consumed steamed or boiled, and its sensory 58 properties are inferior to white rice. The most popular RR-based product is RR baby por-59 ridge, but there are reports on the development of RR-based products such as pasta [10], 60 noodles [11], flakes [12], rice milk [13], and fermented beverages [14]. However, these have 61 not been scaled up or commercially established. 62

OSP (Ipomoea batatas) is one variety of sweet potato with a bright orange flesh color 63 caused by carotenoids, of which high amounts of  $\beta$  carotene,  $\alpha$  carotene, and  $\beta$  cryptoxan-64 thin have been reported [15–17]. In many countries, OSP has been used to eradicate vita-65 min A deficiency due to its high content of beta carotene, a pro-vitamin A carotenoid. 66 Sweet potato was extensively promoted in Africa and some Asian countries with remark-67 able results [18–20]. However, there were difficulties in ensuring its sustainability due to 68 the monotonous way of its preparation, mainly boiled and baked, even though the essen-69 tial nutritional components were reportedly retained after processing [21]. Processing 70 OSP could decrease the beta carotene, but not below the recommended dietary level [22]. 71 Moreover, food prepared from OSP by the traditional method was not attractive to chil-72 dren, who were the main target of the vitamin A intake enhancement. Therefore, innova-73 tive food products need to be developed to promote the consumption of RR and OSP. 74 Numerous research has been published to explore the potency of various plant sources as 75 functional foods. However, the approach mainly investigates the plant materials in the 76 native or raw form. In contrast, the consumer will generally consume after the materials 77 undergo transformations which could affect the characteristics of the products [23]. More-78 over, besides the processing, the bioactive compounds and antioxidant activity will be 79 further affected by the in vivo digestion and the intestinal absorption rate of the body 80 metabolism before providing bioavailable compounds that can be used [24]. This research 81 approach is to closely monitor the changes of the bioactive compound and antioxidant 82 activity from the raw materials to the ready to be consumed food products and aimed to 83 investigate the bioactive compounds and antioxidant activity of raw and cooked RR and 84 OSP and the physicochemical and sensory properties of their developed flake products. 85

## 2. Materials and Methods

## 2.1. Plant materials and Chemicals

A local variety of RR (*Oryza nivara* L.), "Cempo abang," and OSP (*Ipomoea batatas* L.), 88 "Mendut," were collected from farmers in Yogyakarta province, Indonesia. Chemicals 89 used for analysis, including distilled water, Folin Ciocalteu, 1,1-diphenyl-2-picrylhydra-2yl (DPPH), gallic acid, butylated hydroxyl toluene (BHT), enzymes (thermamyl, pancreatin, pepsin), riboflavin, methionine, and nitroblue tetrazolium (NBT), were purchased 92 from Sigma Chemical. Methanol, Whatman 40 filter paper, n-hexane, NaOH, HCl, ethanol, and phosphate buffer (pH 6) were purchased from Merck, Germany. Carotenoid 94

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#### The RR samples were washed, drained, and blended (Philips food processor). The 99 OSP samples were chopped into small pieces. All samples were then freeze-dried, refined, 100 and sieved (30 mesh). Finally, the powdered samples were placed in dark bottles and 101 stored in a refrigerator (4°C) for further usage. 102

standards, including  $\beta$  and  $\alpha$ -carotene,  $\beta$ -cryptoxanthin, lycopene, and lutein, and  $\alpha$  and

 $\gamma$ -tocopherol standards were obtained from Sigma Aldrich.

## 2.2.2. Boiled samples

2.2. Sample preparation

2.2.1. Raw samples

RR (75 g) was cooked with 135 g of tap water (1:1.8; w/w) in a rice cooker (Panasonic) 104 for approximately 45 min. After, the cooked rice was cooled for 10 min. Meanwhile, 150 g 105 of OSP was boiled in an aluminum pot using tap water for 20 min, cooled for 10 min, and 106 then mashed. Both samples were freeze-dried, refined, sieved (30 mesh), and stored  $(4^{\circ}C)$ 107 in a refrigerator. 108

## 2.2.3. Flakes production

RR was placed in a cabinet dryer (60°C) for 1 h. The OSP was peeled, sliced and 110 placed in a cabinet dryer (60°C) for 6 h. The dried OSP and RR were mashed using a 111 blender. The flour was passed through an 80 mesh. Flakes were produced using six dif-112 ferent proportions of OSP and RR, namely 100:0, 80:20, 60:40, 40:60, 20:80, and 0:100. Salt 113 (3% w/w), sugar (30% w/w), and water (150% w/w) were mixed in as additional ingredi-114 ents. The mixture was heated at 75°C for 1 min and pressed at 170°C for 1 min using a 115 customized flake pressing tool. The pressed flakes were cut at 2x2 cm and dried using an 116 oven at 125°C for 5 min. 117

## 2.3. Bioactive compounds and antioxidant activity analysis

## 2.3.1. Methanolic extract of samples

The extraction of samples (raw, boiled, and flakes) was done according to a previ-120 ously published procedure [25]. Briefly, 1 g of sample was weighed, ground, placed in a 121 centrifuge tube, and then extracted with 10 mL of 1% methanol-HCl solution. The mixture 122 was then vortexed for 15 min, centrifuged at 5000 rpm for 15 min, filtered (Whatman No. 123 40), and then used for antioxidant activity analysis, performed in triplicate. 124

## 2.3.2. Phenolic content

The total phenolic content was determined using the Folin Ciocalteau method by Singleton and Rossi as described in other published work [26]. In brief, 0.1 ml of extract was 127 mixed with 0.5 ml 1:1 Folin Ciocalteu reagent and distilled water. After 10 min, 4.5 ml of 128 2% sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>) was added, and the mixture was then vortexed and kept 129 in the dark for 1 h. The blue complex formed was measured using a spectrophotometer at 130 765 nm. Methanol and gallic acid were used as the blank and standard, respectively. The 131 results were calculated as mg gallic acid equivalents (GAE)/100 g dry weight. 132

## 2.3.3. Anthocyanin Content

The total anthocyanin in RR and flake samples was determined spectrophotometri-134 cally using the pH differential method [27]. In brief, 1 mL of extract was diluted in pH 1.0 135 and pH 4.5 buffers. The absorbance was measured at 510 and 710 nm. The final absorbance 136 was calculated using the formula: 137

A = [(A513-A700)pH 1.0 - (A513-A700)pH 4.5)] (1)	1) 138
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The calculated absorbance was then used to calculate the total g of anthocyanins per 139 100 g dry weight, with a molar extinction coefficient of 26,900 and a molecular weight of 140 445.

## 2.3.4. Carotenoid and tocopherol analysis

The carotenoids and tocopherols in the raw and boiled samples of RR and OSP were 143 determined simultaneously using High-Performance Liquid Chromatography (HPLC) 144 based on previously published research [24]. In brief, 0.3 g of finely ground samples were 145 extracted with 0.5 mL of 70% ethanol and 0.4mL of n-hexane under yellow lights. In ad-146 dition, 0.4 mL  $\beta$ -Apo-8' carotenal-O-methyloxim and 0.4 mL  $\alpha$ ,  $\gamma$  to copherol were used as 147 an internal standard for carotenoids and tocopherols, respectively. The mixture was 148shaken for 20 min, centrifuged at 5000 rpm, 4°C for 20min. Next, the upper layer of extract 149 containing a hexane fraction was collected using a micropipette. The extraction was re-150 peated four times using only n-hexane as a solvent. Finally, the hexane fractions were 151 pooled and completely dried using nitrogen gas. 152

Before injection, the extracts were mixed with 200µl of ethanol containing 30µg/ml 153 BHT, then 20µl of the mixture were injected into the HPLC (Varian Pro Star 410, Spark, 154 Holland). A mixture of 82% acetonitrile, 15% dioxan, 3% methanol, 0,1M ammonium ac-155 etate, and 0.1% triethylamine was assigned as the mobile phase and was pumped at a rate 156 of 1.6 ml/min. The solvent was pre-mixed to avoid dependency on reproducible mixing 157 by the pump. For separation, a C18 Spherisorb ODS 2 column (3  $\mu$ m, 250 × 4.6 mm) was 158 applied. In addition, a UV Vis detector at 450nm and a Scanning Fluorescence detector 159 using an excitation wavelength of 295nm and an emission wavelength of 328nm were 160 used to monitor the carotenoids and tocopherols, respectively. Five repetitions were per-161 formed for the HPLC analysis. 162

## 2.3.5. DPPH radical scavenging activity

The radical scavenging activity of the extract was examined by the DPPH method 164 [25]. In brief, 1 mL of extract was mixed with 2 mL of 0,2 M DPPH and 2 mL of methanol 165 in centrifuge tubes, vortexed, and kept in the dark for 1 h. The absorbance was measured 166 spectrophotometrically at 517 nm. As a control, 150 ppm BHT solution was used. The 167 DPPH radical scavenging activity of the extract was expressed as a percentage calculated 168 as follows: % radical scavenging capacity = ((Absorbance of control – Absorbance of sample)/Absorbance of control) \* 100% 170

## 2.3.6. FRAP Assay

The ferric reducing antioxidant power (FRAP) was examined based on a previously 172 published report [28]. In brief, a mixture of 60  $\mu$ l extract, 180  $\mu$ l distilled water, and 1.8 ml 173 FRAP reagent was vortexed and incubated at 37°C for 30 min. The spectrophotometer was 174 used to read the absorbance of the mixture at 593 nm. A standard curve was prepared 175 with Fe [II] (FeSO<sub>4</sub>.7H<sub>2</sub>O, 100–2000 mM) to calculate the reducing power. The result was 176 expressed as mmol Fe[II]/g. In addition, methanol was used for the reagent blank. 177

## 2.3.7. Superoxide Radical Scavenging Capacity

A previous report [29] was followed to examine the superoxide radical scavenging capacity. Firstly, a reagent containing riboflavin, methionine, and NBT in 0.05 M phosphate buffer pH 7.8 was prepared. Then, 100  $\mu$ l of the extract was mixed with 4.9 ml of reagent and illuminated (20 W fluorescent lamp) at 25°C for 25 min. The absorbance was measured at 560 nm. 183

2.4. Physicochemical properties of flake products	184
2.4.1. Moisture and dietary fiber contents	185

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## 2.4.2. Water absorption index

in triplicate.

