

ROLE OF NATURAL POLYSACCHARIDES XANTHAN GUM- LOCUST BEAN GUM IN PER-ORAL DRUG DELIVERY SYSTEMS

by Lannie Hadisoewignyo

FILE	22PI-ROLE_OF_NATURAL_POLYSACCHARIDES.PDF (1.22M)		
TIME SUBMITTED	11-JAN-2021 11:35AM (UTC+0700)	WORD COUNT	2625
SUBMISSION ID	1485546607	CHARACTER COUNT	13738

ROLE OF NATURAL POLYSACCHARIDES XANTHAN GUM- LOCUST BEAN GUM IN PER-ORAL DRUG DELIVERY SYSTEMS

Lannie Hadisoewignyo¹⁾, Achmad Fudholi²⁾

¹⁾Faculty of Pharmacy, Widya Mandala Catholic University
Jl. Dinoyo 42-44, Surabaya, Phone(+62)031-5678478

²⁾Faculty of Pharmacy, Gadjah Mada University

Corresponding author: e-mail: lanhadi@yahoo.com

ABSTRACT

Xanthan gum is a heteropolysaccharide derived from corn starch and produced by fermentation with the bacteria *Xanthomonas campestris*. Locust bean gum is a natural hydrocolloid obtained from the seed endosperm of the carob tree (*Ceratonia siliqua* L.). The structure of locust bean gum consists of a β -(1 \rightarrow 4)-D-mannosa backbone to which D-galactose units are attached α -(1 \rightarrow 6). Locust bean gum is a non-ionic polysaccharide, which can form gel if it is combined with xanthan gum. Locust bean gum is capable of synergistic interactions with xanthan gum.

This combination can be used for a matrix of a sustained release dosage form based on the ability of the combination of xanthan gum and locust bean gum to form gel. In this research, ibuprofen was used as an active ingredient formulated in control release tablet dosage form.

The results exhibited that the combination of xanthan gum and locust bean gum at a ratio of 1:1 gave the longest in vitro release. Kinetic drug release tend to zero order with dominant mechanism is erosion.

Keywords: xanthan gum, locust bean gum, matrix, sustained release.

INTRODUCTION

Xanthan gum is a heteropolisakarida which has a large molecular weight and produced from the fermentation of carbohydrates by the microorganism *Xanthomonas campestris*. It has a double helical structure of antiparallel bimolekular. Glucan backbone structures protected by the side chains are along the sides, which makes xanthan gum is relatively stable on heating as well as in acid or alkaline conditions. Xanthan gum is easily hydrated in

systems containing water as it can dissolve in cold or hot water (Anonymous, 2005^b).

8 Locust bean gum is 8 often also called carob bean gum or a galaktomanan carubin is obtained from the seeds of *Ceratonia siliqua* extraction. Locust bean gum is a galaktomanan, containing chains (1→4) - are associated with β-D-mannopiranosas as the main chain, which in the sixth position associated with α-D-galactose. Locust bean gum is a non-ionic compound so that the ionic strength and pH has no effect on the viscosity, except at extreme pH and high temperature. Locust bean gum partially hydrated in cold water and perfect hydrated when heated (Anonymous, 2005^a).

Xanthan gum when combined with locust bean gum to form a gel when liquid is in an environment (Baichwal, 2001). This behavior is called synergism. Medicine that is often used so frequently in a day that is can cause the patient to forget to take it and less effectiveness in the treatment. To overcome these events experts designing and developing pharmaceutical dosage forms which can release the drug gradually so that the availability of drugs in the blood can be maintained longer and the drug action can be extended. This dosage form is known as a sustained release dosage form (Ansel, 1989).

