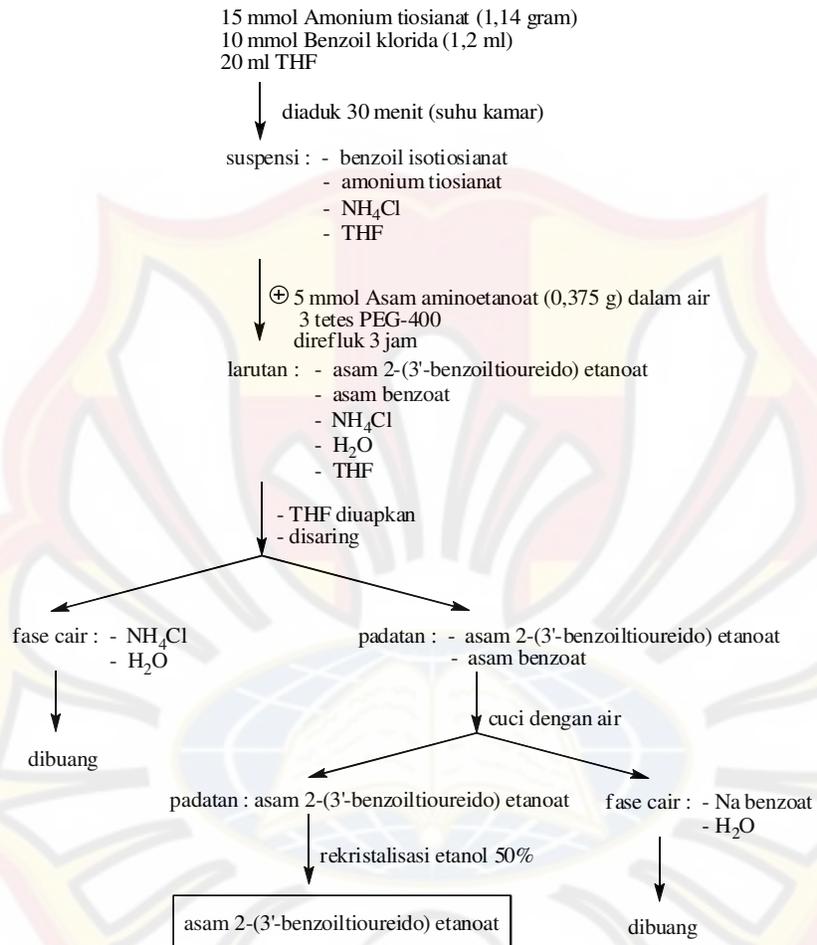


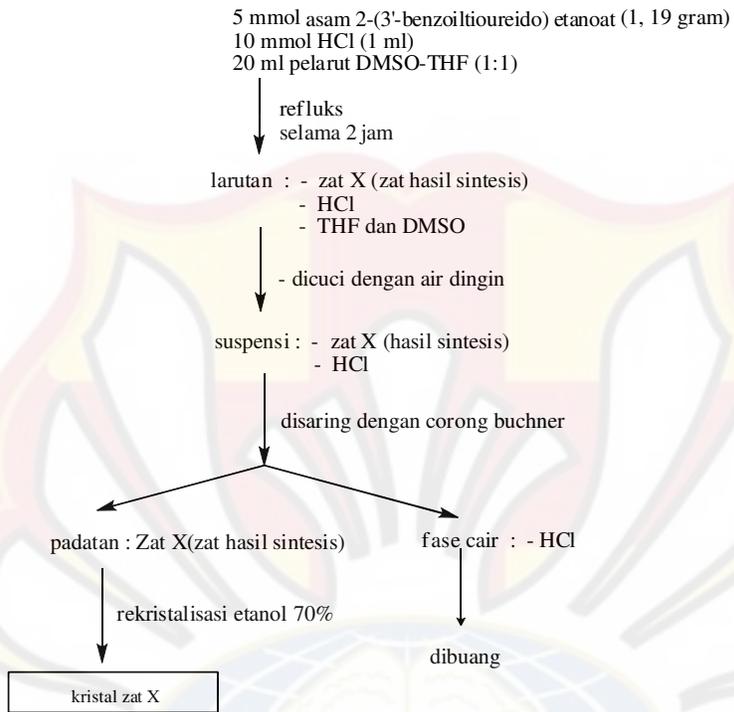
## LAMPIRAN A

### Bagan Alir Sintesis Asam 2-(3-Benzoiltioureido) Etanoat



## LAMPIRAN B

### Bagan Alir Siklisasi Asam 2-(3-Benzoyltioureido) Etanoat



## LAMPIRAN C

### Bagan Alir Sintesis Satu tahap

15 mmol amonium tiosianat (1,142 gram)  
10 mmol benzoil klorida (1,2 ml)  
20 ml THF

↓  
diaduk 30 menit (suhu kamar)

suspensi : - benzoil isotiosianat  
- amonium tiosianat  
-  $\text{NH}_4\text{Cl}$   
- THF

↓  
⊕ 5 mmol Asam aminoetanoat (0,375 g) dalam air  
3 tetes PEG-400  
10 mmol HCl (1 ml)  
refluk 2 jam

larutan : - zat X (zat hasil sintesis)  
- asam benzoat  
-  $\text{NH}_4\text{Cl}$   
-  $\text{NH}_4\text{OH}$   
-  $\text{H}_2\text{O}$   
- THF

↓  
- THF diuapkan  
- disaring

fase cair : -  $\text{NH}_4\text{Cl}$   
-  $\text{H}_2\text{O}$   
-  $\text{NH}_4\text{OH}$

↓  
dibuang