The water absorption index was examined according to previously published 190 method [31]. In brief, 5 g of flakes were placed in a 100 mL beaker, 30 mL of water at 30°C 191 was added. After 10 s of immersion, the flakes were dried. The water absorption index 192 was calculated using the formula: WAI = (wf -wi)/wi, where wi and wf are the initial and 193 final weight of the sample, respectively. 194

The moisture content was measured thermogravimetrically [30]. In brief, 1 g of each

sample was dried at  $105 \pm 0.2$ °C to establish a constant mass. The analysis was performed

## 2.4.3. Fracturability and crispness

The fracturability and crispness of flakes were measured using TA-XT Plus Texture 196 Analyzer (Stable Micro Systems, UK) [32]. The probe used was a ¼ inch spherical stain-197 less-steel probe (P0.25S). The sample was placed on the sample holder, and then the probe 198 was moved down to press the sample. The results were obtained in the form of a graph 199 (force vs. time) (the graph was not shown). The value of the y-axis at the graph's highest 200 point is the maximum force value that can be held by the sample, called the value of frac-201 turability. Crispness can be measured through changes in displacement distance during a 202 drastic decline in the graph pattern from the highest peak to the next peak point. 203

## 2.5. Sensory analysis

The sensory evaluation was conducted by 120 untrained panelists to determine the 205 level of consumer preference for the flakes with various proportions of OSP and RR. The 206 parameters tested were preferences for color, taste, crispness of flakes, and mouthfeel 207 when served with milk. The Hedonic Scale Scoring method (preference test) with a scale 208 ranging from 1 (strongly disliked) to 7 (strongly liked) was used for the sensory test. 209

Samples of flakes for the color preference test were prepared in open white plastic 210 containers. Panelists were asked first to assess aspects of flakes' taste, color, and crispness 211 before serving with milk. The crispness was evaluated based on the panelist's preference 212 for the sound of flakes during biting. For the mouthfeel test, 5 g of flakes were prepared 213 in a small plastic container. Panelists were instructed to pour 10 mL of milk into the con-214 tainer and wait for 1 min. Then, they were asked to assess the mouthfeel of the flakes 215 based on preference level by filling the questionnaire sheet provided. 216

## 2.6. Statistical analysis

The data were statistically analyzed using Anova ( $\alpha = 5\%$ ) followed by Duncan's 218 Multiple Range Test (DMRT) on SPSS software version 19. Spider web chart analysis using Microsoft Excel was used to determine the best proportion of OSP and RR in flakes 220 based on the panelists' preferences. 221

## 3. Results

## 3.1. Bioactive compounds of OSP, RR, and their flake products

Table 1 shows the phenolic compound, anthocyanin, carotenoid, and tocopherol con-224tents of raw and cooked OSP and RR and their flakes containing different proportions of225OSP and RR. The raw RR had a higher phenolic content than OSP. As a result, the higher226the proportion of RR, the higher the phenolic content of the developed flakes. It was also227found that cooking decreased the phenolic content of RR and OSP by approximately22849.34% and 41.08%, respectively.229

Interestingly, the flakes with 100% RR showed a higher phenolic content than the230cooked RR. Anthocyanin was only observed in RR. Furthermore, flakes containing 100%231RR had a lower anthocyanin content than the conventionally cooked RR.232

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	OSP		R	R		Proportions of OSP and RR in flakes				
	Raw	Cook	Raw	Cook	100:0	80:20	60:40	40:60	20:80	0:100
Phenolic (mg	110.68	65.21±	301.89±	152.91±	77.46±	97.34±	102.03±	131.79 ±	146.09±	162.40±
GAE/100 g DW)	±18.3ª	7.3 <sup>b</sup>	24.86 ª	28.92 <sup>b</sup>	8,28 ª	8.79 <sup>b</sup>	11.65 °	10.93 <sup>d</sup>	15.64 °	21.54 <sup>f</sup>
Anthocyanin	1	1	8,81±	8.64±	,		1.74±	2.09±	3.73±	5.81±
(mg/100g DW)	nd	nd	0.05 ª	0.08 a	nd	nd	0.06 c	0.05 <sup>d</sup>	0.07 <sup>e</sup>	$0.04  {}^{\mathrm{f}}$
	278.58	134.17±	13.17 ±	7.37 ±	48,83±	36.27±	25.77±	27.23±	15.69±	3.12±
β-carotene (µg/g)	± 31.5 ª	17.2 <sup>b</sup>	2.62 ª	0.5 <sup>b</sup>	3,31 ª	3.01 <sup>b</sup>	3.45 °	2.72 <sup>d</sup>	2.21 <sup>e</sup>	0.66 <sup>f</sup>
	19.57 ±	23.83 ±	5.53 ±	11.66 ±	15.61±	11.82±	5.31±	2.53±	nd	nd
$\alpha$ -carotene (µg/g)	1.8 ª	1.6 <sup>b</sup>	1.4 ª	1.5 <sup>b</sup>	1.44 a	3.11 <sup>b</sup>	1.59 °	0.87 <sup>d</sup>		nd
β-cryptoxanthine	4.83±	$4.48\pm$	3.67 ±	3.96 ±	2.64±	2.81±	2.97±	2.78±	2.77±	2.81±
(µg/g)	0.2 ª	0.8 ª	2.15 ª	1.9 <sup>a</sup>	0,05 a	0.13 <sup>b</sup>	0.08	0.12 c	0.25 °	0.16 <sup>b,c</sup>
	3.77±	3.81 ±	2.16 ±	1.82 ±	. 1	. 1	. 1	. 1	. 1	. 1
Lutein (µg/g)	0.8 ª	0.7 <sup>a</sup>	0.8 ª	0.5 <sup>b</sup>	na	nd nd	nd	nd	nd	nd
	13,23±	15.11 ±	$34.08 \pm$	57.65 ±	4.58±	7.34±	10.51±	12.45±	16.82±	18.31±
$\alpha$ -tocopherol (µg/g)	1.1 ª	0,5 <sup>b</sup>	2.2ª	2.1 <sup>b</sup>	0.73 <sup>a</sup>	1.49 <sup>b</sup>	1.27 °	1.21 <sup>c</sup>	0.52 °	$0.77^{\rm f}$
	2.40±	5.38±	29.27 ±	40,11 ±	. 1	. 1	3.38±	6.71±	8.06 ±	12.15±
γ-tocopherol (µg/g)	0,2 ª	0.05 <sup>b</sup>	2.4 ª	1.8 <sup>b</sup>	nd	nd	1.22 °	1.19 <sup>d</sup>	0.98 e	$0.73^{\text{ f}}$

Table 1. Bioactive Compounds of Orange Sweet Potato (OSP), Red Rice (RR) and the flake products

. Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–f) denote significantly different values 235 according to Duncan's test (p < 0.05). Comparison was made within each category (OSP, RR, and Flakes) 236

Of the carotenoids, OSP had a higher content of  $\beta$  carotene,  $\alpha$ -carotene,  $\beta$ -cryptoxanthin, and lutein.  $\beta$  carotene and  $\beta$ -cryptoxanthin were the dominant carotenoids observed in RR. The cooking process significantly decreased the content of  $\beta$  carotene in OSP and RR roughly by 48% and 56%, respectively. Overall, flakes containing higher amounts of OSP showed higher carotenoid contents. The processing significantly decreased the carotenoids in the flake products when considering the raw forms and the proportions of OSP and RR, and  $\beta$  carotene was the major carotenoid remaining in the flake products. 244

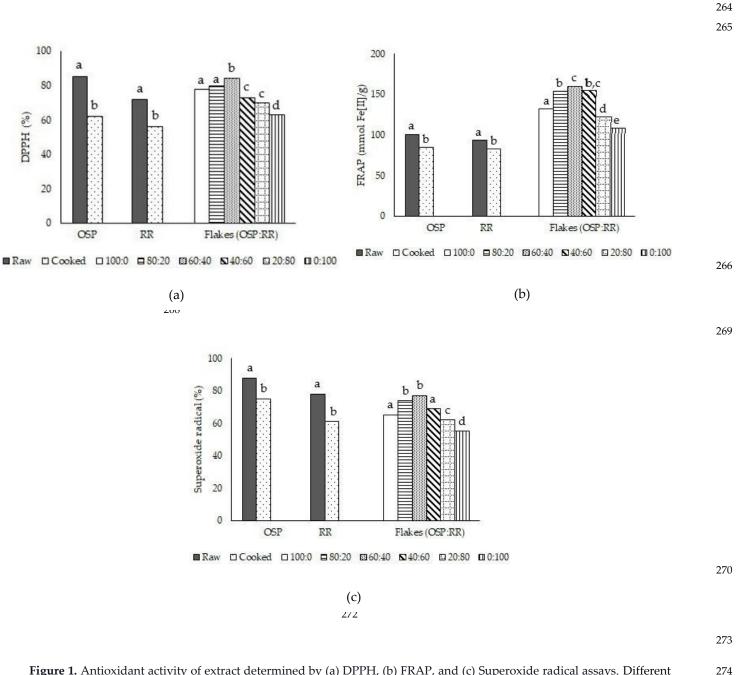
Moreover, raw RR contained higher  $\alpha$ -tocopherol than OSP in raw forms. Unlike the 245 decreasing trend observed in other bioactive compounds due to processing, the tocopherol and  $\alpha$ -carotene contents of both RR and OSP increased after conventional cooking. On 247 the other hand, the two-step thermal processing decreased the tocopherol content of 248 flakes. 249

### 3.2. Antioxidant activity of raw and cooked OSP and RR and their flake products

The antioxidant activity of raw and cooked OSP and RR and their flake products 251 were examined using DPPH, FRAP, and Superoxide radical scavenging activity methods. 252 Figure 1a shows the DPPH scavenging activity of methanolic extract of RR, OSP, and the 253 flake products. Boiling affected the ability of the methanolic extract to scavenge DPPH 254radicals. Approximately 16% and 23% decreases were observed in cooked RR and OSP, 255 respectively. The combination of OSP and RR in the ratio of 60:40 resulted in flakes with 256 the highest antioxidant activity (84%). The results trend indicated that the higher propor-257 tion of OSP contributed to the more robust antioxidant capacity of the extract. Further-258 more, the DPPH result was in agreement with FRAP (Figure 1b) and Superoxide scaveng-259 ing capacity (Figure 1c). Thus, conventional cooking and flake processing methods reduce 260 the antioxidant activity of extracts of OSP and RR compared to their raw forms, and the 261

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right combination of OSP and RR in the flake formulation is critical for higher antioxidant 262 activity. 263

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Figure 1. Antioxidant activity of extract determined by (a) DPPH, (b) FRAP, and (c) Superoxide radical assays. Different274superscript letters (a–f) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made275within each category (OSP: orange sweet potato, RR: red rice, and Flakes).276

## 3.3. *Physicochemical properties of OSP and RR-based flake products*

The proportion of OSP and RR in the flake formulation affected the moisture content278of flakes. A lower OSP proportion, resulted in a lower moisture content of flakes (Table2792). The dietary fiber content increased with a higher proportion of RR. The dietary fiber280content of flake products ranged from  $9.47 \pm 0.01\%$  to  $13.86 \pm 0.73\%$ .281

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			Proportions of C	SP and RR in fla	kes	
	100:0	80:20	60:40	40:60	20:80	0:100
Moisture content (%)	5.71 ± 0,07 <sup>a</sup>	$5.31 \pm 0.10^{\mathrm{b}}$	5.09 ± 0.06 °	4.87 ± 0.01 <sup>d</sup>	$4.43 \pm 0.03^{\mathrm{e}}$	$4.25 \pm 0.03$ f
Dietary fiber (%)	9.47 ± 0,01 <sup>a</sup>	$9.9 \pm 0.02$ <sup>b</sup>	$10.9\pm0.05^{\circ}$	$11.63 \pm 0.34$ <sup>d</sup>	$12,73 \pm 0.26$ $^{\rm e}$	$13.86 \pm 0.73$ f
Water absorption index	$1.69 \pm 0,03^{a}$	$1.14 \pm 0.02^{\mathrm{b}}$	$1.06\pm0.03^{\rm c}$	$0.96 \pm 0.03$ <sup>d</sup>	$1.09\pm0.03^{\rm \ b,c}$	$1.12 \pm 0.02^{b,c}$
Fracturability	$8.48 \pm 0,09$ a	$5.35 \pm 0.85$ b	$3.34 \pm 0.34$ °	$2.27\pm0.04^{\rm \ d}$	$3.17 \pm 0.09^{\mathrm{e}}$	$4.64 \pm 0.12$ f
Crispness	$3.9 \pm 0.03^{a}$	$2.4 \pm 0.02^{\mathrm{b}}$	$1.5 \pm 0.02$ °	$1.9 \pm 0.03^{d}$	$3.21 \pm 0.05^{\text{ e}}$	$3.7\pm0.03$ f

 Table 2. The physicochemical properties of flakes produced from different ratios of Orange Sweet Potato (OSP), Red

 Rice (RR).

Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–f) denote significantly different values 286 according to Duncan's test (p < 0.05). Comparison was made within each row 287

The water absorption index of flakes was lowest at an OSP to RR ratio of 40:60, with 289  $0.96 \pm 0.03$  %, and was generally higher at combination ratios of 100:0, 80:20, 0:100, and 290 20:80. In addition, the texture characteristic of flakes was determined by the fracturability 291 and crispness. The highest fracturability value was found in flakes made from 100% OSP 292  $(8.48 \pm 0.09)$ . The reduction of the OSP proportion in flakes caused a decrease in fractura-293 bility until the proportion of  $40:60 (2.27 \pm 0.04)$ , beyond which the fracturability of flakes 294 increased. A similar trend was observed in the crispness value of flakes. The lowest crisp-295 ness value was detected in flakes with an OSP to RR ratio of 60:40, while higher values 296 were obtained at ratios of 100:0, 0:100, 20:80, and 80:20. 297

Furthermore, the color of flakes was affected by the color of OSP and RR. The color profile of the flakes is shown in Table 3.

**Table 3.** Color profiles of flakes produced from different ratios of Orange Sweet Potato (OSP), Red301Rice (RR)302

	Proportions of OSP and RR in flakes					
_	100:0	80:20	60:40	40:60	20:80	0:100
L	44.0±0.1	47.3±0.2	51.5±0.1	52.7±0.4	51.8±0.2	51.8±0.7
a*	8.2±0.3	8,5±0.3	8.9±0.4	$9.4 \pm 0.4$	10.2±0.6	$10.8\pm0.4$
b*	16.5±0.3	14.7±0.2	13.4±0.2	10.3±0.2	$9.0\pm0.4$	$5.9 \pm 0.4$
٥h	63.574	59.9622	56.4087	47.6158	41.4237	28.6476
С	18.47	16.9685	16.0703	13.898	13.56	12.3145

## 3.4. Sensory characteristics of flakes

A preference test was conducted to determine the sensory characteristics of the 305 flakes. The results are presented in Table 4. The highest level of color preference was found 306 in flakes with OSP to RR ratios of 60:40, 40:60, and 20:80, while flakes containing 100% 307 OSP had the lowest level of acceptance for color. The preference scores for taste and crisp-308 ness of the flakes with various proportions of OSP and RR were comparable. Flakes con-309 taining 100% OSP and 100% RR received the highest preference score for mouthfeel, with 310 the former having a significant edge. Therefore, mixing the OSP and RR lowers the mouth-311 feel acceptance of the flake products. 312

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		Prop	ortions of OS	P and RR in	flakes	
	100:0	80:20	60:40	40:60	20:80	0:100
Color	3.35±1.42ª	$4.40 \pm 1.22^{b}$	5.14±1.12 <sup>c</sup>	4.93±1.21°	4.89±1.30°	$4.30 \pm 1.12^{b}$
Taste	$4.43 \pm 1.34^{a}$	$4.69 \pm 1.28^{ab}$	$5.08 \pm 1.18^{\circ}$	5.09±1.20°	$5.05 \pm 1.26^{bc}$	$4.72 \pm 1.29^{\text{abc}}$
Crispness	$4.76 \pm 1.16^{b}$	$4.85 \pm 1.04^{b}$	$4.76 \pm 1.40^{b}$	$4.08 \pm 1.17^{a}$	$4.96 \pm 1.07^{b}$	$5.00 \pm 1.04^{b}$
Mouthfeel	$5.41 \pm 0.91^{d}$	5.04±1.18°	$4.84 \pm 1.31^{bc}$	$3.91 \pm 1.59^{a}$	4.66±1.32 <sup>b</sup>	5.05±1.03°

**Table 4.** Preference test of flakes produced from different ratios of Orange Sweet Potato (OSP), Red315Rice (RR)316

Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–d) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each 318 row.</th>317319319

## 4. Discussion

Phenolic compounds are the most widely found bioactive compounds in plants [33]. 321 Some are produced in response to stress conditions as a defense mechanism of the plant. 322 They have been extensively investigated due to their antioxidant activity and anti-degen-323 erative disease effects [34]. In this research, methanol-HCl (1%) was used because acidic 324 methanol can penetrate deeply into cells, disrupting the cell membrane. In addition, acidic 325 methanol can dissolve and stabilize polar compounds such as phenolics and anthocyanins 326 [35]. This research shows that RR and OSP are rich sources of phenolic compounds. It 327 has previously been reported that RR has a high content of phenolic compounds [36], and 328 that these compounds are primarily accumulated in the aleurone layer and bran of RR 329 [37]. Thus, the rice milling process to remove the husk plays a vital role in preventing the 330 loss of various beneficial compounds. Previously, it was suggested that ferulic acid, p-331 coumaric acid, and procatechuic acid are the most abundantly found phenolic compounds 332 in RR [37]. 333

Here, cooking led to a 49% decrease in the phenolic compounds of RR. Heating of RR 334 generally destroys the structure of phenolic compounds by breaking the esterified and 335 glycosylated bonds, thus decreasing the quantified content of phenolic compounds in 336 cooked RR [38]. Our results agree with previous findings [39,40], which reported high 337 levels of total phenolic compounds, mostly gallic acid, chlorogenic acid, pro-catechuic 4-338 hydroxybenzoic acid, and salicylic acid, in different varieties of OSP. However, phenolic 339 contents were broken down by the heating process. Regarding the flake products, flakes 340 containing a higher proportion of RR showed a higher content of phenolic compounds. 341 Nevertheless, the processing lowered the phenolic compounds in flakes when compared 342 to raw OSP and RR. Boiling and baking have previously been reported to be responsible 343 for the loss of phenolic compounds of RR-based products [41] 344

A similar trend was observed in the anthocyanin content of RR. The cooking process 345 resulted in a significant decrease in the anthocyanin content due to the unstable property 346 of anthocyanins when exposed to high temperatures [42]. Anthocyanins were only de-347 tected in RR in this research. Anthocyanins such as cyanidin 3 glucoside, delphinidin 3 348 glucoside, and peonidin are reported to have health-promoting properties [43]. Thus, ex-349 posure to high temperatures for extended periods should be avoided to reduce the risk of 350 anthocyanin breakdown. The anthocyanin content in flakes was lower than in the raw 351 samples, and increasing the proportion of RR resulted in a higher anthocyanin content of 352 flakes. The decrease in the anthocyanin content of flakes is possibly due to the heat treat-353 ment during flake production, typically involving two high temperature processing steps 354 of pregelatinization and flaking. It has been reported that the high-temperature treatment 355 used for food processing can lead to a reduction of the anthocyanin content [44] 356

Moreover, there is a growing research interest in the conversion of  $\beta$ -carotene and  $\alpha$ carotene absorbed in the duodenum to retinol by intestinal enzymes [45]. The high rate of vitamin A deficiency in the world and the detrimental effects caused by the condition have necessitated the search for foods that can supply sufficient amounts of daily vitamin 360

A requirement. In this research, OSP had the highest content of  $\beta$ -carotene. However, boil-361 ing of OSP decreased the  $\beta$ -carotene by approximately 41% of the  $\beta$ -carotene available. 362 This phenomenon could be due to the thermal breakdown of  $\beta$ -carotene. Moreover, carot-363 enoids are well known as substances that are sensitive to light and high temperatures [46]. 364 RR contains different types of carotenoids [47]. Here, an increase in carotenoids after boil-365 ing was found, which could be related to the thermal disruption of the protein-carotenoid 366 complex and the consequent release of carotenoids from the matrix. Similar findings have 367 been published [48]. Similar trends were also observed in the OSP and RR-based flake 368 products. The  $\beta$ -carotene was significantly lower in the cooked sample compared to the 369 raw sample. The OSP proportion in the flake formulation affected the carotenoid content. 370 The higher the OSP proportion, the greater the carotenoid content of the flakes. Also, the 371 carotenoid contents of the flakes were significantly lower than the raw material. The sim-372 ultaneous heating process from pre-gelatinization to flake pressing could further break 373 down the carotenoids. This finding is supported by previously published work, which 374 shows that heat treatment during food processing is responsible for the loss of carotenoids 375 to degradation[49] 376

Vitamin E deficiency could lead to severe neurological problems. The main vitamin 377 E compounds are tocopherols, with both  $\alpha$  and  $\gamma$  tocopherol providing most vitamin E 378 activity. Both samples had high contents of  $\alpha$  and  $\gamma$  tocopherol. A significant increase 379 (74%) in  $\alpha$  tocopherol was found in OSP after boiling. This result indicates that heat treat-380 ment could be beneficial for the bioaccessibility of tocopherol. Furthermore, heat treat-381 ment can assist in the breakdown of complex foods. Thus, tocopherol can be quickly re-382 leased from its binding site. On the contrary, heat treatment was reported to reduce the 383 tocopherol content in corn [50]. The increase in tocopherol after boiling indicates that to-384 copherols in the sample are more stable to heat treatment than other foods. In this re-385 search, RR had a considerably high tocopherol content. Therefore, boiling could have re-386 leased the tocopherols from their binding site, facilitating their extraction and detection. 387 Moreover, the structure of the rice grains could have played a role. RR has a compact 388 structure. Therefore, cooking could assist the extraction of tocopherols, increasing the ex-389 tractable tocopherol content, although some of the tocopherols might be lost to high tem-390 perature. The tocopherol contents of the different flake products were lower than the raw 391 and boiled OSP and RR. It could be due to the simultaneous or prolonged exposure to 392 heat treatment. Unlike the conventional method of cooking, which increases the tocoph-393 erol content, extended heating breaks the matrix structure flakes and significantly affects 394 the tocopherols. Thus, prolonged heating should be avoided in the processing of healthy 395 food products to retain their tocopherols. Heat treatment combined with mechanical treat-396 ment in specific conditions could release the tocopherols from the food matrix. However, 397 extended exposure will lead to the breakdown of carotenoids in the sample. 398