Ibuprofen [(±)-2 - (p-isobutylfenil) propionic acid) is a nonsteroidal anti-inflammatory (NSAID) that is often used. Ibuprofen has a short biological half-life of approximately two hours that causes need to be used repeatedly in a day. In the form of tablets are generally used in doses of 200 mg to 800 mg, three to four times daily or 600 mg to 3200 mg per day. This is why ibuprofen is formulated in a dosage suitable for slow release (Kurumaddali et al., 1994, Tehrani and shobeiri, 1995; Shehab and Richards, 1996).

6 Pembuatan suatu matriks dengan bahan obat berada di dalamnya atau tercampur homogen dengan bahan matriks merupakan salah satu cara yang dapat digunakan untuk membuat sediaan lepas lambat. Bahan pembentuk matriks dapat berupa bahan yang tidak larut dalam air maupun bahan yang larut dalam air dan membentuk gel. Salah satu polimer yang sekarang mulai banyak digunakan untuk matriks sediaan lepas lambat adalah kombinasi dari *xanthan gum* dan *locust bean gum*.

One way to make sustained released preparation use a matrix system. The matrix can be developing 12 with water soluble material and form a gel when liquid is in environment. Combination of *xanthan gum* and *locust bean gum* can be consider to be use as matrix sustained release tablet (Anonim, 2005).

According to Higuchi (1963), the release of drug dispersed in a homogeneous solid and inert matrix can be describe following equations (1):

$$Q = [D(2A - C_s)C_s.t]^{1/2} \dots\dots\dots (1)$$

The rate of drug release at once (shortly) at time t obtained by differentiate equation (1), and obtained:

$$\frac{dQ}{dt} = \frac{I}{2} \left[\frac{D(2A - C_s)C_s}{t} \right]^{1/2} \dots\dots\dots (2)$$

In general, $A \gg C_s$, so that equation (1) becomes

$$Q = (2ADC_s t)^{1/2} \dots\dots\dots (3)$$

and equation (2) becomes

$$\frac{dQ}{dt} = \left[\frac{ADC_s}{2t} \right]^{1/2} \dots\dots\dots (4)$$

with Q is the amount of drug released after time t per unit area, D is the diffusion coefficient of drug in the matrix, A is the total amount of drug in the matrix per unit volume, and C_s is the solubility of drug in the matrix.

Type of matrix diffusion and development in general do not follow the Fickian diffusion mechanism. The simple equation that can describe the behavior of drug release from hydrophilic matrix systems are:

$$\frac{Mt}{M_\infty} = kt^n \dots\dots\dots (5)$$

Mt / M_∞ is the fraction of drug release in time t , k is a constant and n is the exponential diffusion which indicatives of the mechanism of drug release (Wise, 2000).

METHODS

Materials

Ibuprofen (USP., BP., Hubei Biocause Helan Pharmaceutical CO., LTD.), Xanthan gum (degussa Texturant Systems, France), Locust bean gum (degussa Texturant Systems, France), Avicel® PH 101 (Asahi Kasei Chemical Corp., Tokyo, Japan), Polivinilpirolidon K-30 (ISP Technologies Inc., Wayne, New Jersey), Magnesium stearat (Peter Greven CV, Nederland), talk (Haichen, China), Natrium hidroksida (pa, Merck), Kalium dihidrogen fosfat (pa, Merck).

Equipments

Single punch tableting machine (Type TDP, Shanghai, China), analytical balance (Sartorius), hardness tester (Schleuniger Hardness Tester, tipe 6D-30, Germany), friablation tester (Erweka, type TA-3, Germany), dissolution tester (Erweka, type DT70, Germany), fluid bed dryer (Glatt, Germany), spectrophotometer UV-VIS (Hitachi, type U-1100, Japan).

Tablet Manufacturing Method

Ibuprofen tablets made from a variety of formulas. The difference formula based on different combinations of the matrix. Combination matrix used is xanthan gum and locust bean gum. The concentration used for each combination were: xanthan gum, locust bean gum = 1: ½: 1:1, 1:1 ½. Composition formula ibuprofen tablets can be seen in Table I.