padatan : - zat X (zat hasil sintesis)  
- asam benzoat

↓  
cuci dengan larutan bikarbonat 10%

padatan : zat X (zat hasil sintesis)

↓  
rekristalisasi etanol 70%

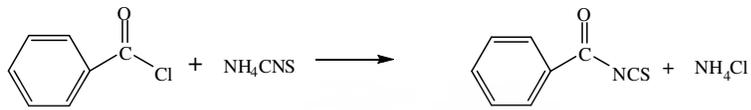
↓  
kristal zat X (zat hasil sintesis)

fase cair : - Na benzoat  
-  $\text{Na}_2\text{CO}_3$   
-  $\text{H}_2\text{O}$

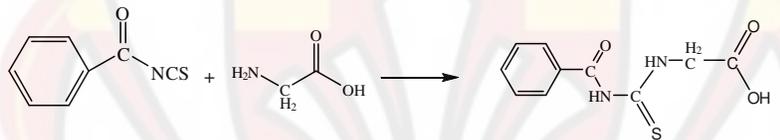
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dibuang

## LAMPIRAN D

### Perhitungan Hasil Sintesis Asam 2-(3'-Benzoiltioureido) Etanoat Secara Teoritis



Benzoil klorida	Amonium tiosianat	Benzoil isotiosianat
BM 140, 59 g/mol	BM 76, 12 g/mol	BM 163, 202 g/mol
10 mmol	15 mmol	10 mmol



Benzoil isotiosianat	Asam amino etanoat	Asam 2-(3'-benzoiltioureido) etanoat
BM 163, 202 g/mol	BM 75, 03 g/mol	BM 238,26 g/mol
10 mmol	5 mmol	5 mmol

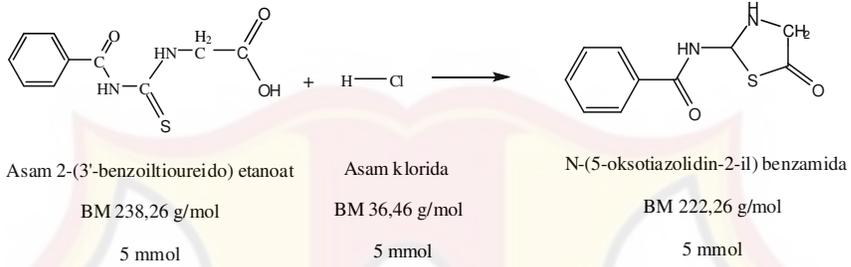
Berat molekul asam 2-(3'-benzoiltioureido) etanoat = 238,26 g/mol