This research measured the antioxidant activity of raw and cooked OSP and RR and 399 their flake products using DPPH, FRAP, and Superoxide radical scavenging capacity as-400 says. The result showed that heat treatment was responsible for decreasing the antioxi-401 dant activity of OSP, RR, and the flake products. A positive correlation was observed be-402 tween the reduction of bioactive compounds and antioxidant capacity. Most bioactive 403 compounds are heat sensitive, which influences their antioxidant activity. A previous 404 study reported that an increase in temperature accelerated the initiation of oxidation, pre-405 venting antioxidant compounds from working optimally [51]. The bioactive compounds 406 were degraded, experiencing structural changes, and wholly transformed into inactive 407 substances. Nevertheless, due to the complex nature of antioxidant compounds in plants, 408 their thermal stability varies. Some compounds such as procatechuic acid, p-coumaric 409 acid, and ferulic acid have high thermal stability, which facilitates their extraction and 410 antioxidant activity [52]. The heat treatment process assists in the release of such com-411 pounds without affecting their activity. Moreover, the flake products exhibited lower an-412 tioxidant activity than the raw or boiled products. Based on the percentage values of 413 DPPH and Superoxide radical scavenging capacity and the content of Fe [II] formed, 414 flakes containing only OSP or RR had lower antioxidant activity. Interestingly, the combination of OSP and RR increased the antioxidant activity, probably due to the synergistic effect of bioactive compounds from OSP and RR [53]. Even though the processing reduced the antioxidant activity in boiled samples and flake products, the remaining antioxidant activity was still considerably high. Therefore, boiled OSP, RR, and their flake products are a good source of bioactive compounds and antioxidants. 410

Regarding the physicochemical and sensory properties of the flakes, it was observed 421 that the moisture content of flake products was mainly influenced by their composition. 422 Starch is the dominant carbohydrate found in OSP and RR, and OSP has a lower amylose 423 content compared to RR. According to Wang et al. [54], the amylose content of OSP is 424 18.71%, while Markus et al. [55] reported that RR has 23% amylose content. Amylose is a 425 linear polymer of glucose, which forms starch. The higher the amylose content, the greater 426 the moisture content of the dough due to a higher capability to absorb water. The absorbed 427 water will promote dough gelatinization during heating. The water absorbed by the 428 dough will then evaporate during the flaking process due to the network's inability to 429 entrap water during the pre-gelatinization heat treatment. The high flaking temperature 430 will detach water from the matrix structure of the flakes, resulting in increased evapora-431 tion and a lower moisture content of the flake products. 432

The dietary fiber content of the flakes ranged between 9.47 and 13.86%, comparable 433 to values commonly found in breakfast cereals such as corn flakes, rice, quinoa, millet, 434 and amaranth flakes [56]. The proportion of RR affected the dietary fiber of flakes. The 435 fiber content of flakes was also influenced by the heating and pressing processes. Heat 436 treatment caused the degradation of the fiber matrix and the glycosidic bond. The degradation affected the solubility level of the fiber, i.e., the ratio between soluble and insoluble 438 fiber, thus resulting in the reduction of total fiber in the product. 439

Water absorption index (WAI) is a physical property associated with the ability of 440 flakes to absorb water molecules within a particular time. Absorbed water molecules 441 could be bound or detained in matrix pores of flakes. The water absorption index is crucial 442 because it is associated with the quality of the flakes. Consumers can experience the crispy 443 and crunchy sensation of the flakes after soaking in milk or water. In contrast, the higher 444 WAI is interrelated with the unwanted soggy texture of flakes. The WAI is influenced by 445 the porosity of the matrix on flakes, thickness, and hygroscopicity of flakes. In addition, 446 the presence of fiber and protein could assist the flakes in absorbing water into their struc-447 ture. The result shows that increasing the proportion of RR reduces the WAI due to a 448 decrease in hygroscopicity. OSP is a rich source of sugar; thus, it has higher hygroscopicity 449 compared to RR. In addition, the presence of RR affects the network construction of flakes 450 by inhibiting the formation of the starch-protein structure, which could entrap gas. The 451 compact structure created by RR starch inhibits water absorption into the matrix of the 452 product. On the other hand, RR has higher fiber and protein contents, which help improve 453 water absorption [57]. Moreover, the suitable fragmentation of the amylose and amylo-454 pectin chain in sweet potato could also affect the water absorption capacity. The heating 455 processes such as roasting, flaking, and extrusion will induce starch fragmentation with 456 sufficient water. The gelatinization process converts starch to a digestible material and 457 plays a vital role in determining the structural properties of flakes and their ability to ab-458 sorb moisture. The presence of RR in the flake dough disturbs the composition of starch, 459 thus inhibiting the flakes from forming a porous structure and affecting their water ab-460 sorption capacity. 461

Fracturability is a physical property related to deformation conditions when a specific maximum force is applied. A higher fracturability value represents the ability of food products to maintain their structure when force is applied. According to Table 2, flakes with 100% OSP has a higher fracturability value compared to others as the homogenous matrix of starch, protein, and fiber in OSP enables the interaction between the matrix and water molecules, leading to the firm, sturdy and rigid texture of flakes. The rigid texture is related to the evolution phase of starch from amorphous conditions, which indicates complete disorganization of the crystalline structure of starch. Increasing the proportion 469 of RR in the flakes reduces the fracturability value. The mixture of OSP and RR decreases 470 the rigidity of the flakes due to the different structural properties of the two samples, cre-471 ating flakes that are susceptible to fracture. Crispness is a complex texture attribute be-472 cause it comprises a combination of sensory analysis, acoustical procedure, and instru-473 mental analysis. The instrumental analysis revealed that flakes with OSP to RR ratios of 474 100:0 and 0:100 had higher crispness values than others. Similarly to fracturability, the 475 crispness value decreased with an increasing RR proportion in the flakes. The mixture of 476 ingredients with different structural properties can affect the crispness value of flakes. 477

The color profile shown in Table 3 revealed that the color of flakes was affected by 478 the pigments in the raw materials used for their production. The red color of RR is asso-479 ciated with anthocyanins found in its bran layer. The yellowness of OSP is linked to ca-480 rotenoids, primarily  $\beta$ -carotene. Previous research reported that carotenoids are easily ox-481 idized and undergo color degradation due to thermal treatment [58]. The hue results 482 showed a flake color range between yellow and red. The color of flakes was also affected 483 by the Maillard reaction product. The higher the Maillard reaction product, the darker the 484 appearance of flakes. 485

The sensory analysis involved a preference test of the color, taste, crispness, and 486 mouthfeel. Flakes containing 100% OSP had the lowest color preference score, associated 487 with the orange appearance, and the lowest brightness score. Most panelists perceived the 488 dark orange color as less fresh, less attractive, and less tasty. The color preference was 489 increased with the addition of RR, which also helped improve the product lightness and 490 redness values. The panelists were mostly in favor of flakes that appeared brighter and 491 reddish. The higher brightness level is attributed to the white endosperm color of RR, 492 while the redness is related to the anthocyanins in the bran of RR. The mouthfeel prefer-493 ence of flakes is associated with the water absorption index. The higher the ability of flakes 494 to absorb water, the greater the plasticizing effect due to the presence of more hydrophilic 495 components such as the phosphate monoester found in sweet potato starch [59]. The pan-496 elists generally preferred flakes with soft mouthfeel. The taste and crispness preferences 497 for OSP and RR-based flakes were in the range of "indifferent" and "slightly likes." Increas-498 ing the RR proportion in flakes decreases the bitterness intensity and increases the savory 499 taste. The perception of savory taste is generally influenced by the moisture content and 500 the flavor of RR. Niu et al. [60] suggests that sweet potato with a low dry matter content 501 has a bitter taste, and increasing the dry matter reduces the bitterness. On the other hand, 502 a decrease in invertase activity may support the bitter aftertaste of the sweet potato. 503

This research has successfully monitored the changes of the bioactive compound and 504 antioxidant activity of OSP and RR in their native form and the flake product. It can be 505 observed that individual compounds were acted differently to processing methods. 506 Therefore, it can be suggested that research on the development of functional foods 507 should address the products ready to be consumed instead of solely focusing on the raw 508 materials due to the changes that have taken place during the transformation. This ap-509 proach should be implemented for other potential materials rich in bioactive compounds. 510 Moreover, further consideration of the bioaccessibility and bioavailability that are affected 511 by digestion and absorption in the human metabolism system should also be considered 512 [61]. 513

## 5. Conclusions

OSP and RR are rich sources of bioactive compounds, especially  $\beta$  carotene, for OSP, 515 and phenolic compounds and anthocyanins, for RR. The boiling process significantly decreased most of the bioactive compounds, except tocopherols and  $\alpha$ -carotene. The level of bioactive compounds in the flake products was dependent on the proportion of OSP and 518 RR. Heat treatment resulted in a decrease in antioxidant activity, even though the remaining activity was still considerably high. The mixture of OSP and RR can produce flakes 520 with low moisture and high fiber contents. The optimum flake water absorption index, 521

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	fracturability, and crispness were obtained by combining 40% OSP and 60% RR. Moreover, the ratio of OSP and RR influenced the color and sensory preferences of the panelists.
	<b>Author Contributions:</b> "Conceptualization, I.R.A.P.J; methodology, I.R.A.P.J, S.R., T.I.P.S., and L.M.Y.D.D.; formal analysis, I.R.A.P.J and C.W.; investigation, S.R., I.R.A.P.J.; resources, L.M.Y.D.D., T.I.P.S; data curation, I.R.A.P.J and L.M.Y.D.D; writing—original draft preparation, I.R.A.P.J and L.M.Y.D.D.; writing—original draft preparation, C.W; project administration, I.R.A.P.J., S.R.; funding acquisition, I.R.A.P.J., S.R., T.I.P.S All authors have read and agreed to the published version of the manuscript."
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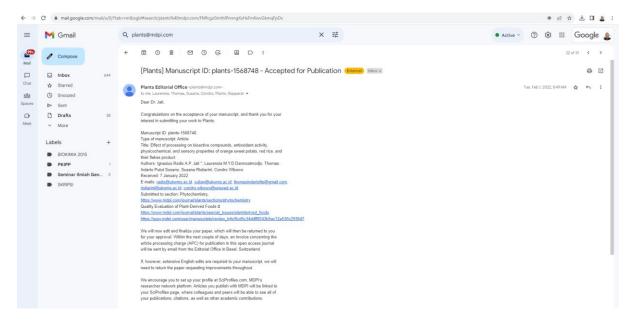
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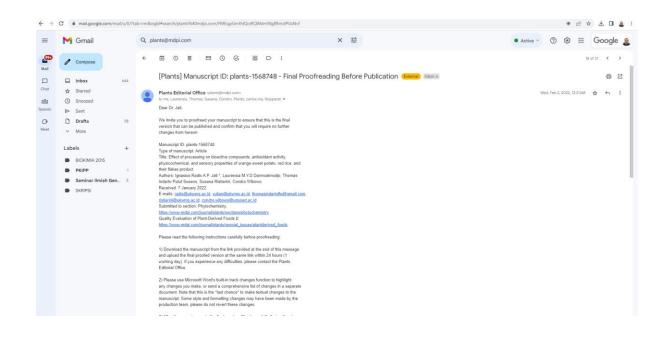
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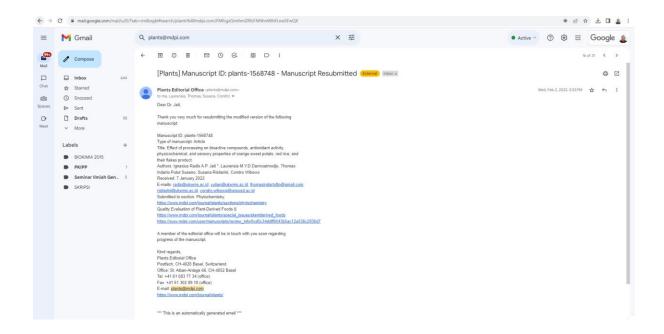
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Article