Table I. Composition (mg per tablet) of Sustained Release Tablet Ibuprofen (600 mg)

Ingredients	Amount (mg/tablet)		
	Formula I	Formula II	Formula III
Ibuprofen	400	400	400
Xanthan Gum	12.5	12.5	12.5
Locust Bean Gum	6.25	12.5	18.75
Avicel® PH 101	91.25	85	78.75
PVP K-30	30	30	30
Mg Stearat	6	6	6
Talk	54	54	54
Weight of one tablet	600	600	600

Formula I : Xanthan gum : Locust bean gum = 1 : ½

Formula II : Xanthan gum : Locust bean gum = 1 : 1

Formula III: Xanthan gum : Locust bean gum = 1 : 1½

The combination of matrix and Avicel® PH 101 is mixed until homogeneous and added water to form granules, then sifted with a sieve mesh 16. The granules are dried using a fluid bed dryer at 55 ° C for 30 minutes. Sifted dry granules by sieve mesh 18. Dried granules are mixed with ibuprofen and added to a solution of polyvinylpyrrolidone to form the mass of granules, and sieved with a sieve mesh 16. The granules are dried by fluid bed dryer at a temperature of 55 ° C for 30 minutes, then sieved with a sieve mesh 20. Added magnesium stearate and talk, then mixed and homogeneous granules were observed physical properties. Then 600 mg of the prepared granules was compressed using single punch tablet machine with the same hardness was 17-18 Kp.

Characterization of Granules and Tablets

The flow characteristics, bulk and tapped density of granules were measured to verify the improvement in physical characteristics of granules for tableting. Physical properties of tablets include tablet weight uniformity, tablet hardness, tablet friability and the disintegration time of tablet, other than that conducted the assay of ibuprofen tablets.

Drug Release Testing

The dissolution test was carried out using a dissolution apparatus II method USP ⁵ over a 6-h period. The dissolution media was phosphate buffer pH 7.2. The temperature of medium and the rotation speed of the paddle were maintained at 37 °C (±0.5 °C) and 100 r/min, respectively. Six tablets were used for each test. Samples (at 5 mL) were withdrawn and replaced with 5 mL fresh phosphate buffer according to following time scheme: 30, 60, 90, 120, 180, 240, 300, and 360 min. The samples were filtered using 0.45 µm Milipore filter. The filtered sample was measured spectrophotometrically at 265 nm to determine the amount of released drug.

RESULTS AND DISCUSSION

Table II shows that bulk density has a smaller value than the value of tapped density and has a significant difference between bulk density and tapped density. This indicates that these granules have large void between particles. From the data of Carr's index indicated that all formulas have a good flow properties because of Carr's index gained less than 20% and Hausner ratio values less than 1.25. Thus a mixture of granular flow is expected to be free flowing and to give homogeneity of content of ibuprofen in tablets.

Table II. Physical Properties of Granules

Formula	Bulk density (g/mL)	Tapped density (g/mL)	Carr's Index (percent)	Hausner Ratio
I	0.34	0.38	12	1.12
II	0.35	0.39	11	1.11
III	0.32	0.36	12	1.13

Percent friability of all formulas to meet the requirements of less than 0.8% showing resistance to the effects of tablet manufacturing process. Physical strength of the exterior of the tablet in the fight against mechanical shock is what determines the size of the fragility.

Table III. Physical Properties of Tablet

Formula	Tablet Hardness (Kp)	Tablet Friability (percent)	Disintegration Time (min)	Content Uniformity (percent)	cv of Content Uniformity (percent)
I	17.7	0.1	117	94.8	0.41
II	18.1	0.1	170	85.9	2.87
III	17.5	0.2	150	96.0	0.78

The disintegration time of all formulas using a matrix formula in which a combination of xanthan gum and locust bean gum is affected by formulation. With the same relative hardness tablet disintegration time obtained varies. The longest disintegration time of the formula II, this is probably due to the ratio of xanthan gum and locust bean gum = 1:1 causing the internal structure of the tablet will have a smaller porosity which causes the water will be difficult to penetrate and increase the disintegration time of tablets.