Berat asam 2-(3'-benzoiltioureido) etanoat

5 mmol x 238,26 g/mol = 1,1913 g

## LAMPIRAN E.

### Perhitungan Hasil Siklisasi Asam 2-(3'-Benzoiltioureido) Etanoat Secara Teoritis



Berat molekul senyawa hasil siklisasi = 222,26 g/mol

Berat Senyawa Hasil Siklisasi

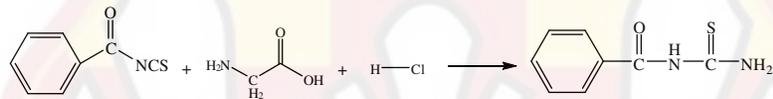
5 mmol x 222,26 g/mol = 1,1113 g

## LAMPIRAN F

### Perhitungan Hasil Sintesis Benzoiltiourea Secara Teoritis



Benzoil klorida	Amonium tiosianat	Benzoil isotiosianat
BM 140, 59 g/mol	BM 76, 12 g/mol	BM 163, 202 g/mol
10 mmol	15 mmol	10 mmol



Benzoil isotiosianat	Asam amino etanoat	Asam klorida	Benzoiltiourea
BM 163, 202 g/mol	BM 75, 03 g/mol	BM 36,46 g/mol	BM 180, 234 g/mol
10 mmol	5 mmol	10 mmol	5 mmol

Berat molekul Benzoiltiourea = 180, 234 g/mol

Berat Benzoiltiourea = 5 mmol x 180, 234 g/mol = 0,901 g

## LAMPIRAN G

### Contoh Perhitungan Rendemen Hasil Asam 2-(3'-Benzoiltioureido) Etanoat dan Senyawa Hasil Siklisasi

$$\text{Rendemen hasil} = \frac{\text{berat praktis}}{\text{berat teoritis}} \times 100 \%$$

#### a. Asam 2-(3'-Benzoiltioureido) Etanoat

$$\begin{aligned} \text{Sintesis I} & : \text{ Berat praktis} & = & 0,980 \text{ gram} \\ & & \text{ Berat teoritis} & = 1,1913 \text{ gram} \\ & & \text{ Persentase hasil} & \\ & & = \frac{0,980}{1,1913} \times 100 \% & = 82,26\% \end{aligned}$$

$$\text{Sintesis I} : \text{ Persentase hasil} = 81,84 \% \text{ (0,975 gram)}$$

$$\text{Sintesis III} : \text{ Persentase hasil} = 81,93\% \text{ (0,976 gram)}$$

Persentase hasil rata-rata

$$\begin{aligned} & : \frac{82,26\% + 81,84\% + 81,93\%}{3} = 82,01\% \end{aligned}$$

**b. Senyawa Hasil Siklisasi**

$$\begin{aligned} \text{Sintesis I} & : \text{ Berat praktis} & = & 0,610 \text{ gram} \\ & & \text{ Berat teoritis} & = 1,1113 \text{ gram} \end{aligned}$$