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## Effect of Processing on Bioactive Compounds, Antioxidant Activity, Physicochemical, and Sensory Properties of Orange Sweet Potato, Red Rice, and Their Application for Flake Products

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Abstract: Orange sweet potato (OSP) and red rice (RR) are rich sources of health benefit-associated substances and can be conventionally cooked or developed into food products. This research approach was to closely monitor the changes of bioactive compounds and their ability as antioxidants from the native form to the food products which are ready to be consumed. Moreover, this research explored the individual carotenoids and tocopherols of raw and cooked OSP and RR and their developed flake products, and also investigated their antioxidant activity, physicochemical properties, and sensory properties. Simultaneous identification using the liquid chromatographic method showed that OSP, RR, and their flake products have significant amounts ( $\mu g/g$ ) of  $\beta$ -carotene (278.58-48.83), α-carotene (19.57-15.66), β-cryptoxanthin (4.83-2.97), α-tocopherol (57.65-18.31), and also γ-tocopherol (40.11–12.15). Different responses were observed on the bioactive compound and antioxidant activity affected by heating process. Meanwhile, OSP and RR can be combined to form promising flake products, as shown from the physicochemical analysis such as moisture (5.71-4.25%) and dietary fiber (13.86-9.47%) contents, water absorption index (1.69-1.06), fracturability (8.48-2.27), crispness (3.9-1.5), and color. Those quality parameters were affected by the proportions of OSP and RR in the flake products. Moreover, the preference scores (n = 120 panelists) for the flakes ranged from slightly liked to indifferent. It can be concluded that OSP and RR are potential sources of bioactive compounds which could act as antioxidants and could be developed into flake products that meet the dietary and sensory needs of consumers.

Keywords: orange sweet potato; red rice; flakes; bioactive compound; antioxidant activity; physicochemical; sensory properties

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### 1. Introduction

Modern food trends and lifestyle changes have strongly influenced the dietary habits of society. The demands for ready-to-eat and simple-to-prepare foods are increasing rapidly, providing an excellent opportunity for food industries to play a significant role in supplying such food products. Flakes, one of the most popular foods made from cereals, typically oat, corn, and barley, are commonly served for breakfast with milk in Europe and the USA, and their global appeal is gaining traction. In addition, healthy eating has become a new trend in modern culture, with consumers increasingly opting for healthy

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food options. Secondary metabolites found in plants have been shown to decrease the risks of degenerative diseases such as coronary heart disease, diabetes, cancer, and stroke [1–4]. Thus, innovative functional food products rich in bioactive compounds are needed to promote a healthy diet and reduce disease risks. Asia has the most rapidly growing food product market, so using local ingredients can help open a new market for flakes and decrease the reliance on imported foods such as oat and barley. Commodities that can potentially be developed into flakes are red rice (RR) and orange sweet potato (OSP).

RR (*Oryza nivara* L.) is a variety of rice with red pericarp caused by anthocyanins in the aleurone layer. It is a rich source of anthocyanins such as cyanidin 3-O-glucoside and peonidin 3-O-glucoside [2]. Anthocyanins have also been reported to inhibit plaque formation [3] and to exhibit hypocholesterolemic [4] and anticancer effects [5] in RR. The health benefits of RR have also been linked to bioactive compounds such as tocopherol and tocotrienols [6,7] and dietary fiber [8]. RR also contains higher essential minerals than white rice, including iron, zinc, and vitamins, which are especially important for babies and toddlers, [9]. Despite the various health benefits of RR, its consumption remains low. It is generally considered inferior to white rice due to the hard texture and unpleasant aroma when cooked. Traditionally, RR is consumed steamed or boiled, and its sensory properties are inferior to white rice. The most popular RR-based product is RR baby porridge, but there are reports on the development of RR-based product such as pasta [10], noodles [11], flakes [12], rice milk [13], and fermented beverages [14]. However, these have not been scaled up or commercially established.

OSP (Ipomoea batatas) is one variety of sweet potato with a bright orange flesh color caused by carotenoids, of which high amounts of  $\beta$ -carotene,  $\alpha$ -carotene, and  $\beta$ -cryptoxanthin have been reported [15–17]. In many countries, OSP has been used to eradicate vitamin A deficiency due to its high content of beta-carotene, a pro-vitamin A carotenoid. Sweet potato was extensively promoted in Africa and some Asian countries with remarkable results [18-20]. However, there were difficulties in ensuring its sustainability due to the monotonous method of preparation, mainly boiled and baked, even though the essential nutritional components were reportedly retained after processing [21]. Processing OSP could decrease the beta-carotene, but not below the recommended dietary level [22]. Moreover, food prepared from OSP by the traditional method was not attractive to children, who were the main target of the vitamin A intake enhancement. Therefore, innovative food products need to be developed to promote the consumption of RR and OSP. Numerous research has been published to explore the potency of various plant sources as functional foods. However, the approach mainly investigates the plant materials in the native or raw form. In contrast, the consumer will generally consume after the materials undergo transformations which could affect the characteristics of the products [23]. Moreover, besides the processing, the bioactive compounds and antioxidant activity will be further affected by the in vivo digestion and the intestinal absorption rate of the body metabolism before providing bioavailable compounds that can be used [24]. This research approach was to closely monitor the changes of the bioactive compound and antioxidant activity from the raw materials to the ready to be consumed food products, and aimed to investigate the bioactive compounds and antioxidant activity of raw and cooked RR and OSP and the physicochemical and sensory properties of their developed flake products.

#### 2. Materials and Methods

#### 2.1. Plant Materials and Chemicals

A local variety of RR (*Oryza nivara* L.), "Cempo abang", and OSP (*Ipomoea batatas* L.), "Mendut", were collected from farmers in Yogyakarta province, Indonesia. Chemicals used for analysis, including distilled water, Folin–Ciocalteu, 1,1-diphenyl-2-picrylhydrazyl (DPPH), gallic acid, butylated hydroxyl toluene (BHT), enzymes (thermamyl, pancreatin, pepsin), riboflavin, methionine, and nitroblue tetrazolium (NBT), were purchased from Sigma Chemical. Methanol, Whatman 40 filter paper, n-hexane, NaOH, HCl, ethanol, and phosphate buffer (pH 6) were purchased from Merck, Germany. Carotenoid standards, including β- and α-carotene, β-cryptoxanthin, lycopene, and lutein, and α- and γ-tocopherol standards were obtained from Sigma-Aldrich.

#### 2.2. Sample Preparation

#### 2.2.1. Raw Samples

The RR samples were washed, drained, and blended (Philips food processor). The OSP samples were chopped into small pieces. All samples were then freeze-dried, refined, and sieved (30 mesh). Finally, the powdered samples were placed in dark bottles and stored in a refrigerator (4 °C) for further usage.

#### 2.2.2. Boiled Samples

RR (75 g) was cooked with 135 g of tap water (1:1.8; w/w) in a rice cooker (Panasonic) for approximately 45 min. After, the cooked rice was cooled for 10 min. Meanwhile, 150 g of OSP was boiled in an aluminum pot using tap water for 20 min, cooled for 10 min, and then mashed. Both samples were freeze-dried, refined, sieved (30 mesh), and stored (4 °C) in a refrigerator.

#### 2.2.3. Flakes Production

RR was placed in a cabinet dryer (60 °C) for 1 h. The OSP was peeled, sliced and placed in a cabinet dryer (60 °C) for 6 h. The dried OSP and RR were mashed using a blender. The flour was passed through an 80 mesh. Flakes were produced using six different proportions of OSP and RR, namely 100:0, 80:20, 60:40, 40:60, 20:80, and 0:100. Salt (3% *w/w*), sugar (30% *w/w*), and water (150% *w/w*) were mixed in as additional ingredients. The mixture was heated at 75 °C for 1 min and pressed at 170 °C for 1 min using a customized flake pressing tool. The pressed flakes were cut at 2 × 2 cm and dried using an oven at 125 °C for 5 min.

#### 2.3. Bioactive Compounds and Antioxidant Activity Analysis

### 2.3.1. Methanolic Extract of Samples

The extraction of samples (raw, boiled, and flakes) was performed according to a previously published procedure [25]. Briefly, 1 g of sample was weighed, ground, placed in a centrifuge tube, and then extracted with 10 mL of 1% methanol–HCl solution. The mixture was then vortexed for 15 min, centrifuged at 5000 rpm for 15 min, filtered (Whatman No. 40), and then used for antioxidant activity analysis, performed in triplicate.

#### 2.3.2. Phenolic Content

The total phenolic content was determined using the Folin–Ciocalteu method by Singleton and Rossi as described in other published work [26]. In brief, 0.1 mL of extract was mixed with 0.5 mL 1:1 Folin–Ciocalteu reagent and distilled water. After 10 min, 4.5 mL of 2% sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>) was added, and the mixture was then vortexed and kept in the dark for 1 h. The blue complex formed was measured using a spectrophotometer at 765 nm. Methanol and gallic acid were used as the blank and standard, respectively. The results were calculated as milligram gallic acid equivalents (GAE)/100 g dry weight.

#### 2.3.3. Anthocyanin Content

The total anthocyanin in RR and flake samples was determined spectrophotometrically using the pH differential method [27]. In brief, 1 mL of extract was diluted in pH 1.0 and pH 4.5 buffers. The absorbance was measured at 510 and 710 nm. The final absorbance was calculated using the formula:

A = [(A513-A700)pH 1.0 - (A513-A700)pH 4.5)]

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The calculated absorbance was then used to calculate the total grams of anthocyanins per 100 g dry weight, with a molar extinction coefficient of 26,900 and a molecular weight of 445.

#### 2.3.4. Carotenoid and Tocopherol Analysis

The carotenoids and tocopherols in the raw and boiled samples of RR and OSP were determined simultaneously using High-Performance Liquid Chromatography (HPLC) based on previously published research [24]. In brief, 0.3 g of finely ground samples were extracted with 0.5 mL of 70% ethanol and 0.4 mL of n-hexane under yellow lights. In addition, 0.4 mL  $\beta$ -Apo-8′ carotenal-O-methyloxim and 0.4 mL  $\alpha$ -,  $\gamma$ -tocopherol were used as an internal standard for carotenoids and tocopherols, respectively. The mixture was shaken for 20 min, centrifuged at 5000 rpm, 4 °C for 20 min. Next, the upper layer of extract containing a hexane fraction was collected using a micropipette. The extraction was repeated four times using only n-hexane as a solvent. Finally, the hexane fractions were pooled and completely dried using nitrogen gas.