Table III shows that the content of ibuprofen in tablets of each formula has a cv of less than 5%, this means that content of ibuprofen in each tablet in each formula is uniform. This content uniformity data illustrate that tableting process goes well and produces tablets with active substances which approached the ideal levels.

Dissolution profiles was obtained by plotting the percent drug released *vs.* time as shown in Figs.1. The profile of drug released showed that the formula III has the percent of drug release of the most high despite the destruction of much longer time when compared with formula I. This is because formula III shows the mikrogranular while the formula I are makrogranular. While the profile of drug release on the formula II shows the release of the longest when compared with other formulas.

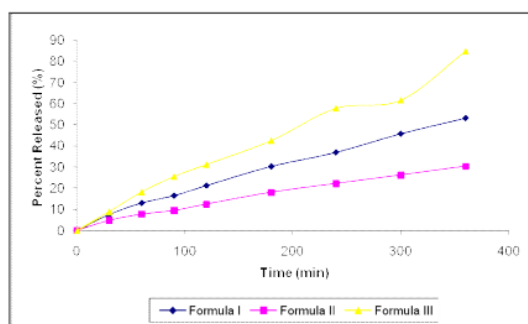


Figure 1. Release profile of ibuprofen sustained release tablet

Table IV. Mathematical Modeling and Drug Release Kinetics of Ibuprofen Sustained Release Tablet

Formula	Zero Order		Higuchi	First Order	Exponential Diffusion (n)
	k (mg/min)	r	r	r	
I	0.5190	0.9991	0.9922	0.9975	0.873
II	0.2679	0.9980	0.9933	0.9993	0.603
III	0.8130	0.9919	0.9830	0.9447	0.570

Table IV shows that both diffusion and erosion mechanisms applicable to the release of ibuprofen from matrix xanthan gum-locust bean gum in medium phosphate buffer pH 7.2, with the dominant mechanism is erosion. Characteristics of drug release, represented by the value of n . All formulas have the value of n greater than 0.45 and smaller than 0.89 means that the release characteristics of ibuprofen following the anomalous diffusion transport mechanism, where the rate of diffusion and polymer relaxation are comparable. N value above also indicates that the controlled drug release more than one process.

CONCLUSION

Xanthan gum in combination with locust bean gum can be used as a slow release matrix tablet of ibuprofen with the kinetics of drug release tends to zero order and the dominant mechanism is erosion. The mechanism of drug transport that occurs is anomalous diffusion, which states that the controlled drug release more than one process. Levels of locust bean gum is used combination matrix may affect the rate of dissolution, where the ratio of 1:1 between Xanthan gum and locust bean gum provides drug release profile of the longest with the lowest dissolution rate compared to other formulas.

REFERENCES

- Anonim. 2005a. *Water Structure and Behavior : Locust Bean Gum*. Available at: <http://sbu.ac.uk/water/hyloc.html>. Accessed 28 January 2003.
- Anonim. 2005b. *Water Structure and Behavior : Xanthan Gum*. Available at: <http://www.sbu.ac.uk/water/hyxan.html>. Accessed 28 January 2003.
- Ansel, H.C. 1989. *Introduction to Pharmaceutical Dosage Forms*. 4th Ed., Lea&Febiger, Philadelphia.
- Baichwal, A, R. 2001. *Sustained Release Heterodisperse Systems-amorphous* Drugs. US Patent 9, 879, 296.
- Higuchi, W.I. 1963. Mechanism of Sustained Action Medication. Theoretical Analysis of Rate of Release of Solid Drugs Disperse in Solid Matrices. *J. Pharm. Sci.*, 52, 1145-1149.
- Kurumaddali, K.R., Ravis, W.R., and Betageri, G.V. 1994. Preparation and Evaluation of Sustained Release Ibuprofen Beads. *Drug Development and Industrial Pharmacy*. 20(17): 2659-2669.
- Tehrani, M.R., and Shobeiri, N. S. 1995. Effect of Various Polymers on formulation of Controlled Release (CR) Ibuprofen Tablets by Fluid Bed Technique,. *Drug Development and Industrial Pharmacy*. 21 (10): 1193-1202.
- Shehab, M. A., and Richards, J. H. 1996. Studies on the In Vitro Release of Ibuprofen from Polyethylene Glycol-Polyvinyl Acetat Mixtures Liquid Filled into Hard Gelatin Capsules. *Drug Development and Industrial Pharmacy*. 22(7): 645-651.
- Wise, D. L.,2000. *Handbook of Pharmaceutical Controlled Release Technology*. Marcel Dekker, Inc., New York.