Persentase hasil

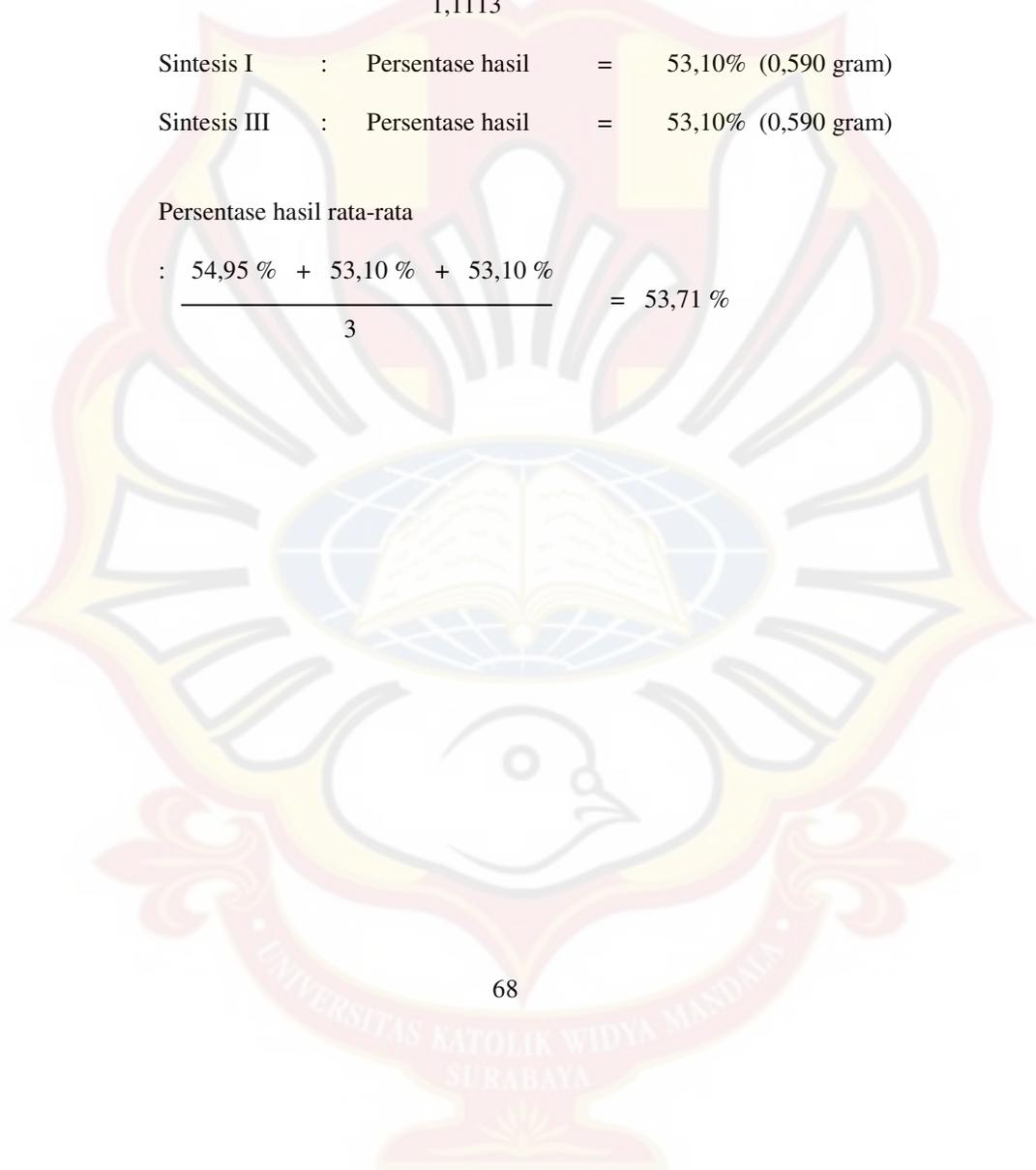
$$= \frac{0,610}{1,1113} \times 100 \% = 54,95 \%$$

$$\text{Sintesis I} : \text{ Persentase hasil} = 53,10\% (0,590 \text{ gram})$$

$$\text{Sintesis III} : \text{ Persentase hasil} = 53,10\% (0,590 \text{ gram})$$

Persentase hasil rata-rata

$$: \frac{54,95 \% + 53,10 \% + 53,10 \%}{3} = 53,71 \%$$





## LAMPIRAN H

### Contoh Perhitungan Rendemen Hasil Benzoiltiourea

$$\text{Rendemen hasil} = \frac{\text{berat praktis}}{\text{berat teoritis}} \times 100 \%$$

#### Benzoiltiourea

$$\text{Sintesis I} : \text{ Berat praktis} = 0,465 \text{ gram}$$

$$\text{Berat teoritis} = 0,901 \text{ gram}$$

Persentase hasil

$$= \frac{0,465}{0,901} \times 100 \% = 51,61\%$$

$$\text{Sintesis II} : \text{ Persentase hasil} = 49,94 \% \text{ (0,450 gram)}$$

$$\text{Sintesis III} : \text{ Persentase hasil} = 51,61 \% \text{ (0,465 gram)}$$

Persentase hasil rata-rata

$$: \frac{51,61 \% + 49,94 \% + 51,61 \%}{3} = 51,05 \%$$