Before injection, the extracts were mixed with 200  $\mu$ L of ethanol containing 30  $\mu$ g/mL BHT, then 20  $\mu$ L of the mixture was injected into the HPLC (Varian Pro Star 410, Spark, The Netherlands). A mixture of 82% acetonitrile, 15% dioxan, 3% methanol,  $0|\mu$  M ammonium acetate, and 0.1% triethylamine was assigned as the mobile phase and was pumped at a rate of 1.6 mL/min. The solvent was pre-mixed to avoid dependency on reproducible mixing by the pump. For separation, a C18 Spherisorb ODS 2 column (3  $\mu$ m, 250 × 4.6 mm) was applied. In addition, a UV Vis detector at 450 nm and a Scanning Fluorescence detector using an excitation wavelength of 295 nm and an emission wavelength of 328 nm were used to monitor the carotenoids and tocopherols, respectively. Five repetitions were performed for the HPLC analysis.

## 2.3.5. DPPH Radical Scavenging Activity

The radical scavenging activity of the extract was examined by the DPPH method [25]. In brief, 1 mL of extract was mixed with 2 mL of 0,2 M DPPH and 2 mL of methanol in centrifuge tubes, vortexed, and kept in the dark for 1 h. The absorbance was measured spectrophotometrically at 517 nm. As a control, 150 ppm BHT solution was used. The DPPH radical scavenging activity of the extract was expressed as a percentage calculated as follows: % radical scavenging capacity = ((Absorbance of control Absorbance of sample)/Absorbance of control N0%

#### 2.3.6. FRAP Assay

The ferric reducing antioxidant power (FRAP) was examined based on a previously published report [28]. In brief, a mixture of 60  $\mu$ L extract, 180  $\mu$ L distilled water, and 1.8 mL FRAP reagent was vortexed and incubated at 37 °C for 30 min. The spectrophotometer was used to read the absorbance of the mixture at 593 nm. A standard curve was prepared with Fe [II] (FeSO4.7H<sub>2</sub>O, 100–2000 mM) to calculate the reducing power. The result was expressed as mmol Fe[II]/g. In addition, methanol was used for the reagent blank.

#### 2.3.7. Superoxide Radical Scavenging Capacity

A previous report [29] was followed to examine the Superoxide radical scavenging capacity. Firstly, a reagent containing riboflavin, methionine, and NBT in 0.05 M phosphate buffer pH 7.8 was prepared. Then, 100  $\mu$ L of the extract was mixed with 4.9 mL of reagent and illuminated (20 W fluorescent lamp) at 25 °C for 25 min. The absorbance was measured at 560 nm.



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2.4. Physicochemical Properties of Flake Products

2.4.1. Moisture and Dietary Fiber Contents

The moisture content was measured thermogravimetrically [30]. In brief, 1 g of each sample was dried at  $105 \pm 0.2$  °C to establish a constant mass. The analysis was performed in triplicate.

#### 2.4.2. Water Absorption Index

The water absorption index was examined according to previously published method [31]. In brief, 5 g of flakes was placed in a 100 mL beaker, 30 mL of water at 30 °C was added. After 10 s of immersion, the flakes were dried. The water absorption index was calculated using the formula: WAI = (wf  $|wi\rangle/wi$ , where wi and wf are the initial and final weight of the sample, respectively.

#### 2.4.3. Fracturability, and Crispness, and color profiles

The fracturability and crispness of flakes were measured using TA-XT Plus Texture Analyzer [Stable Micro Systems, Surrey, UK) [32]. The probe used was a ¼ inch spherical stainless-steel probe (P0.255). The sample was placed on the sample holder, and then the probe was moved down to press the sample. The results were obtained in the form of a graph (force vs. time) (the graph is not shown). The value of the *y*-axis at the graph's highest point is the maximum force value that can be held by the sample, called the value of fracturability. Crispness can be measured through changes in displacement distance during a drastic decline in the graph pattern from the highest peak to the next peak point. Meanwhile,

the color profiles of flakes were measured using color reader Konica Minolta CR-10 (Konica Minolta, Osaka, Japan). The results were expressed as Lightness (L\*), redness (a\*), yellowness (b\*), ohue (oh), and Chroma (C). The fracturability and crispness of flakeswere measured using TA. XT Plus Texture Analyzer (Stable Micro Systems, UK) [32]. The probe used was a ¼ inch spherical stainless steel probe (P0.25S). The sample wasplaced on the sample holder, and then the probe was moved down to press the sample. The results were obtained in the form of a graph (force vs. time) (the graph is notshown). The value of the y axis at the graph's highest point is the maximum force value that can be held by the sample, called the value of fracturability. Crispness can be measured through changes in displacement distance during a drastic decline in the graphpattern from the highest peak to the next peak point.

#### 2.5. Sensory Analysis

The sensory evaluation was conducted by 120 untrained panelists to determine the level of consumer preference for the flakes with various proportions of OSP and RR. The parameters tested were preferences for color, taste, crispness of flakes, and mouthfeel when served with milk. The Hedonic Scale Scoring method (preference test) with a scale ranging from 1 (strongly disliked) to 7 (strongly liked) was used for the sensory test.

Samples of flakes for the color preference test were prepared in open white plastic containers. Panelists were asked first to assess aspects of flakes' taste, color, and crispness before serving with milk. The crispness was evaluated based on the panelist's preference for the sound of flakes during biting. For the mouthfeel test, 5 g of flakes were prepared in a small plastic container. Panelists were instructed to pour 10 mL of milk into the container and wait for 1 min. Then, they were asked to assess the mouthfeel of the flakes based on preference level by filling the questionnaire sheet provided.

#### 2.6. Statistical Analysis

The data were statistically analyzed using ANOVA ( $\alpha$  = 5%) followed by Duncan's Multiple Range Test (DMRT) on SPSS software version 19. Spider web chart analysis using

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Microsoft Excel was used to determine the best proportion of OSP and RR in flakes based on the panelists' preferences.

#### 3. Results

3.1. Bioactive Compounds of OSP, RR, and Their Flake Products

Table 1 shows the phenolic compound, anthocyanin, carotenoid, and tocopherol contents of raw and cooked OSP and RR and their flakes containing different proportions of OSP and RR. The raw RR had a higher phenolic content than OSP. As a result, the higher the proportion of RR, the higher the phenolic content of the developed flakes. It was also found that cooking decreased the phenolic content of RR and OSP by approximately 49.34% and 41.08%, respectively.

Interestingly, the flakes with 100% RR showed a higher phenolic content than the cooked RR. Anthocyanin was only observed in RR. Furthermore, flakes containing 100% RR had a lower anthocyanin content than the conventionally cooked RR.

Table 1. Bioactive compounds of orange sweet potato (C	OSP), red rice (RR) and the flake products.
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	0	SP	RR		Proportions of OSP and RR in Flakes					
	Raw	Cooked	Raw	Cooked	100:0	80:20	60:40	40:60	20:80	0:100
Phenolic (mg	$110.68 \pm$	65.21± 7.3	301.89 ±	$152.91 \pm$	$77.46 \pm$	$97.34 \pm$	$102.03 \pm$	131.79 ±	$146.09 \pm$	$162.40 \pm$
GAE/100 g DW)	18.3 a	b	24.86 a	28.92 <sup>b</sup>	8,28 a	8.79 <sup>b</sup>	11.65 c	10.93 d	15.64 e	21.54 f
Anthocyanin	nd	,	8,81±	$8.64{\pm}~0.08$		$1.74 \pm$	$2.09\pm0.05$	3.73 ±	$5.81 \pm$	
(mg/100 g DW)	nd	nd	0.05 a	а	nd	nd	0.06 c	d	0.07 e	$0.04  \mathrm{f}$
l annataria (u.g./g)	$278.58 \pm$	134.17±	$13.17 \pm$	$7.37\pm0.5$	48,83 $\pm$	$36.27 \pm$	$25.77 \pm$	27.23 ±	$15.69 \pm$	$3.12 \pm$
β-carotene (µg/g)	31.5 ª	17.2 в	2.62 ª	b	3,31 ª	3.01 <sup>b</sup>	3.45 c	2.72 <sup>d</sup>	2.21 <sup>e</sup>	$0.66^{\rm f}$
a anatono (ua/a)	$19.57 \pm$	$23.83 \pm$	$5.53 \pm 1.4$	11.66 ±	$15.61 \pm$	$11.82 \pm$	$5.31 \pm$	$2.53\pm0.87$	nd	nd
α-carotene (µg/g)	1.8 a	1.6 <sup>b</sup>	а	1.5 в	1.44 a	3.11 в	1.59 °	d	na	na
β-cryptoxanthin	$4.83\pm0.2$	4.48± 0.8ª	$3.67 \pm$	$3.96 \pm 1.9$	$2.64 \pm$	$2.81 \pm$	2.97 ±	$2.78\pm0.12$	2.77 ±	$2.81 \pm$
(µg/g)	a	4.40± 0.0°	2.15 ª	a	0,05 a	0.13 <sup>b</sup>	0.08	с	0.25 c	0.16 <sup>b,c</sup>
Lutein (µg/g)	3.77 ± 0.8	3.81 ± 0.7	2.16 ± 0.8	1.82 ± 0.5 ь	nd	nd	nd	nd	nd	nd
	13,23 ±	15.11 ±	$34.08 \pm$	57.65 ±	$4.58 \pm$	$7.34 \pm$	$10.51 \pm$	12.45 ±	$16.82 \pm$	$18.31 \pm$
$\alpha$ -tocopherol (µg/g)	1.1 ª	0,5 <sup>b</sup>	2.2 <sup>a</sup>	2.1 <sup>b</sup>	0.73 <sup>a</sup>	1.49 <sup>b</sup>	1.27 °	1.21 °	0.52 <sup>e</sup>	0.77 f
a to comborol (ug/g)	$2.40\pm0.2$	$5.38 \pm 0.05$	29.27 ±	$40,11 \pm$	nd	nd	$3.38 \pm$	$6.71 \pm 1.19$	$8.06 \pm$	$12.15 \pm$
γ-tocopherol (µg/g)	а	b	2.4 ª	1.8 <sup>b</sup>	nd	nd	1.22 °	d	0.98 <sup>e</sup>	$0.73^{\rm f}$

Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–f) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each category (OSP, RR, and Flakes).

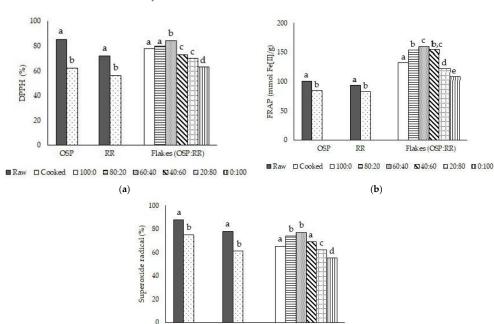
Of the carotenoids, OSP had a higher content of  $\beta$ -carotene,  $\alpha$ -carotene,  $\beta$ -cryptoxanthin, and lutein.  $\beta$ -carotene and  $\beta$ -cryptoxanthin were the dominant carotenoids observed in RR. The cooking process significantly decreased the content of  $\beta$ -carotene in OSP and RR by roughly 48% and 56%, respectively. Overall, flakes containing higher amounts of OSP showed higher carotenoid contents. The processing significantly decreased the carotenoids in the flake products when considering the raw forms and the proportions of OSP and RR, and  $\beta$ -carotene was the major carotenoid remaining in the flake products.