ISBN

ISBN 978-602-95911-1-8



**CENTER FOR PHARMACEUTICAL AND MEDICAL TECHNOLOGIES
DEPUTY FOR AGROINDUSTRY AND BIOTECHNOLOGY
AGENCY FOR THE ASSESSMENT AND APPLICATION OF TECHNOLOGY
BPPT Gedung II, Lantai 15
Jalan M. H. Thamrin No 8, JAKARTA 10340
Telp. / Fax : (021) 316 9505**

2010

ROLE OF NATURAL POLYSACCHARIDES XANTHAN GUM-LOCUST BEAN GUM IN PER-ORAL DRUG DELIVERY SYSTEMS

ORIGINALITY REPORT

%**20**

SIMILARITY INDEX

%**15**

INTERNET SOURCES

%**12**

PUBLICATIONS

%**4**

STUDENT PAPERS

PRIMARY SOURCES

1

vdocuments.site

Internet Source

%**6**

2

**Submitted to Higher Education Commission
Pakistan**

Student Paper

%**2**

3

**Sven Richter. "Gelation Studies, 3",
Macromolecular Rapid Communications,
04/06/2005**

Publication

%**2**

4

www.freepatentsonline.com

Internet Source

%**2**

5

**Jyh-Ding Wei. "Characterisation of fenofibrate
dissolution delivered by a self-microemulsifying
drug-delivery system : Fenofibrate delivered by
SMEDDS", Journal of Pharmacy and
Pharmacology, 12/2010**

Publication

%**1**

6

Internet Source

%1

7

M. L. Vueba, L. A. E. Batista de Carvalho, F. Veiga, J. J. Sousa, Maria Eugénia Pina. "Role of Cellulose Ether Polymers on Ibuprofen Release from Matrix Tablets", Drug Development and Industrial Pharmacy, 2008

Publication

%1

8

Sundaram Gunasekaran, Won Byong Yoon. "Investigation of Elastic Modulus of Xanthan and Locust Bean Gum at Different Concentrations of Mixture Using Cascade Model", Journal of Texture Studies, 2014

Publication

%1

9

repository.ubaya.ac.id

Internet Source

%1

10

jddtonline.info

Internet Source

%1

11

Sven Richter, Volodymyr Boyko, Rolf Matzker, Klaus Schröter. "Gelation Studies: Comparison of the Critical Exponents Obtained by Dynamic Light Scattering and Rheology, 2", Macromolecular Rapid Communications, 2004

Publication

%1

12

Vipul D. Prajapati, Girish K. Jani, Naresh G. Moradiya, Narayan P. Randeria, Bhanu J.

%1

Nagar. "Locust bean gum: A versatile biopolymer", Carbohydrate Polymers, 2013

Publication

13

Submitted to Wageningen University

Student Paper

<% 1

14

maruaryriskaterz-pharmacy.blogspot.com

Internet Source

<% 1

EXCLUDE QUOTES ON

EXCLUDE
BIBLIOGRAPHY ON

EXCLUDE MATCHES < 10
WORDS