Moreover, raw RR contained higher  $\alpha$ -tocopherol than OSP in raw forms. Unlike the decreasing trend observed in other bioactive compounds due to processing, the tocopherol and  $\alpha$ -carotene contents of both RR and OSP increased after conventional cooking. On the other hand, the two-step thermal processing decreased the tocopherol content of flakes.

3.2. Antioxidant Activity of Raw and Cooked OSP and RR and Their Flake Products

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The antioxidant activity of raw and cooked OSP and RR and their flake products were examined using DPPH, FRAP, and Superoxide radical scavenging activity methods. Figure 1a shows the DPPH scavenging activity of methanolic extract of RR, OSP, and the flake products. Boiling affected the ability of the methanolic extract to scavenge DPPH radicals. Approximately 16% and 23% decreases were observed in cooked RR and OSP, respectively. The combination of OSP and RR in the ratio of 60:40 resulted in flakes with the highest antioxidant activity (84%). The results trend indicated that the higher proportion of OSP contributed to the more robust antioxidant capacity of the extract. Furthermore, the DPPH result was in agreement with FRAP (Figure 1b) and Superoxide scavenging capacity (Figure 1c). Thus, conventional cooking and flake processing methods reduce the antioxidant activity of extracts of OSP and RR compared to their raw forms, and the right combination of OSP and RR in the flake formulation is critical for higher antioxidant activity.



■Raw □Cooked □100:0 ■80:20 ■60:40 ■40:60 ■20:80 □0:100

RR

OSF

#### (c)

**Figure 1.** Antioxidant activity of extract determined by (a) DPPH, (b) FRAP, and (c) Superoxide radical assays. Different superscript letters (a- $\frac{1}{12}$ ) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each category (OSP: orange sweet potato, RR: red rice, and flakes).

Flakes (OSP:RR)

#### 3.3. Physicochemical Properties of OSP- and RR-Based Flake Products

The proportion of OSP and RR in the flake formulation affected the moisture content of flakes. A lower OSP proportion resulted in a lower moisture content of flakes (Table 2). The dietary fiber content increased with a higher proportion of RR. The dietary fiber content of flake products ranged from  $9.47 \pm 0.01\%$  to  $13.86 \pm 0.73\%$ .

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	Proportions of OSP and RR in Flakes						
	100:0 80:20 60:40 40:60 20:80					0:100	
Moisture content (%)	$5.71 \pm 0,07$ a	$5.31 \pm 0.10^{\mathrm{b}}$	$5.09 \pm 0.06$ <sup>c</sup>	$4.87 \pm 0.01$ d	$4.43 \pm 0.03^{\text{e}}$	$4.25 \pm 0.03$ f	
Dietary fiber (%)	$9.47 \pm 0.01$ a	$9.9 \pm 0.02^{b}$	$10.9 \pm 0.05$ °	$11.63 \pm 0.34$ <sup>d</sup>	12 <mark>.</mark> 73 ± 0.26 º	$13.86 \pm 0.73$ f	
Water absorption index	$1.69 \pm 0.03$ a	$1.14 \pm 0.02^{\mathrm{b}}$	$1.06 \pm 0.03$ °	$0.96 \pm 0.03$ <sup>d</sup>	$1.09 \pm 0.03^{b,c}$	$1.12 \pm 0.02^{b,c}$	
Fracturability	$8.48 \pm 0.09$ a	$5.35 \pm 0.85^{b}$	$3.34 \pm 0.34$ °	$2.27 \pm 0.04$ <sup>d</sup>	$3.17 \pm 0.09^{\mathrm{e}}$	$4.64 \pm 0.12$ f	
Crispness	$3.9 \pm 0,03^{a}$	$2.4\pm0.02$ b	$1.5 \pm 0.02$ °	$1.9 \pm 0.03$ d	$3.21 \pm 0.05^{\text{e}}$	$3.7 \pm 0.03$ f	
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Table 2. The physicochemical properties of flakes produced from different ratios of Orange Sweet Potato (OSP), Red Rice (RR).

Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–f) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each row.

The water absorption index of flakes was lowest at an OSP to RR ratio of 40:60, with 0.96  $\pm$  0.03%, and was generally higher at combination ratios of 100:0, 80:20, 0:100, and 20:80. In addition, the texture characteristic of flakes was determined by the fracturability and crispness. The highest fracturability value was found in flakes made from 100% OSP (8.48  $\pm$  0.09). The reduction of the OSP proportion in flakes caused a decrease in fracturability until the proportion of 40:60 (2.27  $\pm$  0.04), beyond which the fracturability of flakes increased. A similar trend was observed in the crispness value of flakes. The lowest crispness value was detected in flakes with an OSP to RR ratio of 60:40, while higher values were obtained at ratios of 100:0, 0:100, 20:80, and 80:20.

Furthermore, the color of flakes was affected by the color of OSP and RR. The color profile of the flakes is shown in Table 3.

Table 3. Color profiles of flakes produced from different ratios of orange sweet potato (OSP), red rice (RR).

	Proportions of OSP and RR in Flakes								
_	100:0	80:20	60:40	40:60	20:80	0:100			
L <u>*</u>	$44.0\pm0.1$	$47.3 \pm 0.2$	$51.5 \pm 0.1$	$52.7 \pm 0.4$	$51.8 \pm 0.2$	$51.8 \pm 0.7$			
L <u>*</u> a <mark>*</mark>	$8.2 \pm 0.3$	$8,5 \pm 0.3$	$8.9 \pm 0.4$	$9.4 \pm 0.4$	$10.2 \pm 0.6$	$10.8 \pm 0.4$			
b*	$16.5 \pm 0.3$	$14.7 \pm 0.2$	$13.4 \pm 0.2$	$10.3 \pm 0.2$	$9.0 \pm 0.4$	$5.9 \pm 0.4$			
٥h	63.574	59.9622	56.4087	47.6158	41.4237	28.6476			
С	18.47	16.9685	16.0703	13.898	13.56	12.3145			

L\*: Lightness; a\*: redness; b\*: yellowness; oh: ohue; C: Chroma

#### 3.4. Sensory Characteristics of Flakes

A preference test was conducted to determine the sensory characteristics of the flakes. The results are presented in Table 4. The highest level of color preference was found in flakes with OSP to RR ratios of 60:40, 40:60, and 20:80, while flakes containing 100% OSP had the lowest level of acceptance for color. The preference scores for taste and crispness of the flakes with various proportions of OSP and RR were comparable. Flakes containing 100% OSP and 100% RR received the highest preference score for mouthfeel, with the former having a significant edge. Therefore, mixing the OSP and RR lowers the mouthfeel acceptance of the flake products.

Table 4. Preference test of flakes produced from different ratios of orange sweet potato (OSP), red rice (RR).

	Proportions of OSP and RR in Flakes								
	100:0	80:20	60:40	40:60	20:80	0:100			
Color	$3.35 \pm 1.42$ a	$4.40 \pm 1.22$ <sup>b</sup>	$5.14\pm1.12$ $^{\rm c}$	4.93 ± 1.21 °	4.89 ± 1.30 °	$4.30 \pm 1.12$ <sup>b</sup>			
Taste	$4.43 \pm 1.34$ a	4.69 ± 1.28 <sup>ab</sup>	$5.08\pm1.18$ $^{\rm c}$	5.09 ± 1.20 °	$5.05 \pm 1.26$ bc	$4.72 \pm 1.29$ abc			

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 Crispness
  $4.76 \pm 1.16^{\text{b}} 4.85 \pm 1.04^{\text{b}} 4.76 \pm 1.40^{\text{b}} 4.08 \pm 1.17^{\text{a}} 4.96 \pm 1.07^{\text{b}}$   $5.00 \pm 1.04^{\text{b}}$  

 Mouthfeel
  $5.41 \pm 0.91^{\text{d}} 5.04 \pm 1.18^{\text{c}} 4.84 \pm 1.31^{\text{bc}} 3.91 \pm 1.59^{\text{a}} 4.66 \pm 1.32^{\text{b}}$   $5.05 \pm 1.03^{\text{c}}$  

 Data are presented as mean  $\pm$  standard deviation. Different superscript letters (a–d) denote significantly different values according to Duncan's test (p < 0.05). Comparison was made within each row.

#### 4. Discussion

Phenolic compounds are the most widely found bioactive compounds in plants [33]. Some are produced in response to stress conditions as a defense mechanism of the plant. They have been extensively investigated due to their antioxidant activity and anti-degenerative disease effects [34]. In this research, methanol–HCl (1%) was used because acidic methanol can penetrate deeply into cells, disrupting the cell membrane. In addition, acidic methanol can dissolve and stabilize polar compounds such as phenolics and anthocyanins [35]. This research shows that RR and OSP are rich sources of phenolic compounds. It has previously been reported that RR has a high content of phenolic compounds [36], and that these compounds are primarily accumulated in the aleurone layer and bran of RR [37]. Thus, the rice milling process to remove the husk plays a vital role in preventing the loss of various beneficial compounds. Previously, it was suggested that ferulic acid, p-coumaric acid, and pro-catechuic acid are the most abundantly found phenolic compounds in RR [37].

Here, cooking led to a 49% decrease in the phenolic compounds of RR. Heating of RR generally destroys the structure of phenolic compounds by breaking the esterified and glycosylated bonds, thus decreasing the quantified content of phenolic compounds in cooked RR [38]. Our results agree with previous findings [39,40], which reported high levels of total phenolic compounds, mostly gallic acid, chlorogenic acid, pro-catechuic 4-hydroxybenzoic acid, and salicylic acid, in different varieties of OSP. However, phenolic contents were broken down by the heating process. Regarding the flake products, flakes containing a higher proportion of RR showed a higher content of phenolic compounds. Nevertheless, the processing lowered the phenolic compounds in flakes when compared to raw OSP and RR. Boiling and baking have previously been reported to be responsible for the loss of phenolic compounds of RR-based products [41]

A similar trend was observed in the anthocyanin content of RR. The cooking process resulted in a significant decrease in the anthocyanin content due to the unstable property of anthocyanins when exposed to high temperatures [42]. Anthocyanins were only detected in RR in this research. Anthocyanins such as cyanidin 3 glucoside, delphinidin 3 glucoside, and peonidin are reported to have health-promoting properties [43]. Thus, exposure to high temperatures for extended periods should be avoided to reduce the risk of anthocyanin breakdown. The anthocyanin content in flakes was lower than in the raw samples, and increasing the proportion of RR resulted in a higher anthocyanin content of flakes. The decrease in the anthocyanin content of flakes is possibly due to the heat treatment during flake production, typically involving two high temperature processing steps of pre-gelatinization and flaking. It has been reported that the high-temperature treatment used for food processing can lead to a reduction of the anthocyanin content [44].

Moreover, there is a growing research interest in the conversion of  $\beta$ -carotene and  $\alpha$ carotene absorbed in the duodenum to retinol by intestinal enzymes [45]. The high rate of vitamin A deficiency in the world and the detrimental effects caused by the condition have necessitated the search for foods that can supply sufficient amounts of daily vitamin A requirement. In this research, OSP had the highest content of  $\beta$ -carotene. However, boiling of OSP decreased the  $\beta$ -carotene by approximately 41% of the  $\beta$ -carotene available. This phenomenon could be due to the thermal breakdown of  $\beta$ -carotene. Moreover, carotenoids are well known as substances that are sensitive to light and high temperatures [46]. RR contains different types of carotenoids [47]. Here, an increase in carotenoids after boiling was found, which could be related to the thermal disruption of the protein–carotenoid complex and the consequent release of carotenoids from the matrix. Similar findings have been published [48]. Similar trends were also observed in the OSP- and RR-based flake products. The  $\beta$ -carotene was significantly lower in the cooked sample compared to the raw sample. The OSP proportion in the flake formulation affected the carotenoid content. The higher the OSP proportion, the greater the carotenoid content of the flakes. In addition, the carotenoid contents of the flakes were significantly lower than the raw material. The simultaneous heating process from pre-gelatinization to flake pressing could further break down the carotenoids. This finding is supported by previously published work, which shows that heat treatment during food processing is responsible for the loss of carotenoids to degradation [49].

Vitamin E deficiency could lead to severe neurological problems. The main vitamin E compounds are tocopherols, with both  $\alpha$ - and  $\gamma$ -tocopherol providing most vitamin E activity. Both samples had high contents of  $\alpha$ - and  $\gamma$ -tocopherol. A significant increase (74%) in  $\alpha$ -tocopherol was found in OSP after boiling. This result indicates that heat treatment could be beneficial for the bioaccessibility of tocopherol. Furthermore, heat treatment can assist in the breakdown of complex foods. Thus, tocopherol can be quickly released from its binding site. On the contrary, heat treatment was reported to reduce the tocopherol content in corn [50]. The increase in tocopherol after boiling indicates that tocopherols in the sample are more stable to heat treatment than other foods. In this research, RR had a considerably high tocopherol content. Therefore, boiling could have released the tocopherols from their binding site, facilitating their extraction and detection. Moreover, the structure of the rice grains could have played a role. RR has a compact structure. Therefore, cooking could assist the extraction of tocopherols, increasing the extractable tocopherol content, although some of the tocopherols might be lost to high temperature. The tocopherol contents of the different flake products were lower than the raw and boiled OSP and RR. It could be due to the simultaneous or prolonged exposure to heat treatment. Unlike the conventional method of cooking, which increases the tocopherol content, extended heating breaks the matrix structure in flakes and significantly affects the tocopherols. Thus, prolonged heating should be avoided in the processing of healthy food products to retain their tocopherols. Heat treatment combined with mechanical treatment in specific conditions could release the tocopherols from the food matrix. However, extended exposure will lead to the breakdown of carotenoids in the sample.

This research measured the antioxidant activity of raw and cooked OSP and RR and their flake products using DPPH, FRAP, and Superoxide radical scavenging capacity assays. The result showed that heat treatment was responsible for decreasing the antioxidant activity of OSP, RR, and the flake products. A positive correlation was observed between the reduction of bioactive compounds and antioxidant capacity. Most bioactive compounds are heat sensitive, which influences their antioxidant activity. A previous study reported that an increase in temperature accelerated the initiation of oxidation, preventing antioxidant compounds from working optimally [51]. The bioactive compounds were degraded, experiencing structural changes, and wholly transformed into inactive substances. Nevertheless, due to the complex nature of antioxidant compounds in plants, their thermal stability varies. Some compounds such as pro-catechuic acid, p-coumaric acid, and ferulic acid have high thermal stability, which facilitates their extraction and antioxidant activity [52]. The heat treatment process assists in the release of such compounds without affecting their activity. Moreover, the flake products exhibited lower antioxidant activity than the raw or boiled products. Based on the percentage values of DPPH and Superoxide radical scavenging capacity and the content of Fe [II] formed, flakes containing only OSP or RR had lower antioxidant activity. Interestingly, the combination of OSP and RR increased the antioxidant activity, probably due to the synergistic effect of bioactive compounds from OSP and RR [53]. Even though the processing reduced the antioxidant activity in boiled samples and flake products, the remaining antioxidant activity was still considerably high. Therefore, boiled OSP, RR, and their flake products are a good source of bioactive compounds and antioxidants.

Regarding the physicochemical and sensory properties of the flakes, it was observed that the moisture content of flake products was mainly influenced by their composition. Starch is the dominant carbohydrate found in OSP and RR, and OSP has a lower amylose content compared to RR. According to Wang et al. [54], the amylose content of OSP is 18.71%, while Markus et al. [55] reported that RR has 23% amylose content. Amylose is a linear polymer of glucose, which forms starch. The higher the amylose content, the greater the moisture content of the dough due to a higher capability to absorb water. The absorbed water will promote dough gelatinization during heating. The water absorbed by the dough will then evaporate during the flaking process due to the network's inability to entrap water from the matrix structure of the flakes, resulting in increased evaporation and a lower moisture content of the flake products.

The dietary fiber content of the flakes ranged between 9.47 and 13.86%, comparable to values commonly found in breakfast cereals such as corn flakes, rice, quinoa, millet, and amaranth flakes [56]. The proportion of RR affected the dietary fiber of flakes. The fiber content of flakes was also influenced by the heating and pressing processes. Heat treatment caused the degradation of the fiber matrix and the glycosidic bond. The degradation affected the solubility level of the fiber, i.e., the ratio between soluble and insoluble fiber, thus resulting in the reduction of total fiber in the product.

Water absorption index (WAI) is a physical property associated with the ability of flakes to absorb water molecules within a particular time. Absorbed water molecules could be bound or detained in matrix pores of flakes. The water absorption index is crucial because it is associated with the quality of the flakes. Consumers can experience the crispy and crunchy sensation of the flakes after soaking in milk or water. In contrast, the higher WAI is interrelated with the unwanted soggy texture of flakes. The WAI is influenced by the porosity of the matrix on flakes, thickness, and hygroscopicity of flakes. In addition, the presence of fiber and protein could assist the flakes in absorbing water into their structure. The result shows that increasing the proportion of RR reduces the WAI due to a decrease in hygroscopicity. OSP is a rich source of sugar; thus, it has higher hygroscopicity compared to RR. In addition, the presence of RR affects the network construction of flakes by inhibiting the formation of the starch-protein structure, which could entrap gas. The compact structure created by RR starch inhibits water absorption into the matrix of the product. On the other hand, RR has higher fiber and protein contents, which help improve water absorption [57]. Moreover, the suitable fragmentation of the amylose and amylopectin chain in sweet potato could also affect the water absorption capacity. The heating processes such as roasting, flaking, and extrusion will induce starch fragmentation with sufficient water. The gelatinization process converts starch to a digestible material and plays a vital role in determining the structural properties of flakes and their ability to absorb moisture. The presence of RR in the flake dough disturbs the composition of starch, thus inhibiting the flakes from forming a porous structure and affecting their water absorption capacity.

Fracturability is a physical property related to deformation conditions when a specific maximum force is applied. A higher fracturability value represents the ability of food products to maintain their structure when force is applied. According to Table 2, flakes with 100% OSP have a higher fracturability value compared to others, as the homogenous matrix of starch, protein, and fiber in OSP enables the interaction between the matrix and water molecules, leading to the firm, sturdy and rigid texture of flakes. The rigid texture is related to the evolution phase of starch from amorphous conditions, which indicates complete disorganization of the crystalline structure of starch. Increasing the proportion of RR in the flakes the fracturability value. The mixture of OSP and RR decreases the rigidity of the flakes due to the different structural properties of the two samples, creating flakes that are susceptible to fracture. Crispness is a complex texture attribute because it comprises a combination of sensory analysis, acoustical procedure, and instrumental analysis. The instrumental analysis revealed that flakes with OSP to RR ratios of 100:0 and 0:100 had higher crispness values than others. Similar to fracturability, the

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crispness value decreased with an increasing RR proportion in the flakes. The mixture of ingredients with different structural properties can affect the crispness value of flakes.

The color profile shown in Table 3 revealed that the color of flakes was affected by the pigments in the raw materials used for their production. The red color of RR is associated with anthocyanins found in its bran layer. The yellowness of OSP is linked to carotenoids, primarily  $\beta$ -carotene. Previous research reported that carotenoids are easily oxidized and undergo color degradation due to thermal treatment [58]. The hue results showed a flake color range between yellow and red. The color of flakes was also affected by the Maillard reaction product. The higher the Maillard reaction product, the darker the appearance of flakes.

The sensory analysis involved a preference test of the color, taste, crispness, and mouthfeel. Flakes containing 100% OSP had the lowest color preference score, associated with the orange appearance, and the lowest brightness score. Most panelists perceived the dark orange color as less fresh, less attractive, and less tasty. The color preference was increased with the addition of RR, which also helped improve the product lightness and redness values. The panelists were mostly in favor of flakes that appeared brighter and reddish. The higher brightness level is attributed to the white endosperm color of RR, while the redness is related to the anthocyanins in the bran of RR. The mouthfeel preference of flakes is associated with the water absorption index. The higher the ability of flakes to absorb water, the greater the plasticizing effect due to the presence of more hydrophilic components such as the phosphate monoester found in sweet potato starch [59]. The panelists generally preferred flakes with soft mouthfeel. The taste and crispness preferences for OSP- and RR-based flakes were in the range of "indifferent" and "slightly likes". Increasing the RR proportion in flakes decreases the bitterness intensity and increases the savory taste. The perception of savory taste is generally influenced by the moisture content and the flavor of RR. Niu et al. [60] suggests that sweet potato with a low dry matter content has a bitter taste, and increasing the dry matter reduces the bitterness. On the other hand, a decrease in invertase activity may support the bitter aftertaste of the sweet potato.

This research successfully monitored the changes of the bioactive compound and antioxidant activity of OSP and RR in their native form and in the flake products. It can be observed that individual compounds acted differently to processing methods. Therefore, it can be suggested that research on the development of functional foods should address the products ready to be consumed instead of solely focusing on the raw materials due to the changes that take place during the transformation. This approach should be implemented for other potential materials rich in bioactive compounds. Moreover, further consideration of the bioaccessibility and bioavailability that are affected by digestion and absorption in the human metabolism system should also be considered [61].

#### 5. Conclusions

OSP and RR are rich sources of bioactive compounds, especially  $\beta$ -carotene, for OSP, and phenolic compounds and anthocyanins, for RR. The boiling process significantly decreased most of the bioactive compounds, except tocopherols and  $\alpha$ -carotene. The level of bioactive compounds in the flake products was dependent on the proportion of OSP and RR. Heat treatment resulted in a decrease in antioxidant activity, even though the remaining activity was still considerably high. The mixture of OSP and RR can produce flakes with low moisture and high fiber contents. The optimum flake water absorption index, fracturability, and crispness were obtained by combining 40% OSP and 60% RR. Moreover, the ratio of OSP and RR influenced the color and sensory preferences of the panelists.

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supervision, C.W.; project administration, I.R.A.P.J. and S.R.; funding acquisition, I.R.A.P.J., S.R., and T.I.P.S. All authors have read and agreed to the published version of the manuscript.

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